

***Arboriculture 101***  
***Tree – Soil Relationships***

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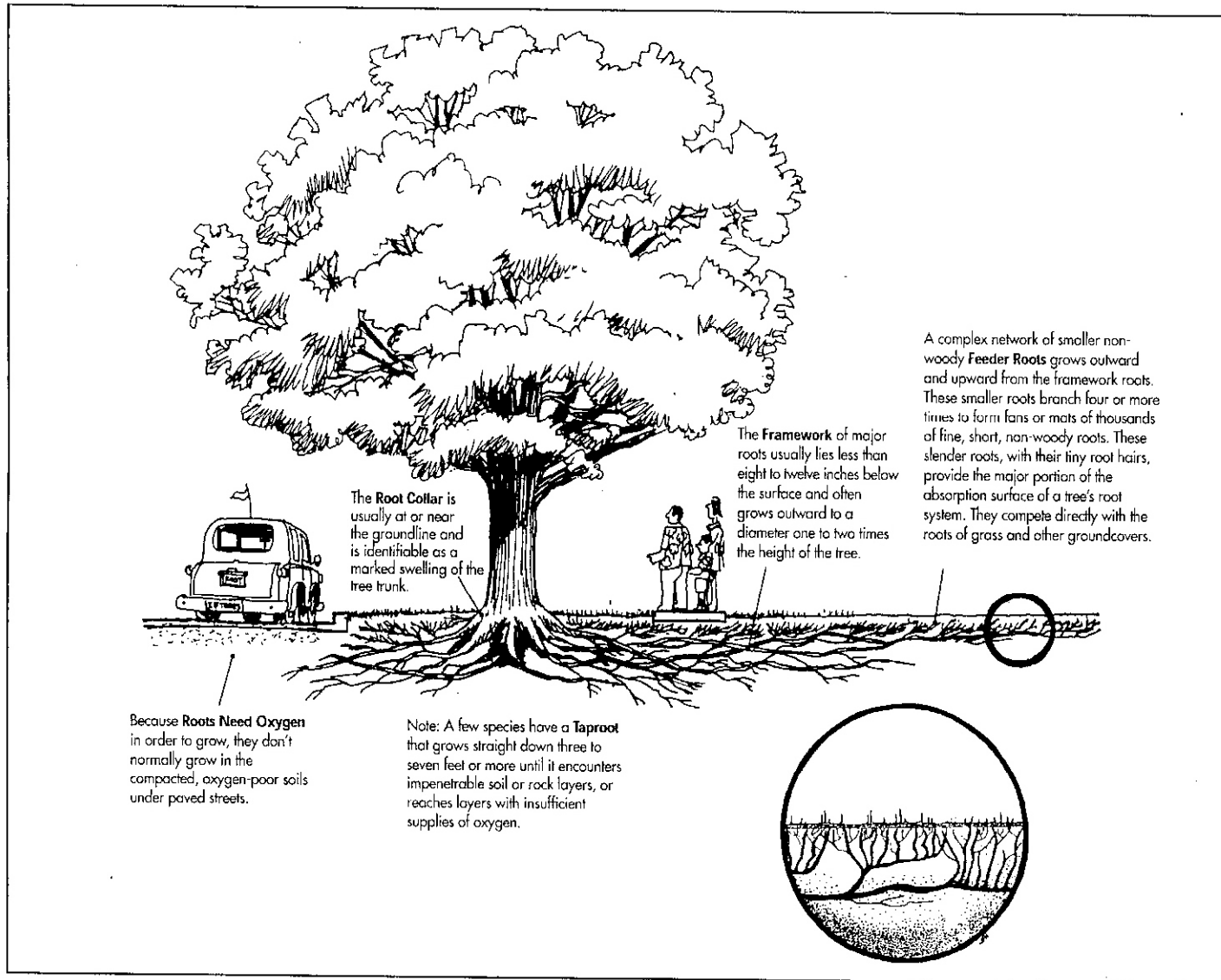


Figure 1.12 Roots grow where water, oxygen, and space are available.



10-10-10 Fall



0-10-10 Spring

Fraser firs planted  
in 2001  
Treatments applied  
annually since  
2002

Photos taken  
3/8/06



No Fertilization



10-10-10 Spring



10-10-10 Spring & Fall















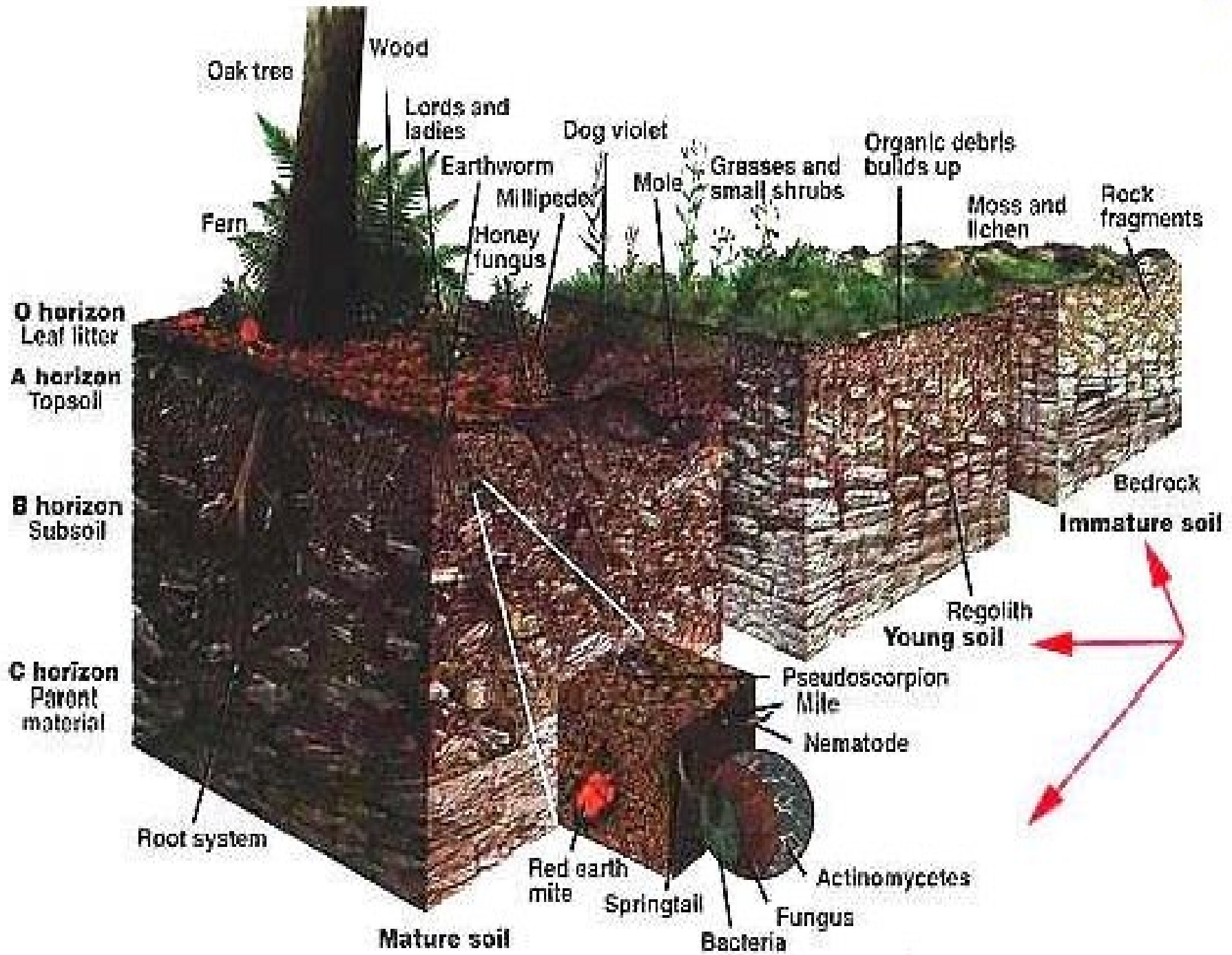






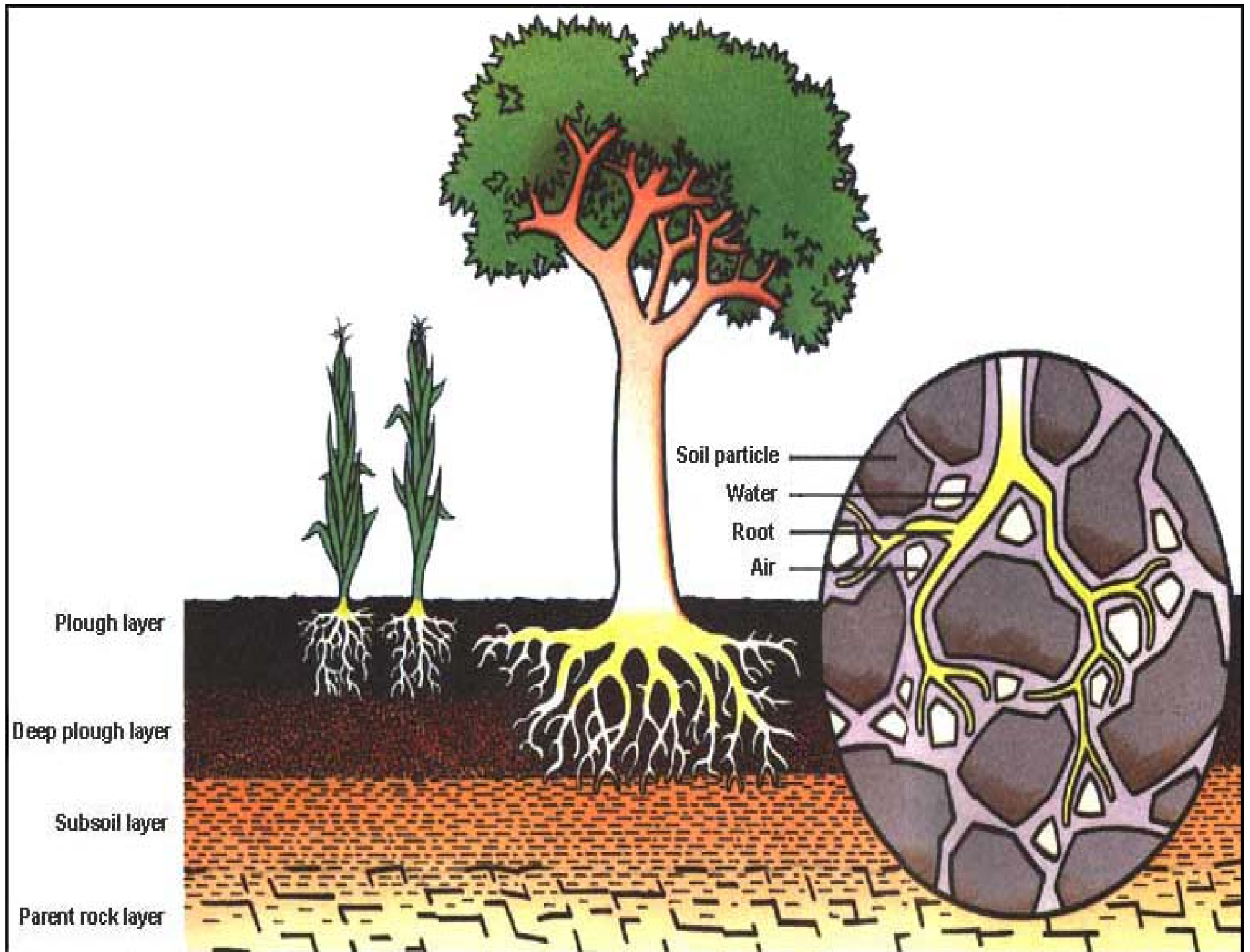












# ***Soil and Tree Fact 1***

Trees take up moisture and soluble nutrients through their roots and those roots must continue to grow in order to continue functioning.

# ***Soil and Tree Fact 2***

In order to grow and function, roots must have a good supply of oxygen (for respiration) in balance with water.



# ***Soil and Tree Fact 3***

Tree roots grow in soil. Soil conditions must be appropriate for healthy root growth all year long or roots will fail.

# ***Soil and Tree Fact 4***

In natural settings, tree roots are part of a community of decaying organic matter, micro-organisms and larger organisms that, together, provide enough nutrition for all.

# ***Soil and Tree Fact 5***

In most landscape settings, the value of the community of organisms is diminished. Care must be taken to create an environment that, at least, mimics some of the aspects of natural settings.

# How a Tree Grows

No



Yes

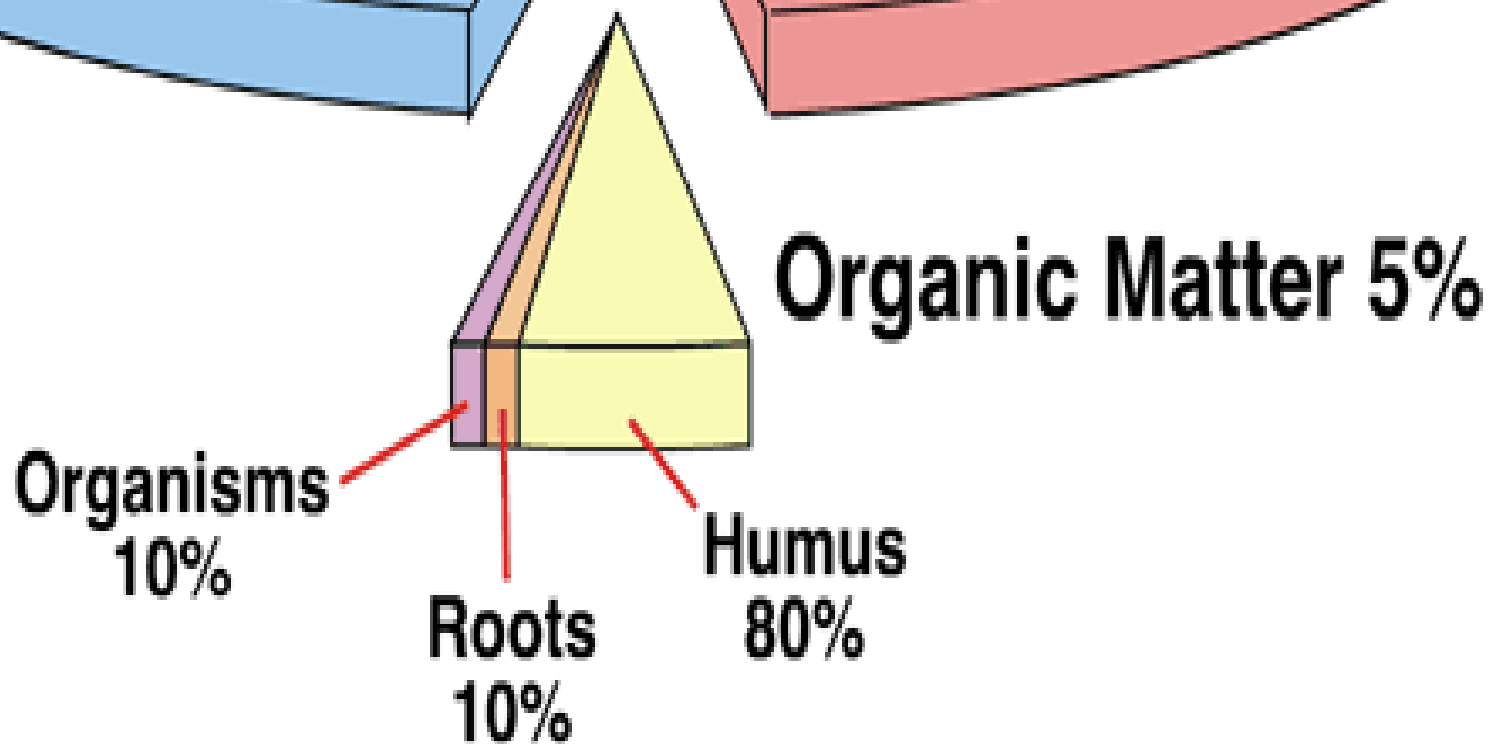
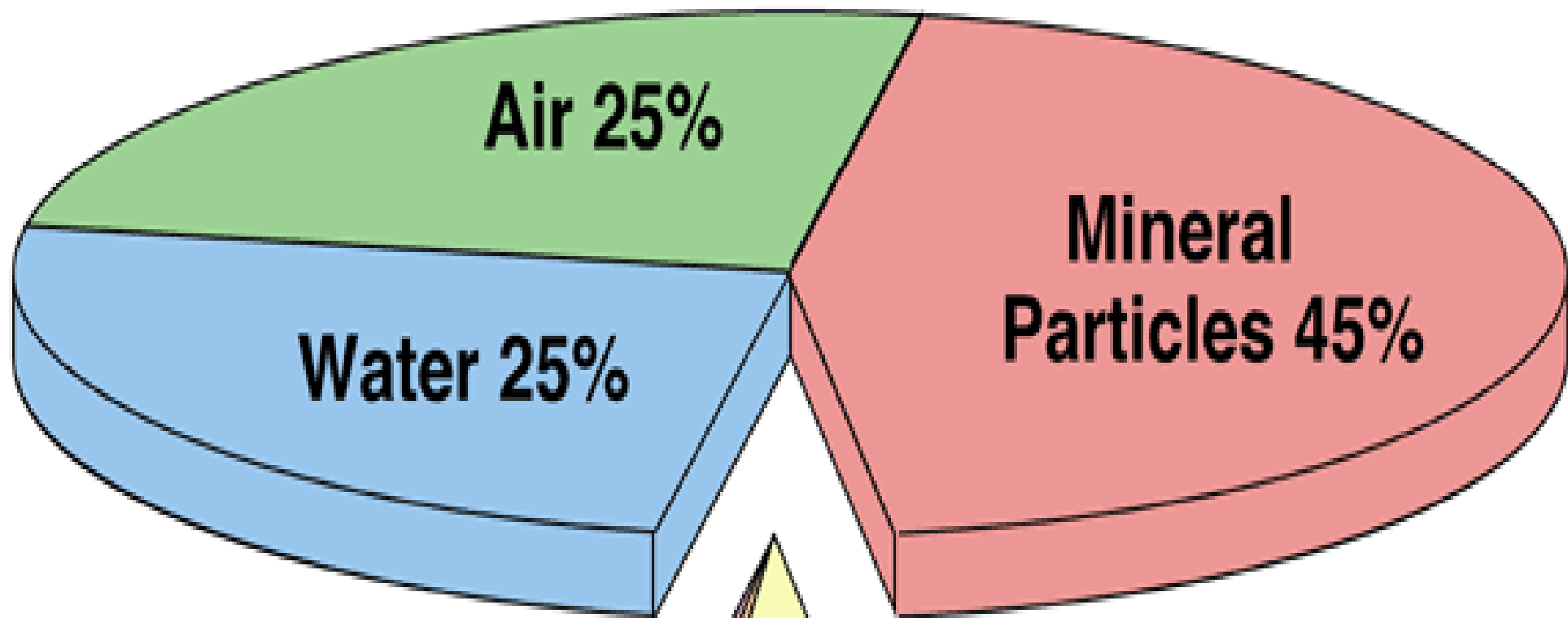




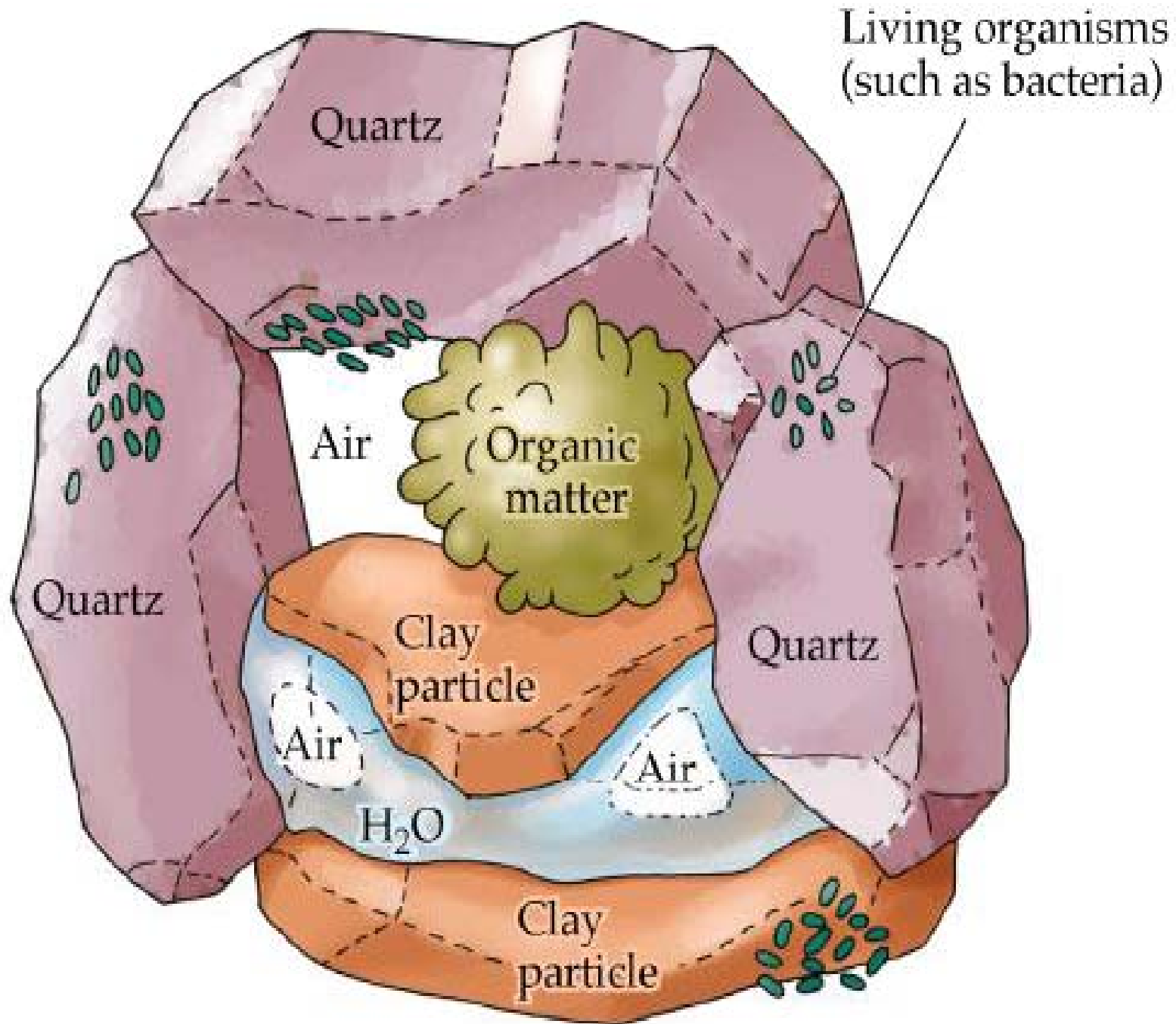


# ***Soil***

Soil is a dynamic, natural system synthesized in profile form at the earth's surface from a variable mixture of weathered minerals and decaying organic matter, under the influence of climatic and topographic conditions and living organisms; which supplies, when containing the proper amounts of air and water, mechanical support and, in part, sustenance for plants.



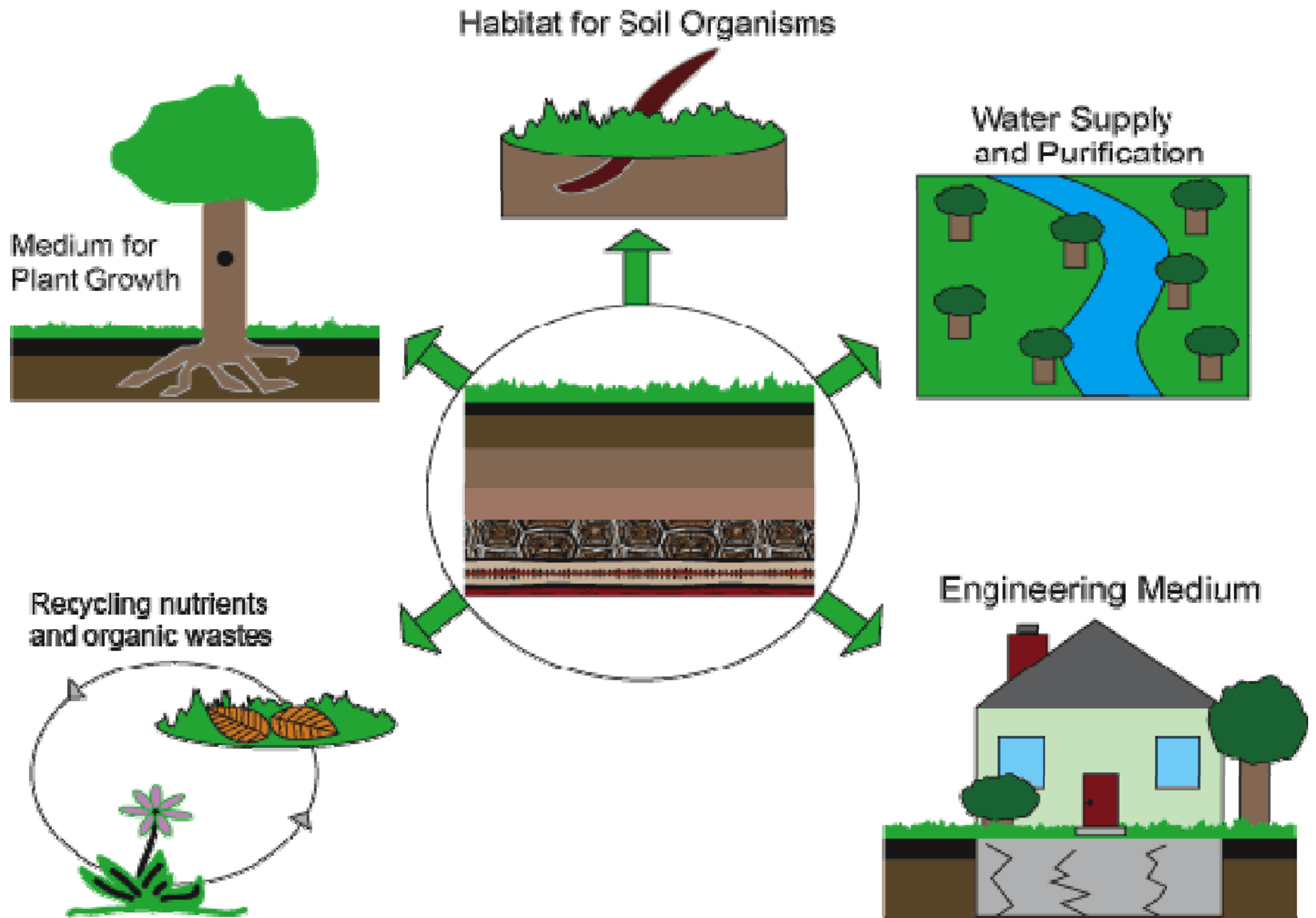




# ***Other soil uses***

- Storehouse of water
- Structural base for houses and other buildings and roads
- Habitat for soil organisms
- Recycling nutrients and organic matter

# The Five Functions of Soil



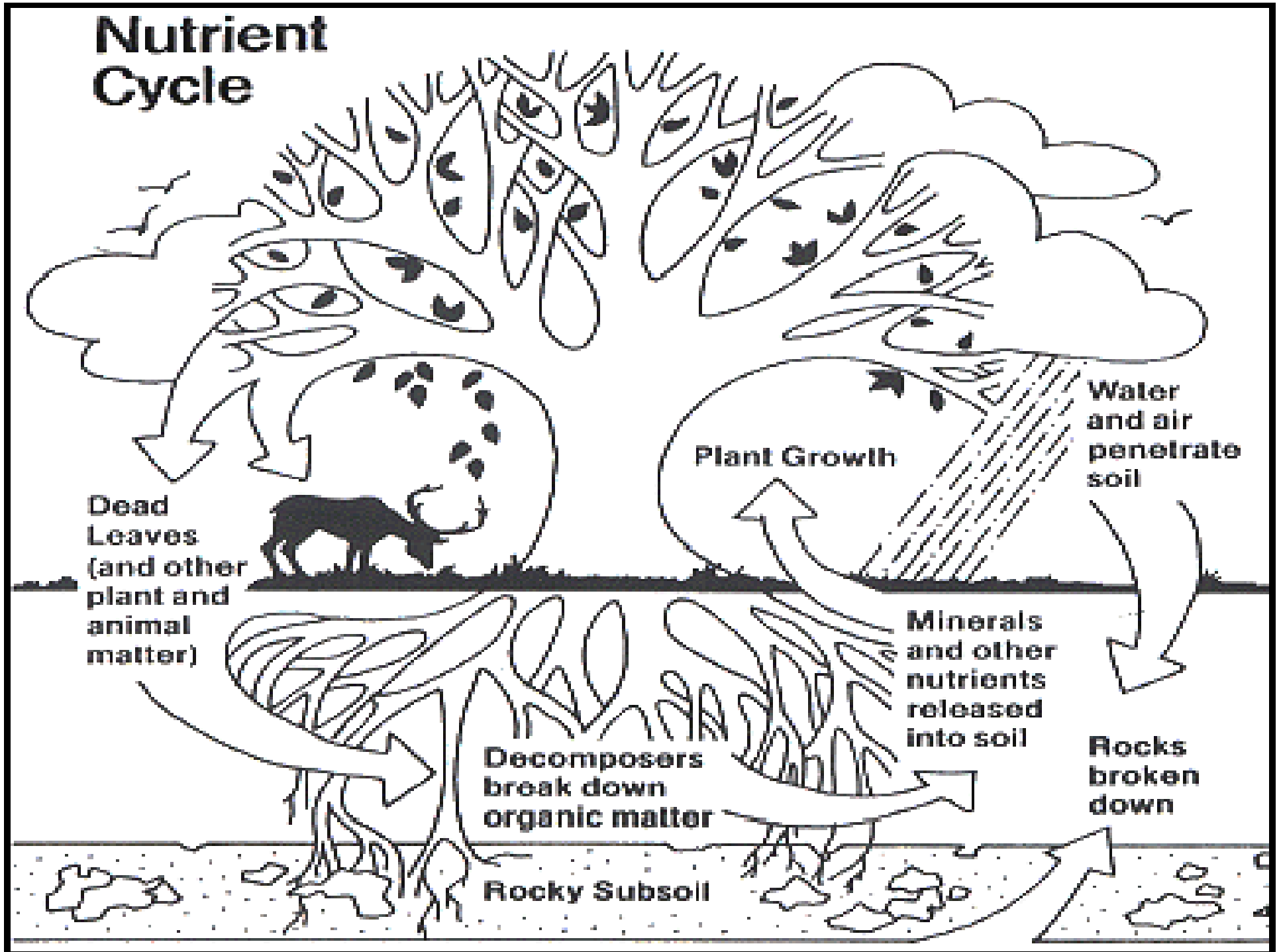
# ***Rhizosphere***

The zone of intense biological activity near actively growing roots. As roots grow through soil, root caps and external layers of cells fall off and organic compounds are released from the root tissue. This is a continuous source of organic matter that is food for microorganisms. Most important moisture and nutrient uptake activities by roots take place in the rhizosphere.

# ***Nutrient Cycling***

As a tree grows, roots absorb nutrient minerals from the soil solution and produce new woody material and leaves. As seasons pass, plants or plant parts die and fall to the soil surface where they weather and are gradually decomposed by soil organisms, releasing nutrients into the soil, available for root uptake again. This is how plants survive in natural settings, not in landscapes.

# Nutrient Cycle



# ***Practical tree/soil health care***

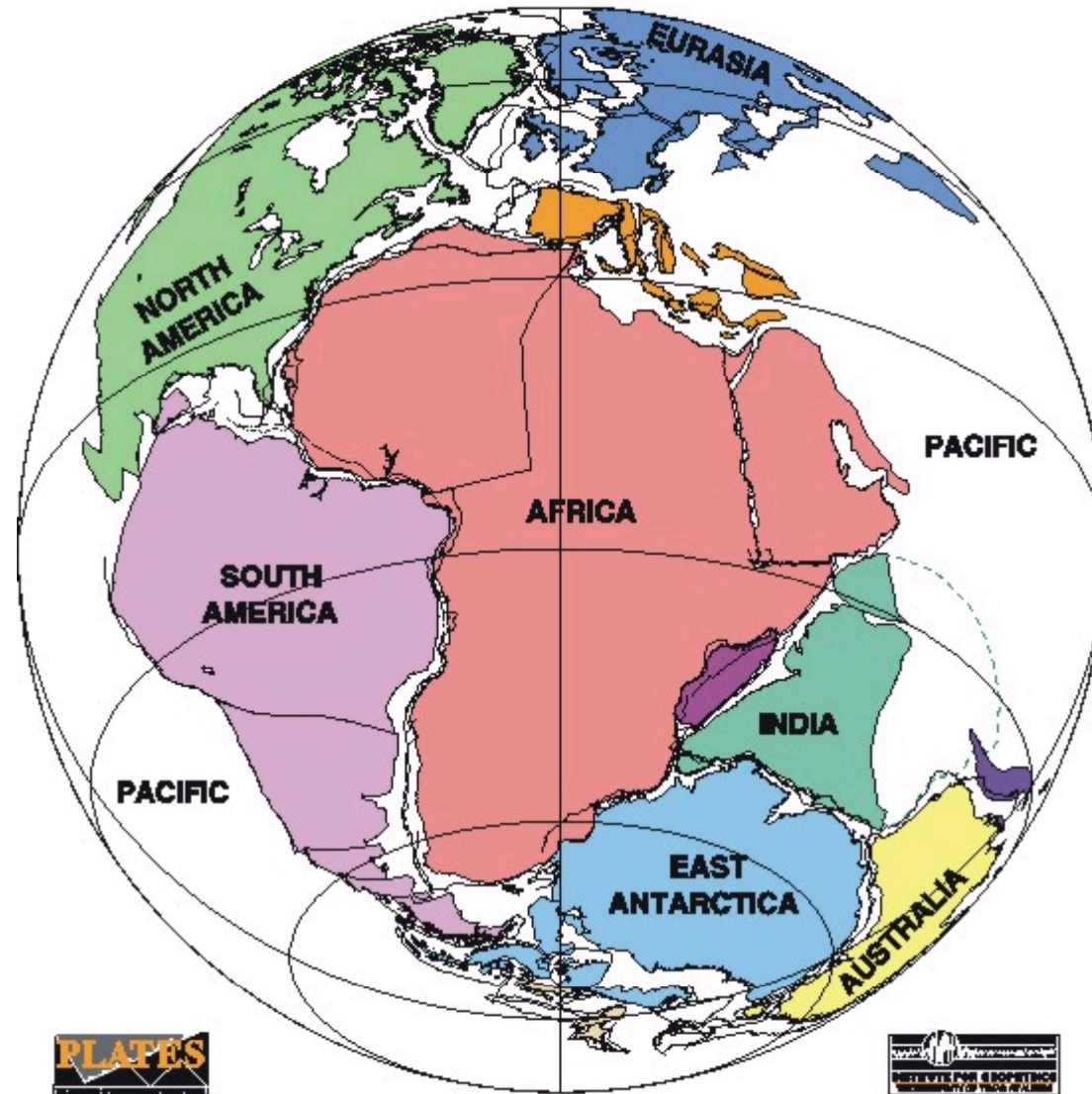
- Develop and carry out methods that optimize healthy root environments
- Duplicate nutrient cycling whenever possible
- Understand that healthy soils can provide most of tree's nutrition without fertilization
- Understand how soil affects water and air availability

# ***Soil Science***

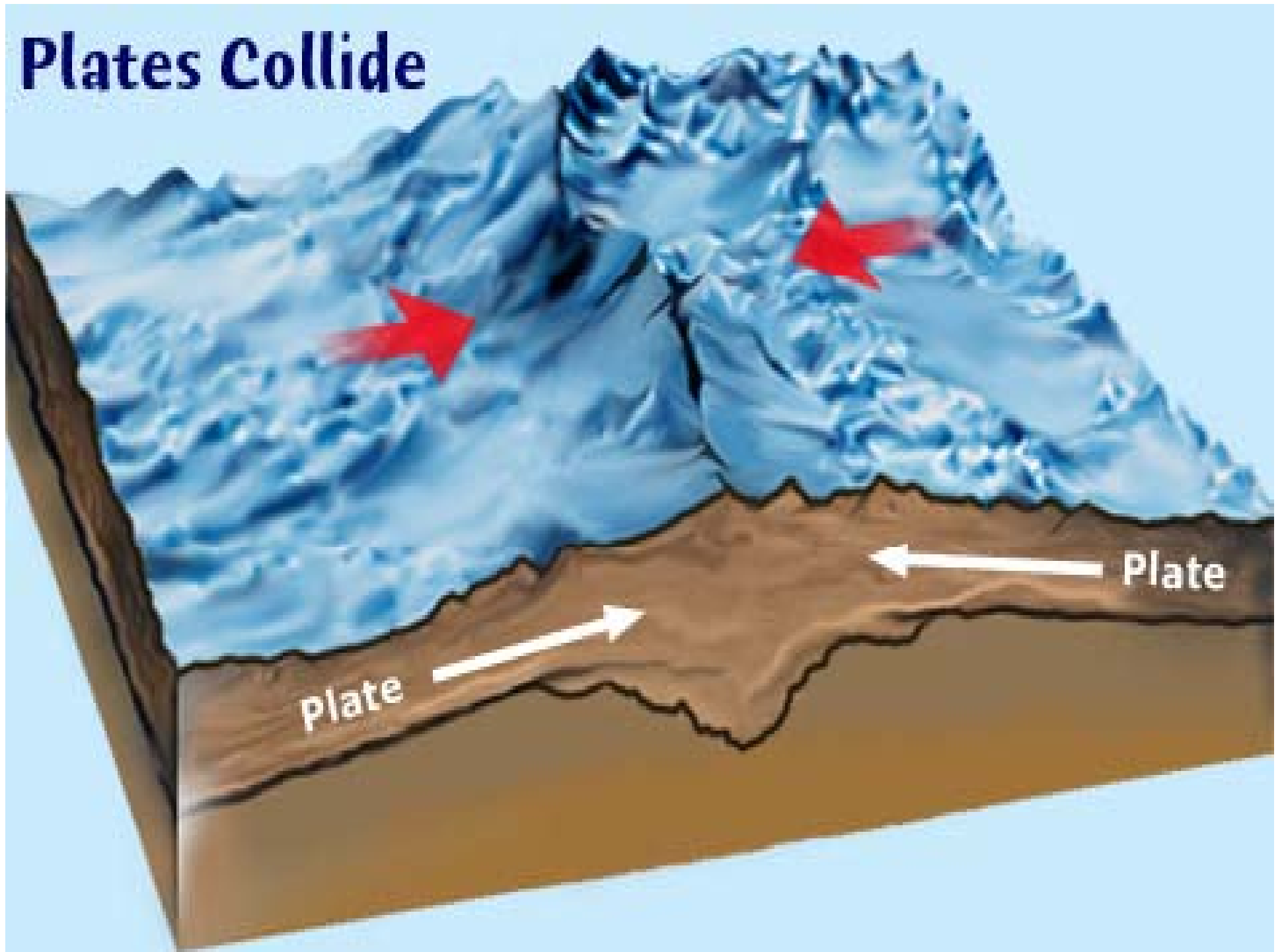
- Physical characteristics
- Chemical characteristics
- Biological characteristics



# PANGEA



# Plates Collide







# ***Mineral soils are formed by weathering of rocks***

## Physical weathering

water (erosion)

freezing/thawing

wind

vegetation

## Chemical weathering

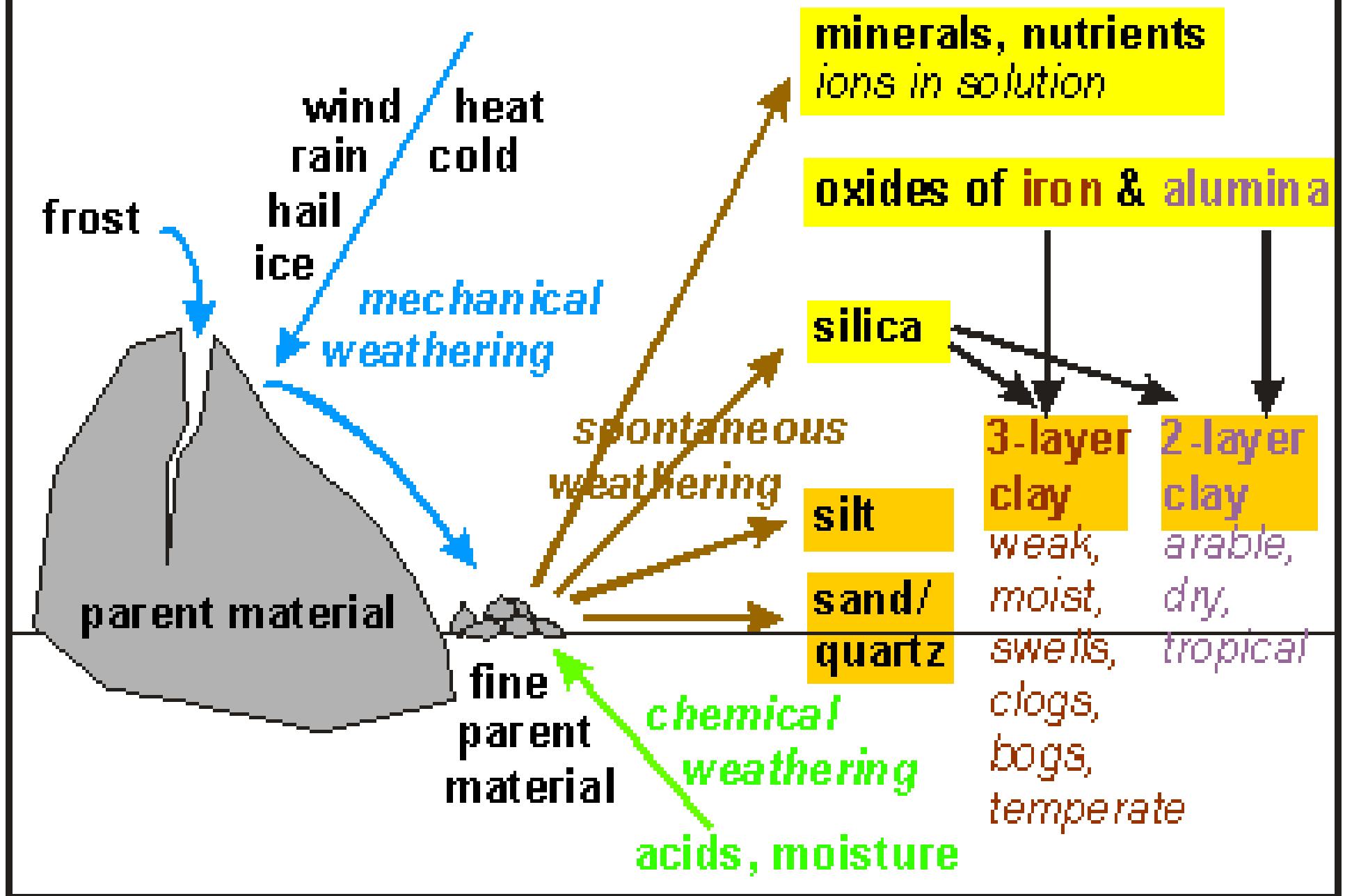
oxidation/reduction

hydration/hydrolysis

solubility

carbonation

# How is soil made? - weathering



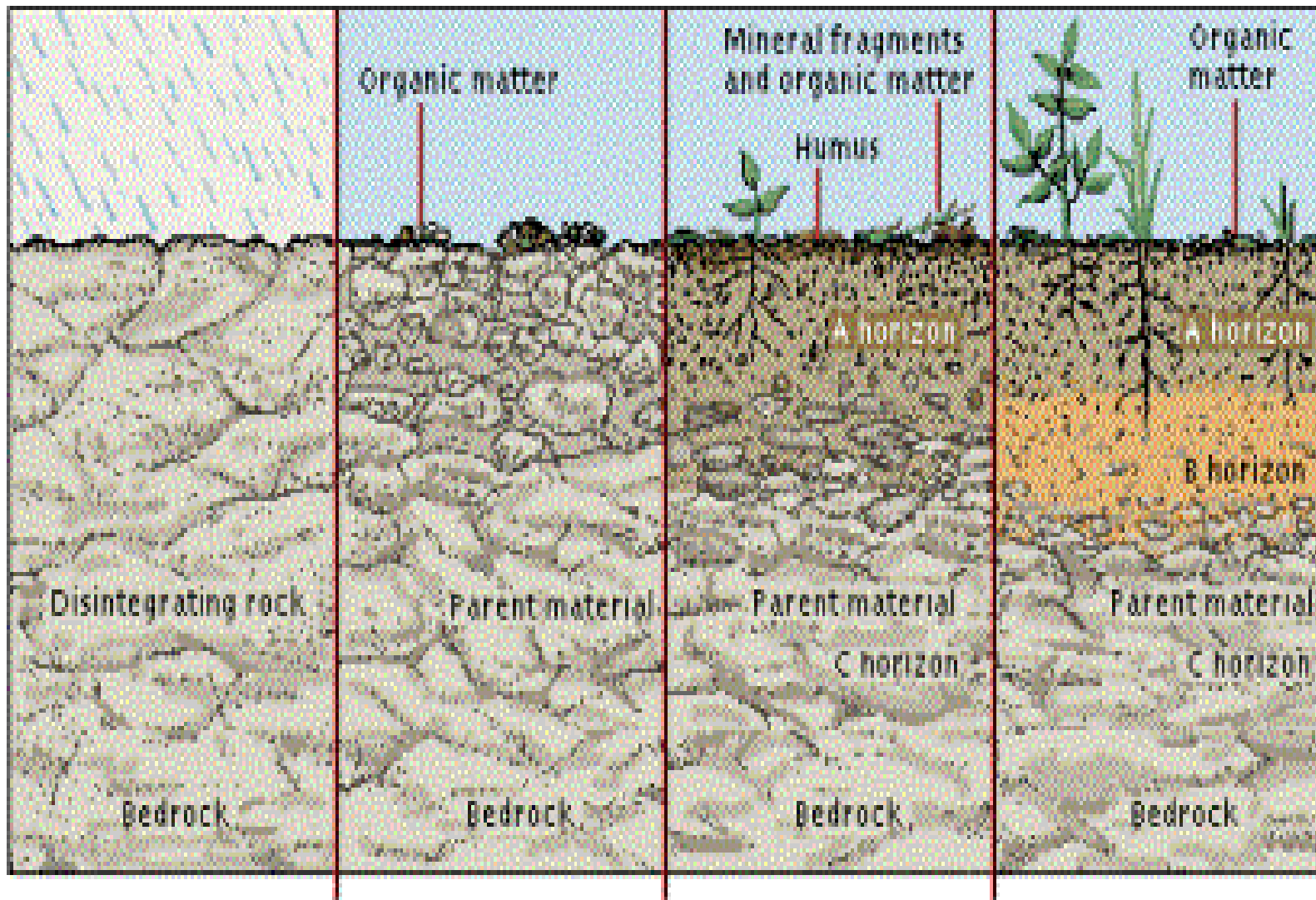
# *Parent material*

- Rocks that are the source of the mineral portion of soil
- Weather at different rates
- Bedrock beneath soil is often the parent material (though not always, thanks to glaciers)
- Sedimentary (sandstone, shale); igneous (basalt); metamorphic (marble, granite, quartz)

# ***Soil and rock are interchangeable***

- Over millions of years, rock becomes soil and soil becomes rock again
- Same total amount since earth was formed
- Continental drift
- Most CT soils are where they are because of glacial activity





Bedrock begins to disintegrate

I

Organic materials facilitate disintegration

II

Horizons form

III

Developed soil supports thick vegetation

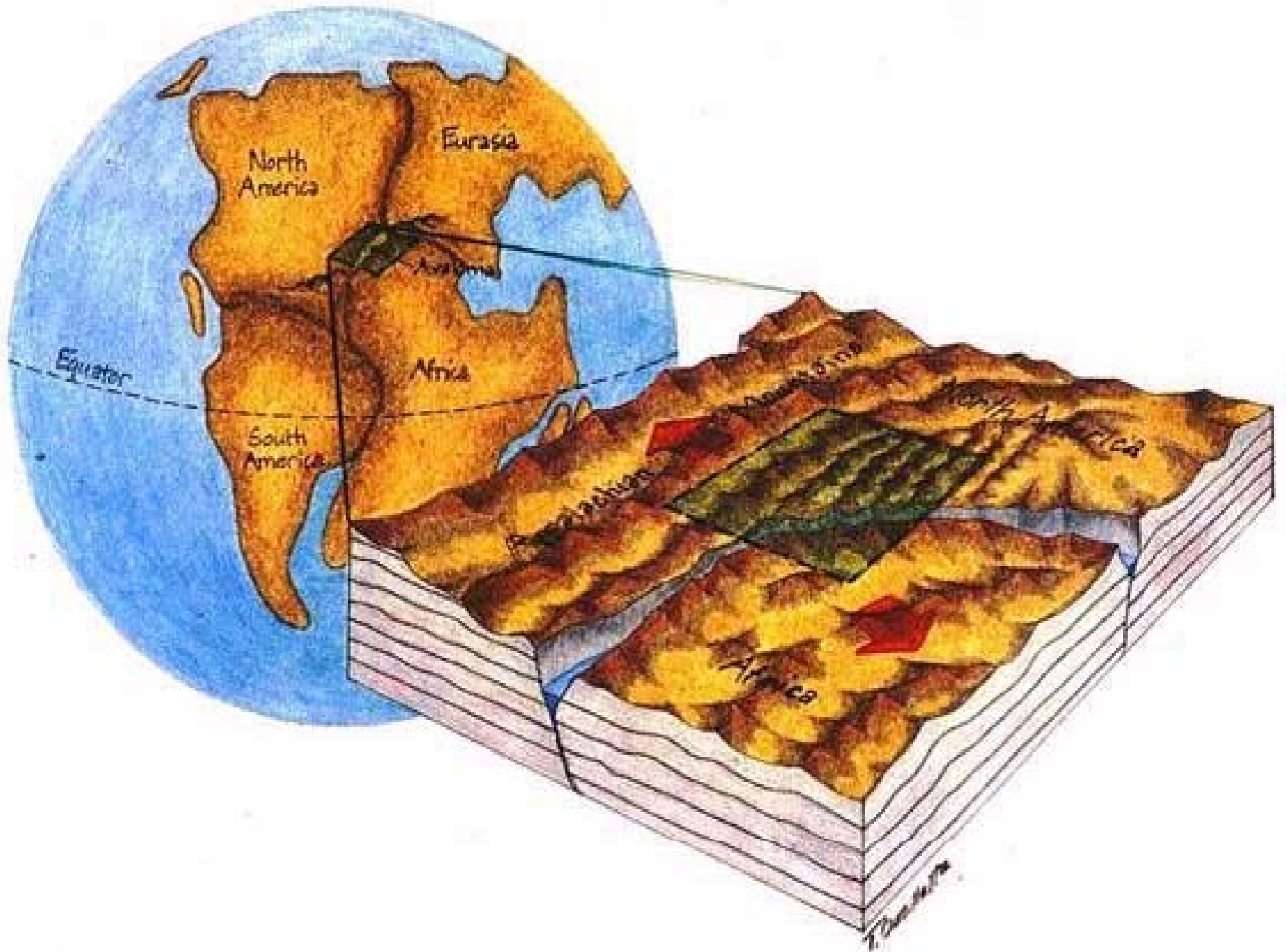
IV

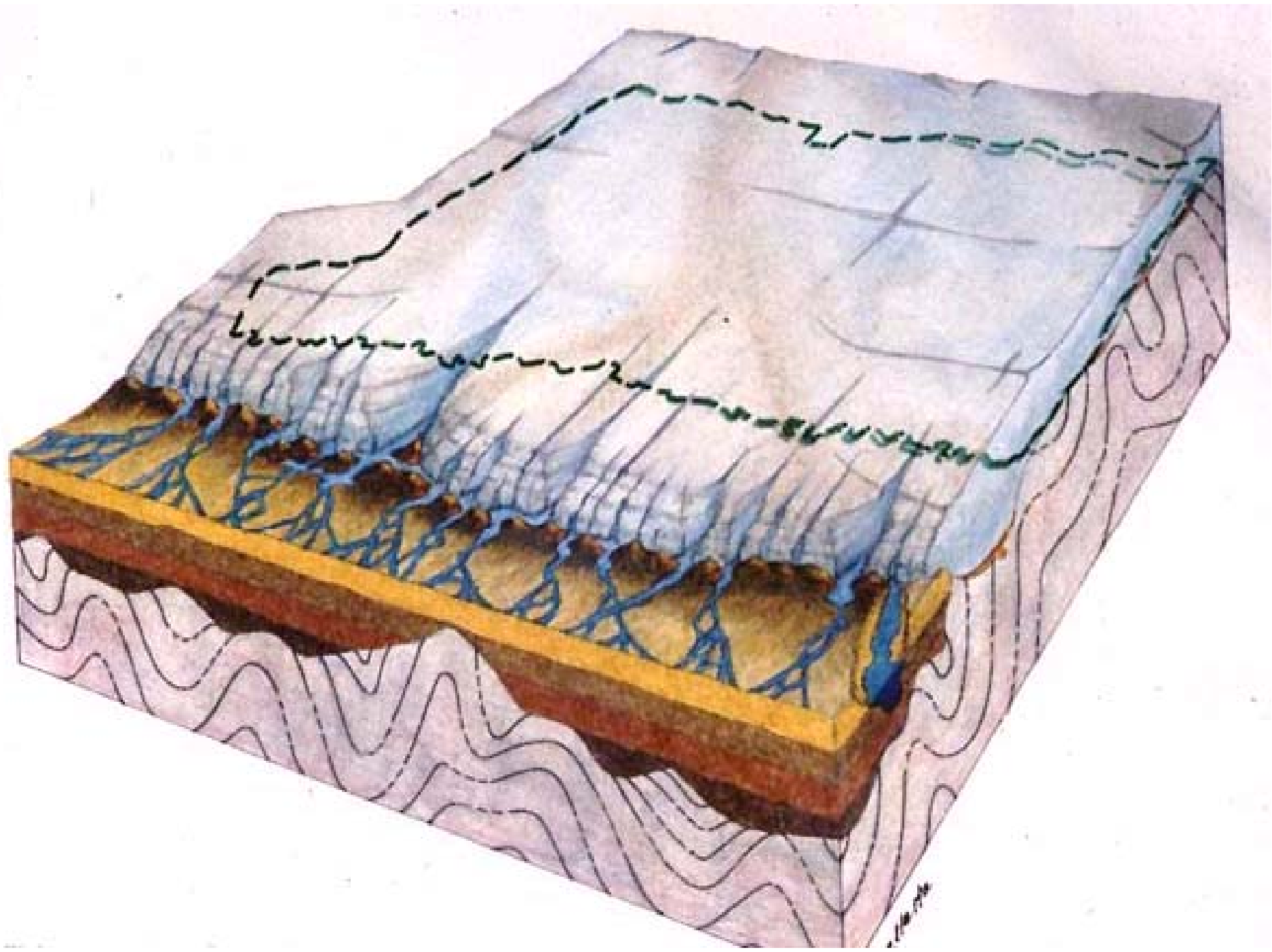
# ***Soil weathering influences***

- Climate
- Vegetation
- Parent material

# *How glaciers moved soil*

- Glacial till – mixed/plowed up, left in profile
- Stratified drift/glacial outwash – caught in flow/meltdown
- Alluvial – carried in water, left behind when dried
- Organic deposits – bogs, mucks





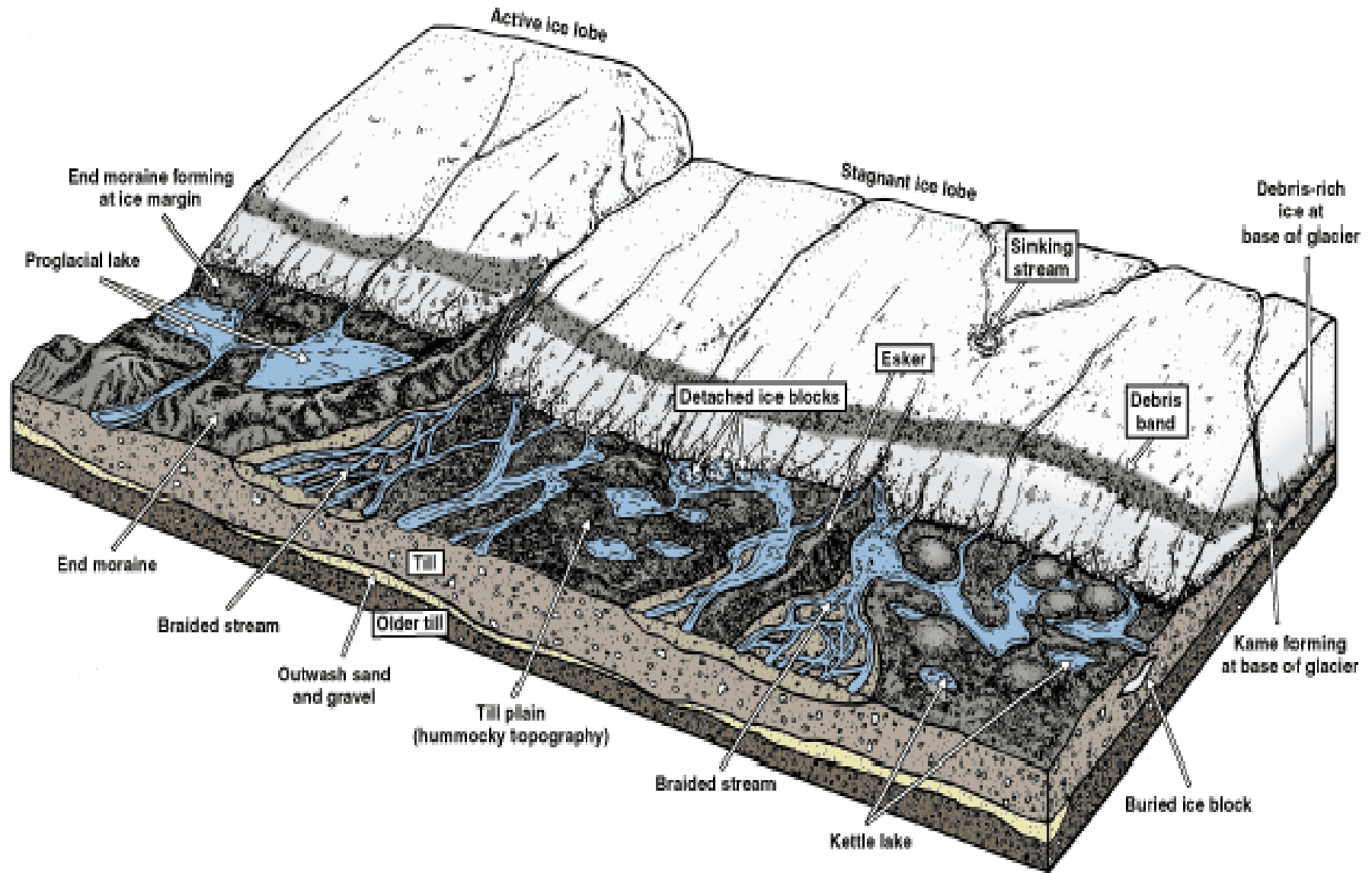


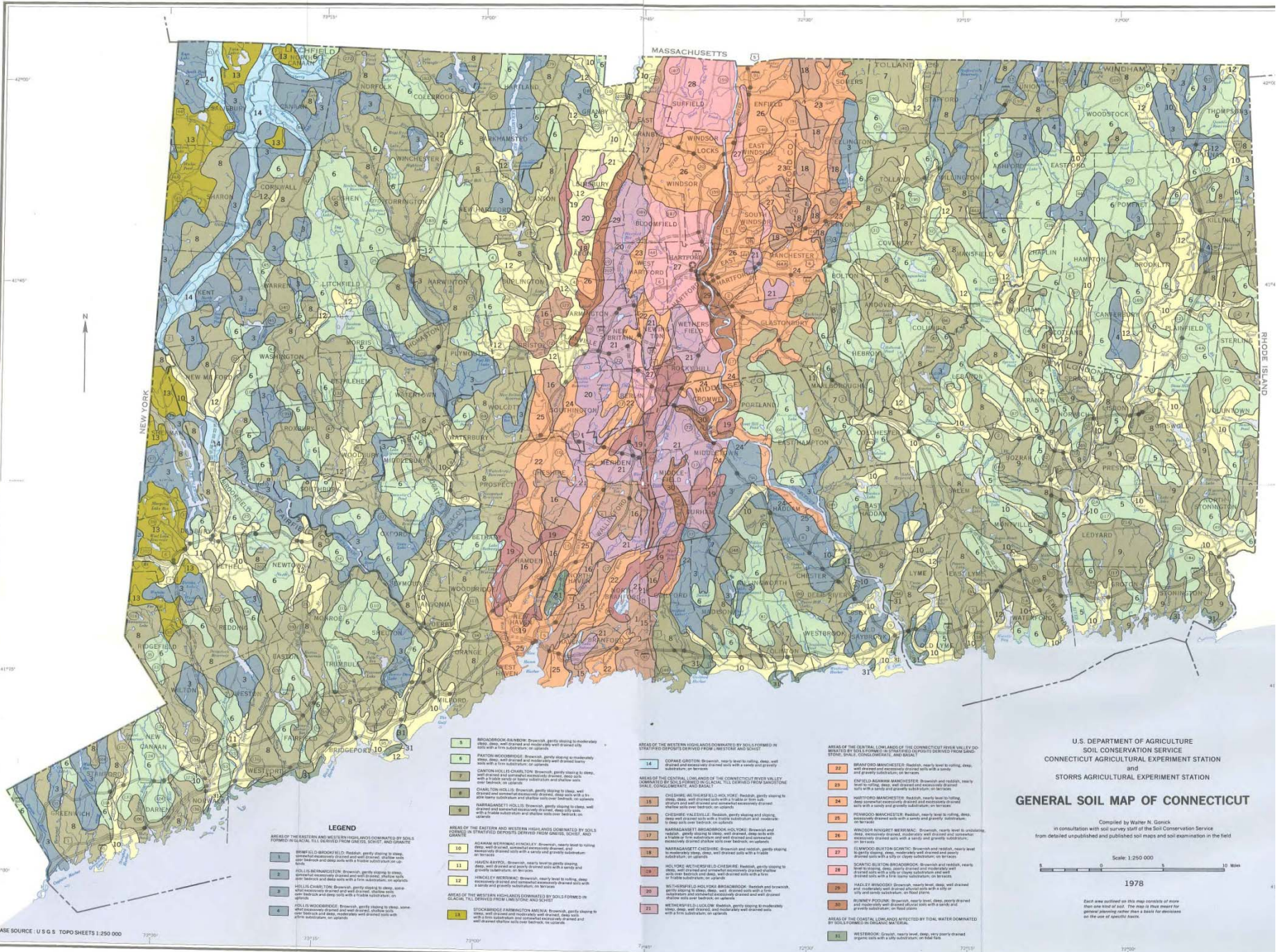






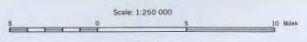
# Continental Glacier





U. S. DEPARTMENT OF AGRICULTURE  
SOIL CONSERVATION SERVICE  
CONNECTICUT AGRICULTURAL EXPERIMENT STATION  
and  
STORRS AGRICULTURAL EXPERIMENT STATION  
**GENERAL SOIL MAP OF CONNECTICUT**

Compiled by Walter R. Gunkel  
in consultation with soil survey staff of the Soil Conservation Service  
from detailed unpublished and published soil maps and soil examination in the field



1978

Each area outlined on this map consists of more than one kind of soil. The map is that meant for general planning rather than a basis for decisions on the use of specific lands.

BASE SOURCE: U. S. G. S. TOPO SHEETS 1:250,000

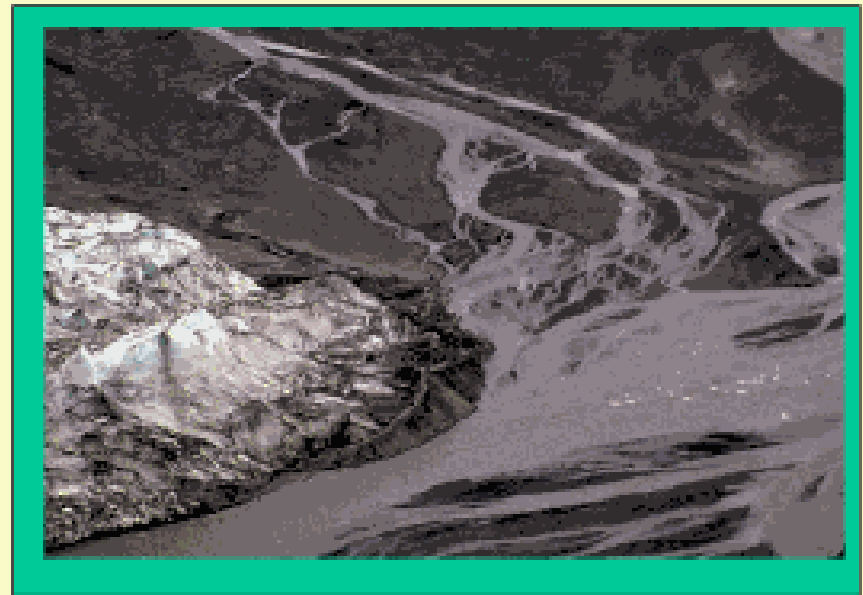
# Glacial Till

- Unsorted/stratified material deposited beneath and within glacial ice.
- Heterogeneous mixture of all particle sizes (boulder to clay).
- Oldest surficial deposit.



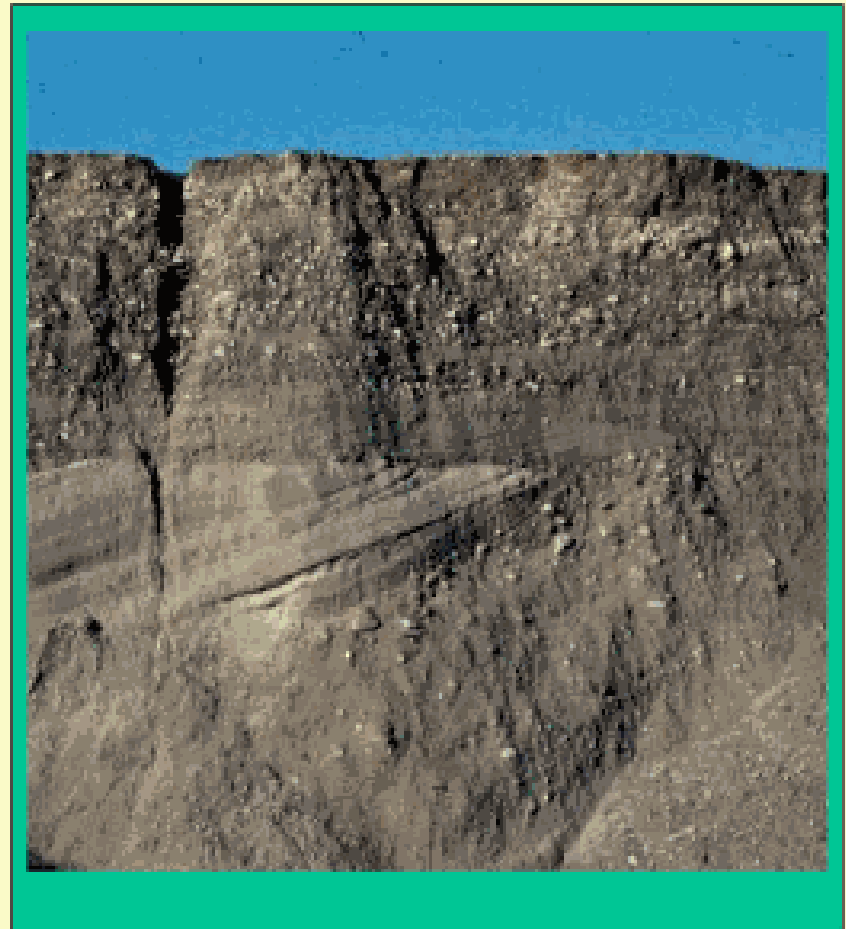
# Glacial Fluvial (Outwash)

- Sediments deposited by glacial meltwater.
- Stratified layers of sand, gravel, and fines.
- Types: Proglacial and Proximal (ice contact).
- Landforms: Plains, eskers, kames, deltas.



# Outwash Properties

- Dominantly sand and gravel sized particles.
- Rapid water movement, associated with aquifers.
- Apparent watertable.
- Few limitations for most uses.



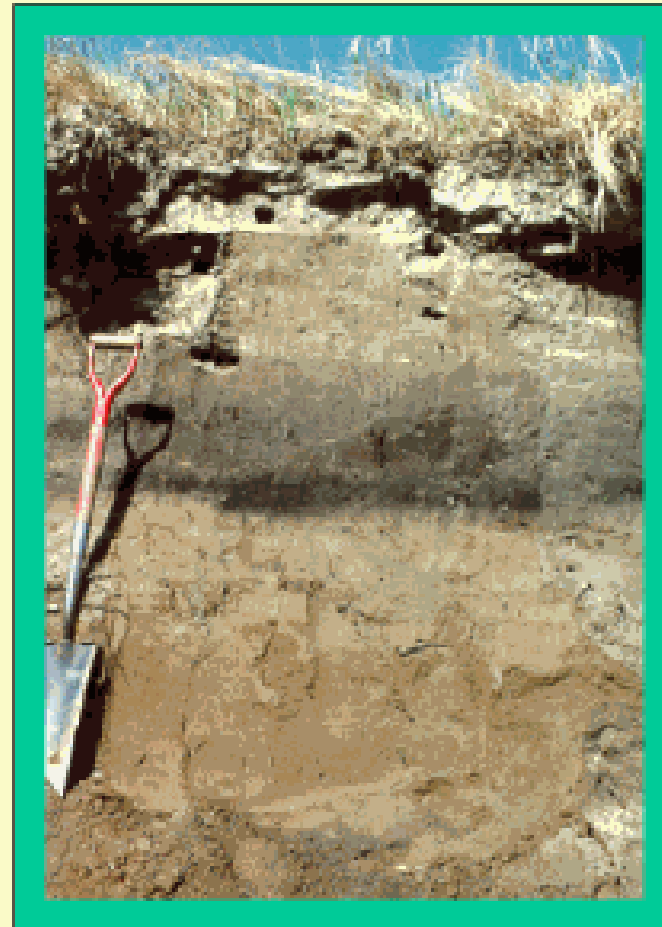
# Post-Glacial Deposits

- Material deposited after glacier left (Holocene-10K BP).
- Eolian - wind deposited sand to silt sized particles.
- Most upland soils in NE have a thin 18-36 inch eolian cap. Deposited rapidly after ice left.



# Post-Glacial Deposits

- Organic deposits - swamps, bogs, marshes.
- Alluvial deposits - modern day floodplains.
- Anthro-transported material.



# Parent Material

Geologic Material Soil  
Formed From.

- Types of minerals.
- Reaction of soil.
- Soil Color.

Types of PM: Glacial,  
Volcanic, Organic,  
Loess, Colluvium,





# New England Parent Materials

- Pleistocene Epoch (Ice Age) - 1.8 MYBP to 8 KYBP.
- 4 Major advances.
- Last- Wisconsinan advance covered all of NE.
- Soil parent materials - glacial & post glacial



# ***Soil physical characteristics***

- Soil is the interface between the atmosphere and the lithosphere
- Most soils in the US are at least 8000 years old
- Many soils as old as 600,000 years
- Some soils are forming as we speak
- Soil is unconsolidated rock

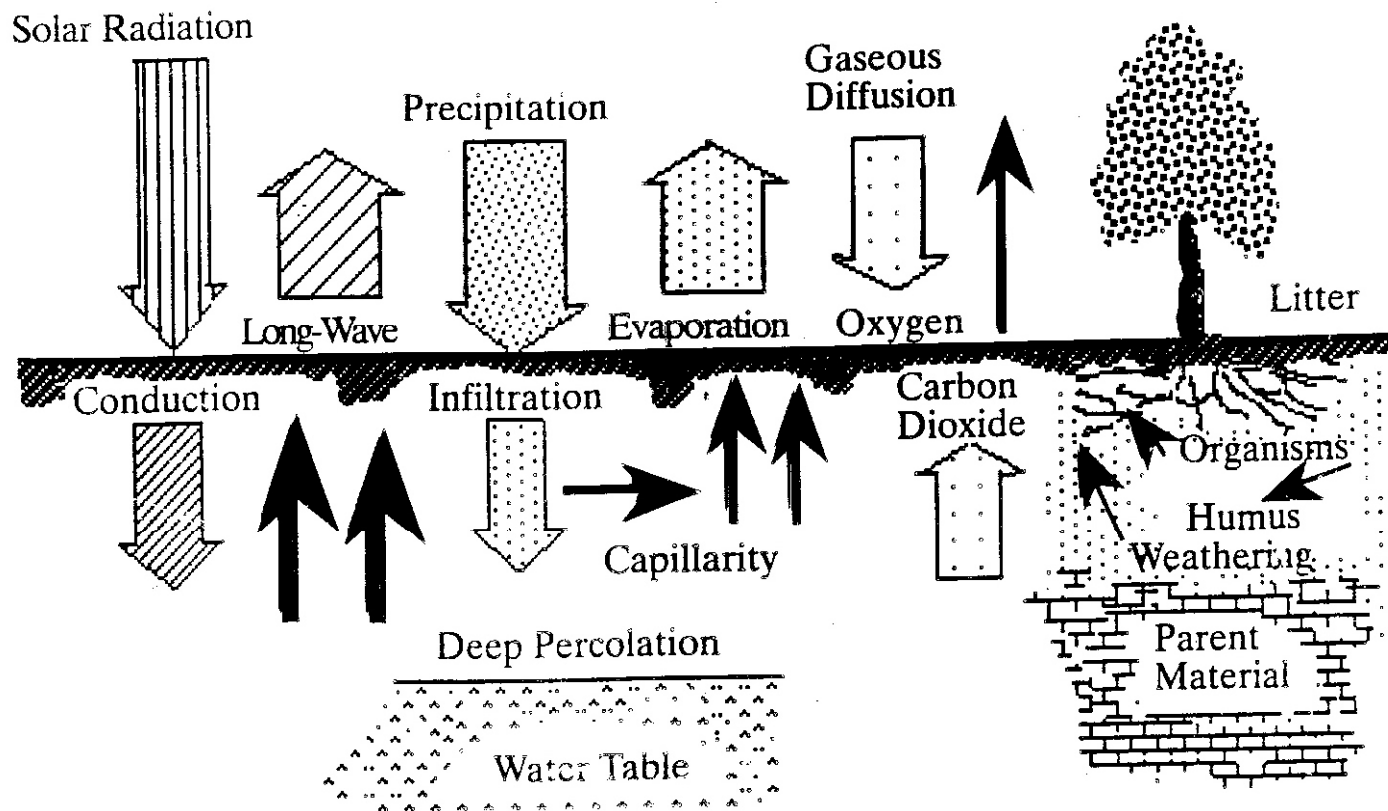
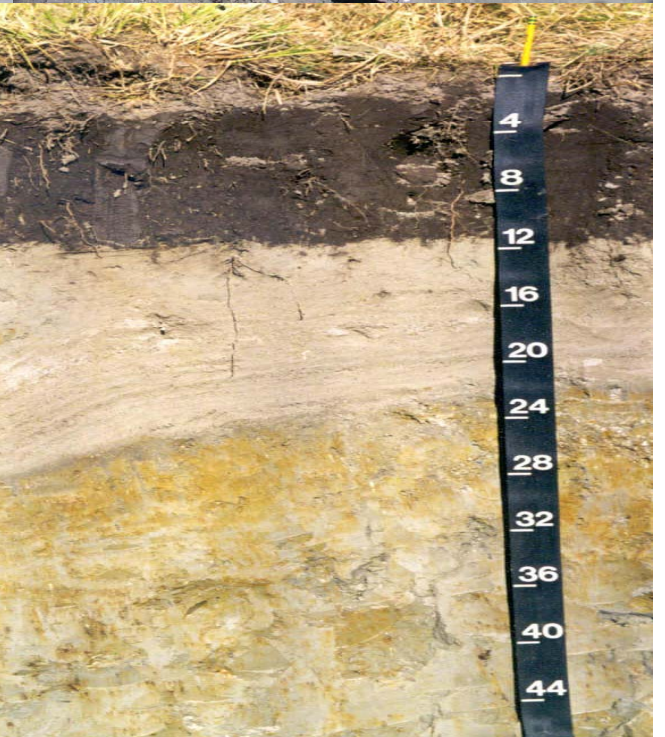
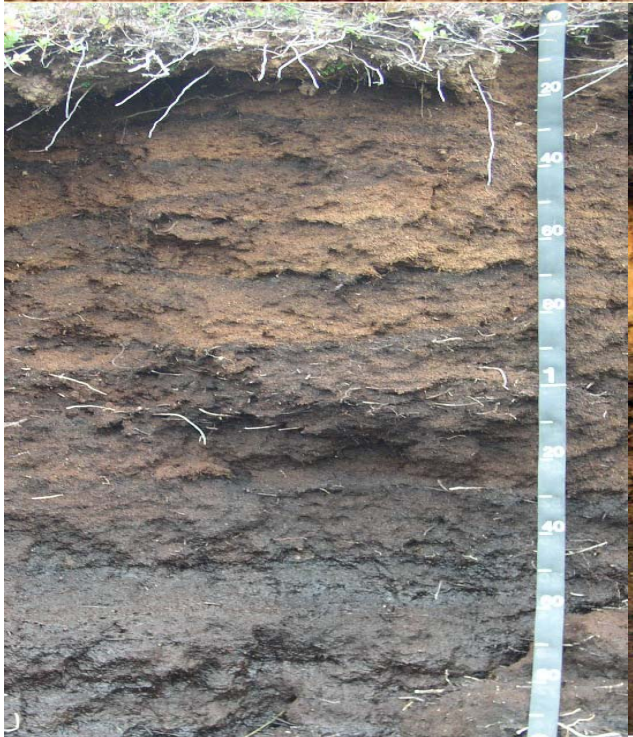
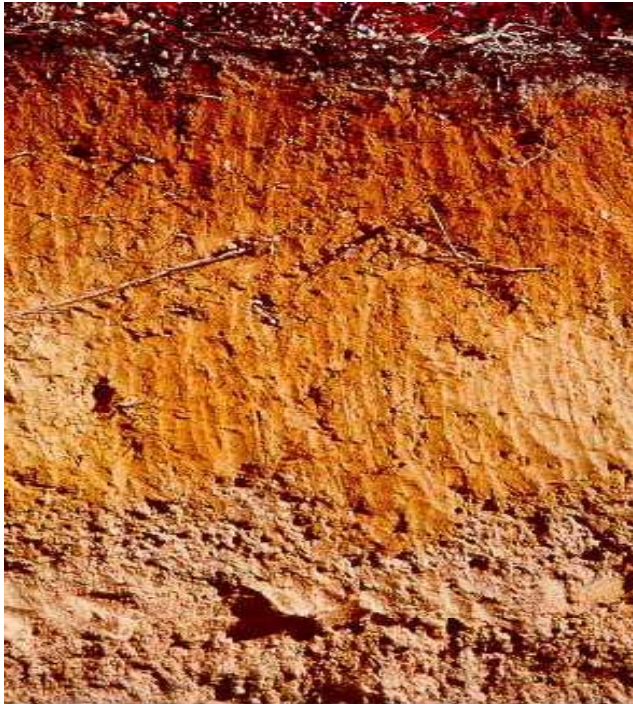
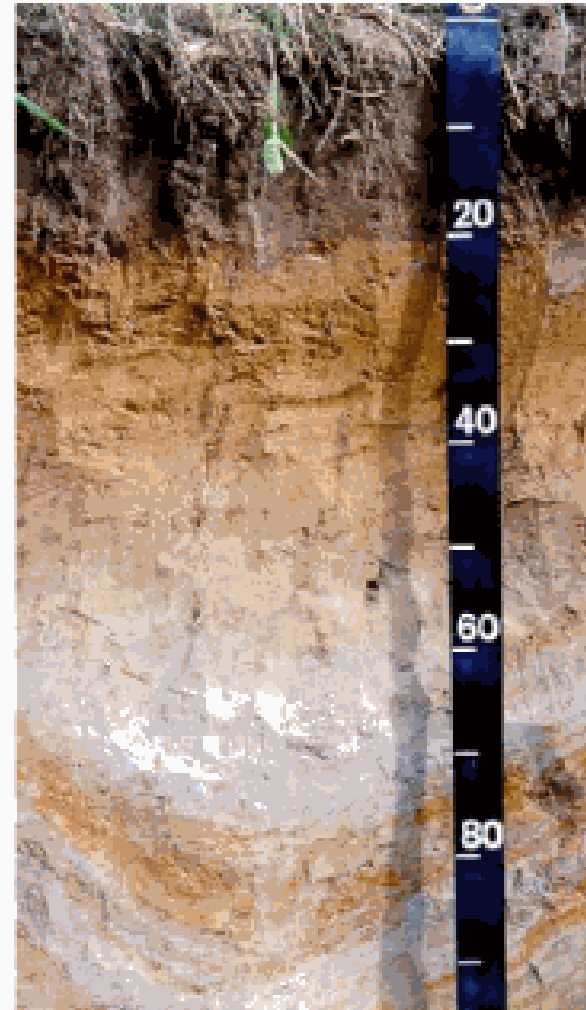


Fig. 1.1. The soil as the interface between the atmosphere and the lithosphere.



# Soil Color

- Easily identified property.
- Used to relate chemical/physical properties such as watertable depth, drainage, chemical constituents, formation, horizons.



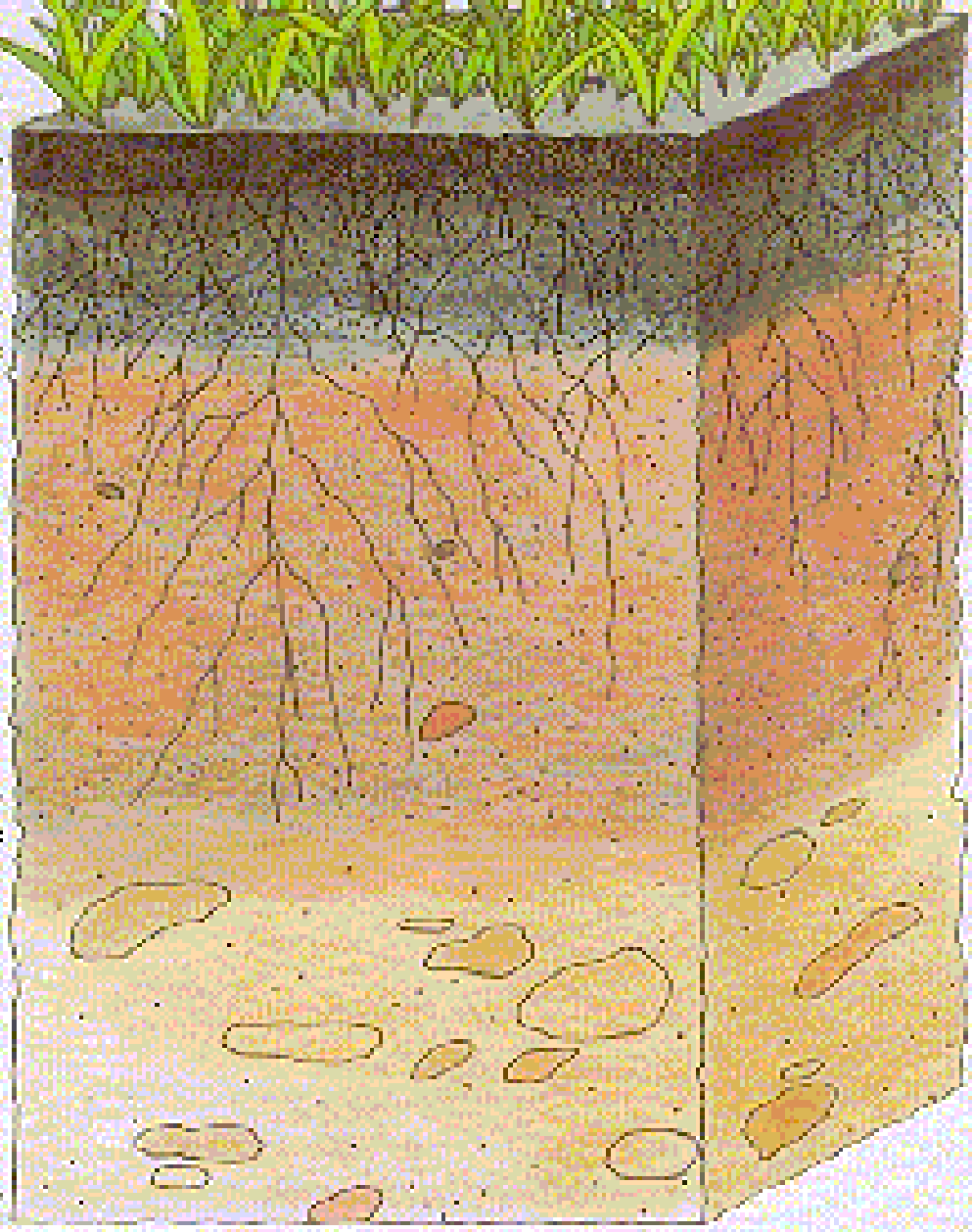
# *Soil Profile*

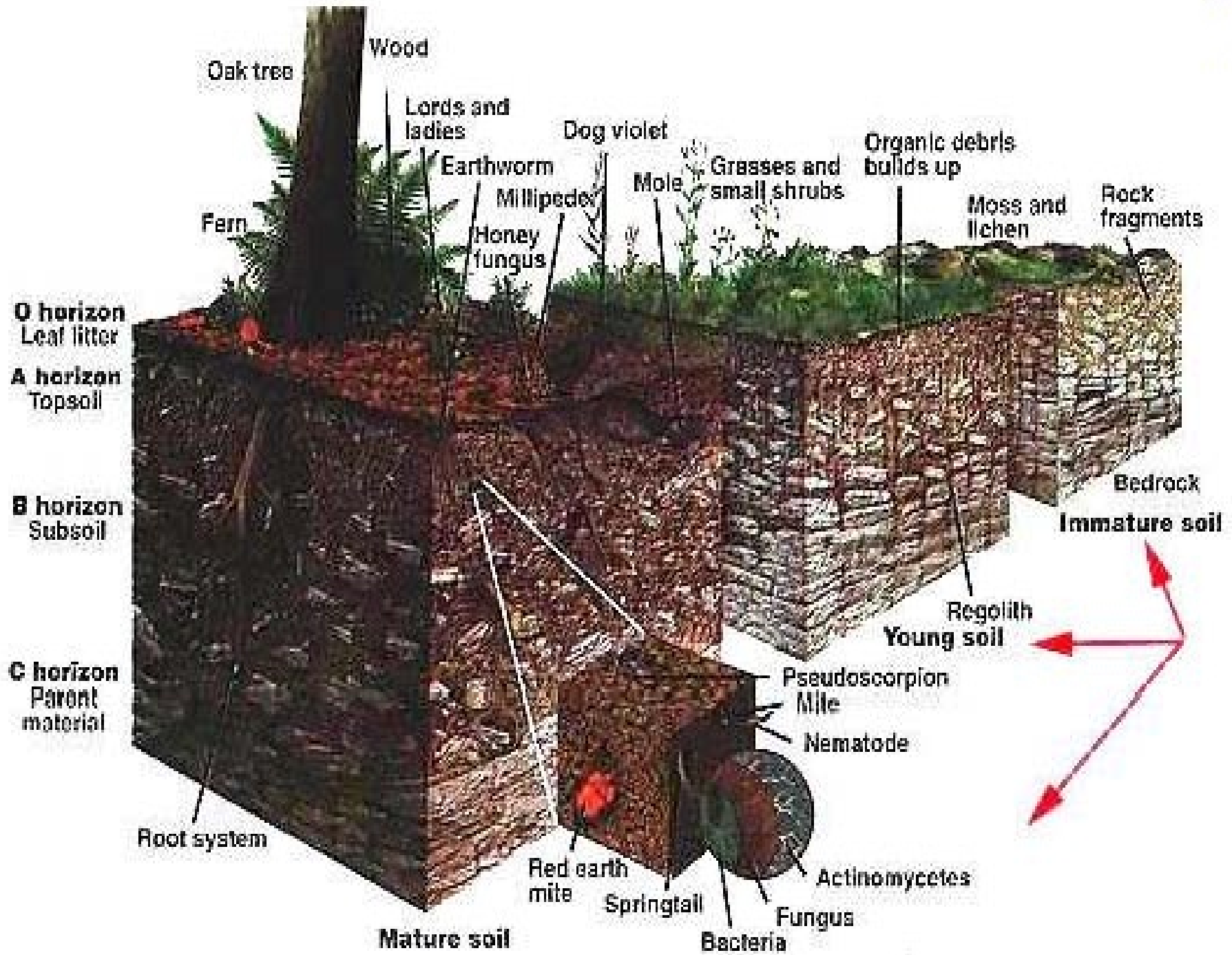
- Layers develop over time – influenced by physical, chemical, biological factors
- Horizons
  - organic layer
  - A – topsoil
  - B – subsoil
  - C - subsoil

# A Soil Profile

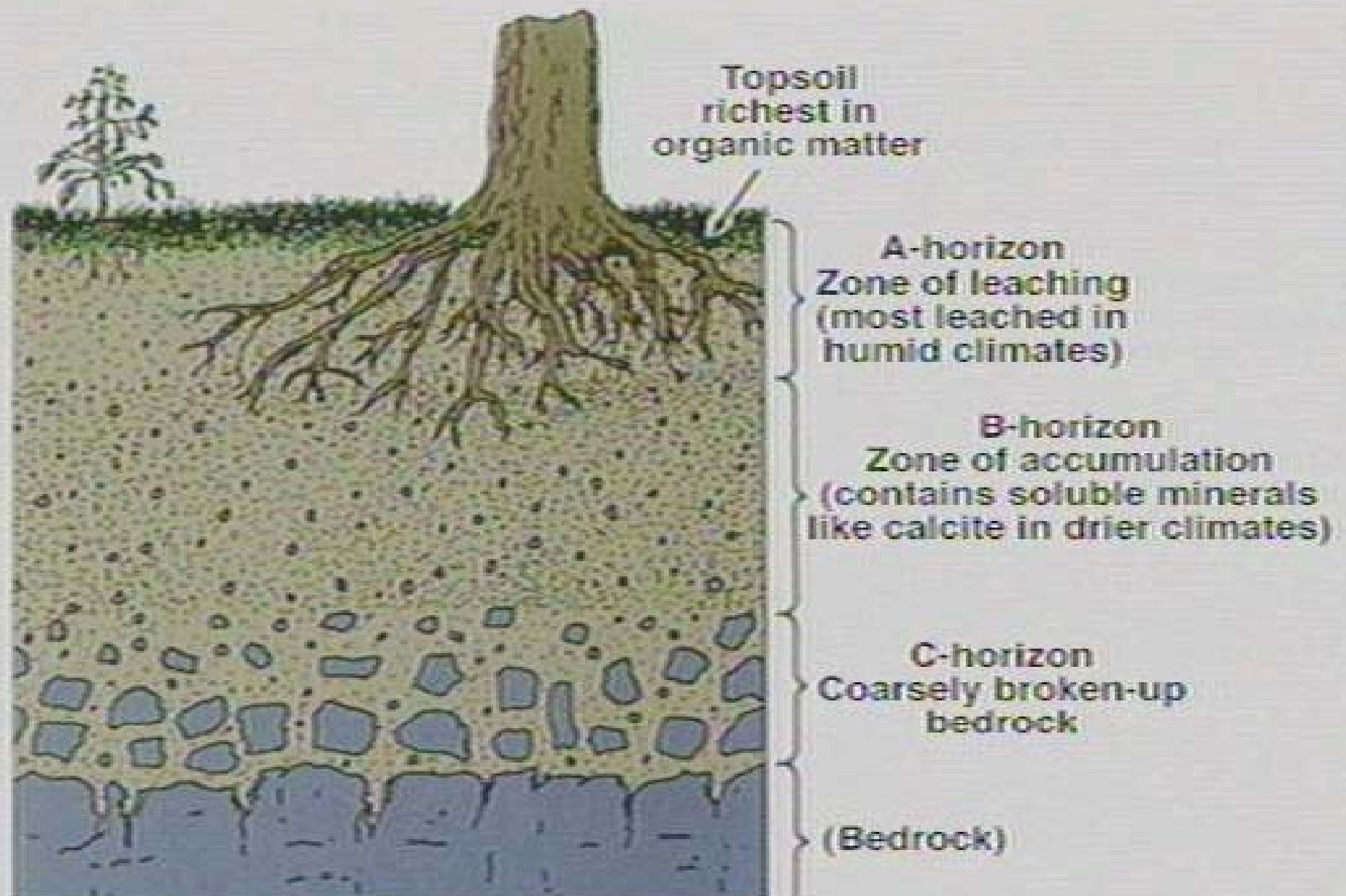
Horizons

O	0"	2"
A		10"
B		30"
C		46"









**Generalized relationship of soil horizons in soil development**



O

A

B

C

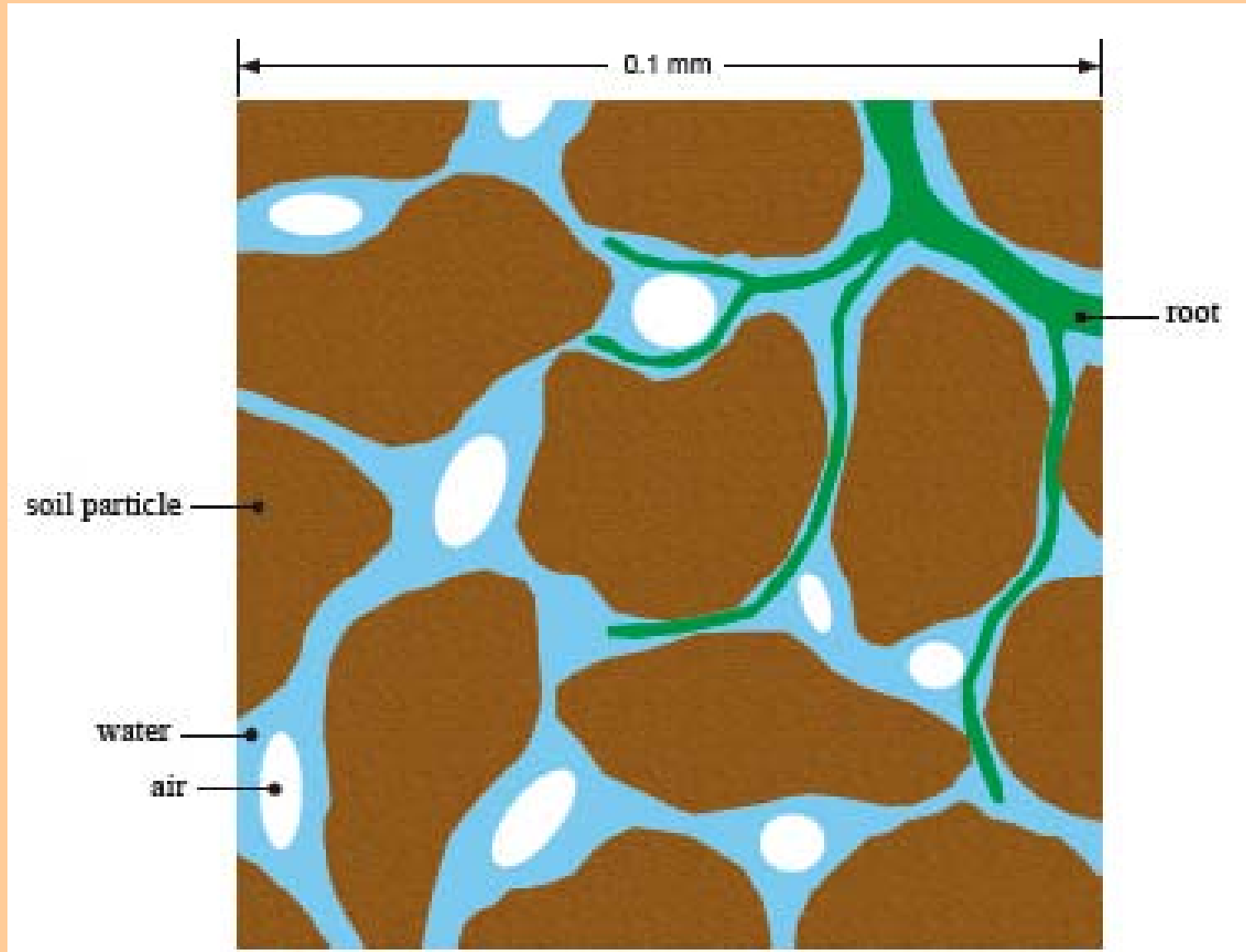


Soil Profile



# *Soil pore space*

- Pore space = space between all soil particles
- Macropores – larger spaces normally occupied by air
- Micropores – smaller spaces usually filled with water



# *Soil texture*

- Relative proportion of sand to silt to clay – based on particle size and reactivity
- Particle sizes arbitrarily chosen
- Applies only to portion of soil smaller than 2mm
- Doesn't include gravel, stones, boulders, etc.

# Soil Properties: Texture

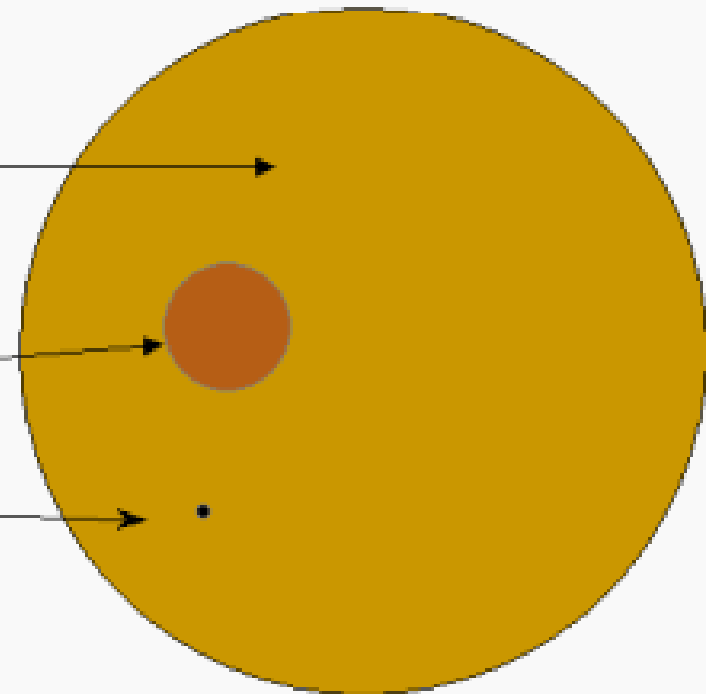
**Soil Texture:** The relative proportions of sand, silt, and clay particles in a mass of soil (material less than 2mm in size).

Very Coarse Sand = 2 to 1 mm

Very Fine Sand = 0.1 to 0.5 mm

Silt = 0.05 to 0.002 mm

Clay = < 0.002 mm







# *Sand*

- Particle size 0.05mm – 2.0mm
- Skeleton/backbone of soil
- Chemically inert
- Non-porous
- Formed by physical weathering
- Low water holding ability
- Good aeration
- Poor fertility

# *Silt*

- Particle size 0.002mm – 0.05mm
- Chemically inert
- Non-porous
- Formed by physical weathering
- Good water holding ability
- Poor aeration
- Poor to moderate fertility

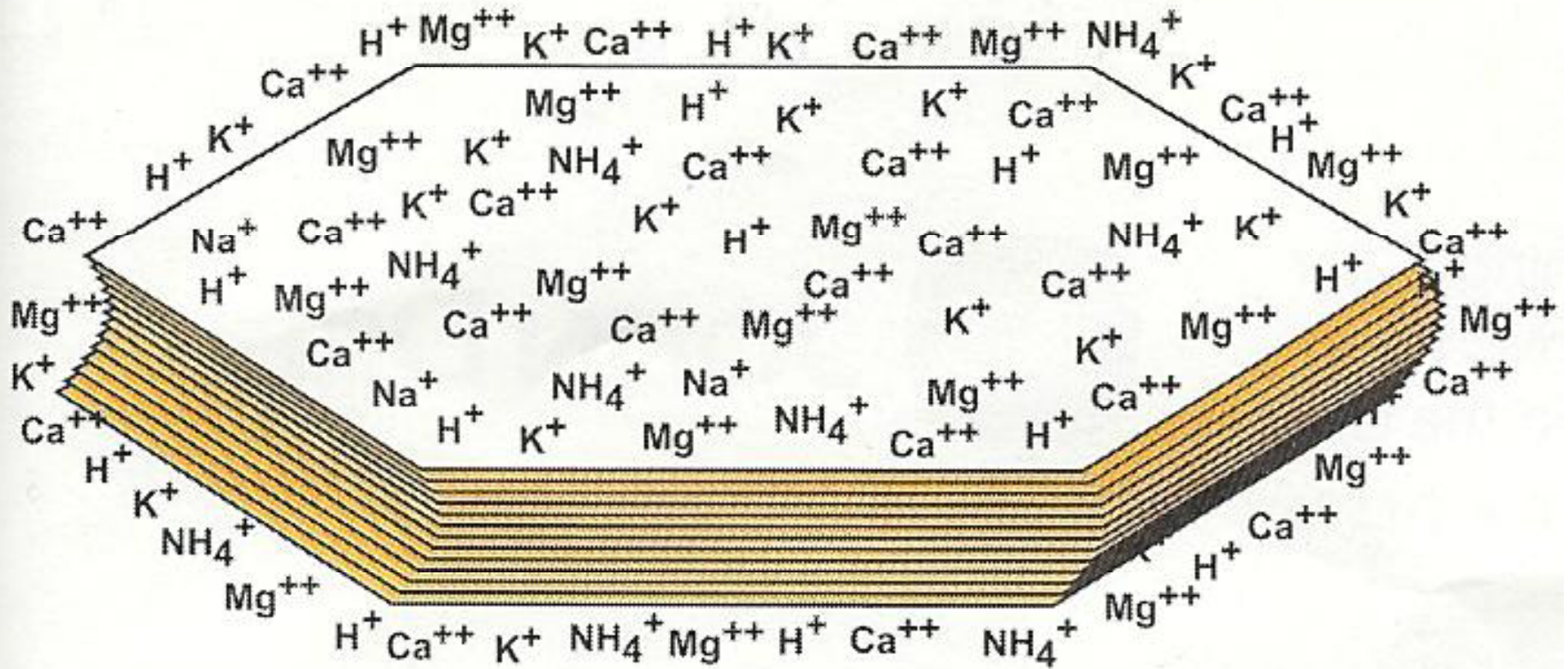
# *Clay*

- Particle size  $<.002\text{mm}$  (lots of surface area)
- Chemically active
- Porous, flat crystals
- Formed by chemical/freeze-dry weathering
- Expands/contracts when wet/dry
- Great water holding ability
- Poor aeration
- Good fertility
- Negatively charged active sites

Figure 1

Paul Marks

# CLAY PARTICLES



Negatively charged clay particles  
shown with typical plate-like  
appearance and swarm of adsorbed cations

# *Soil structure*

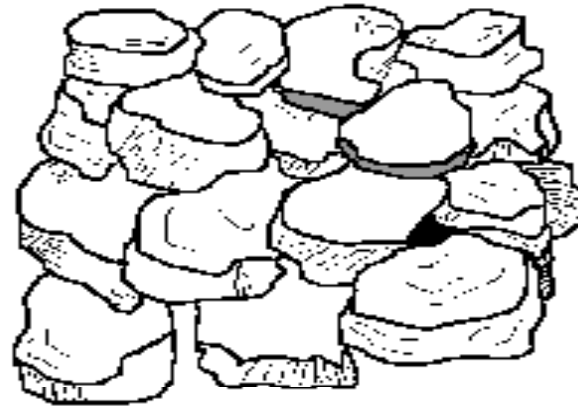
- Arrangement of soil particles into aggregates
- Each soil has characteristic shape/structure
- Sand, silt, clay and organic matter all clumped together by cementing by microorganisms



Jim Urban



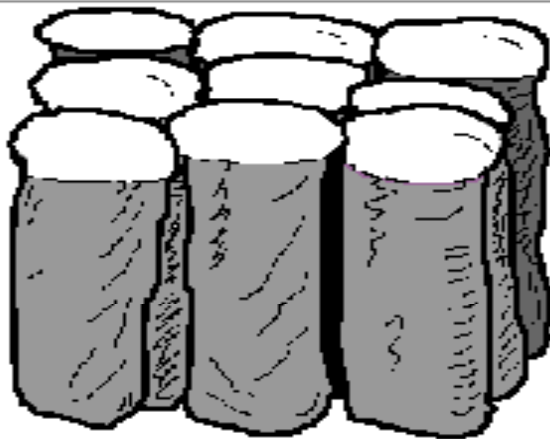
**Granular:** Resembles cookie crumbs and is usually less than 0.5 cm in diameter. Commonly found in surface horizons where roots have been growing.



**Blocky:** Irregular blocks that are usually 1.5 - 5.0 cm in diameter.



**Prismatic:** Vertical columns of soil that might be a number of cm long. Usually found in lower horizons.



**Columnar:** Vertical columns of soil that have a salt "cap" at the top. Found in soils of arid climates.



**Platy:** Thin, flat plates of soil that lie horizontally. Usually found in compacted soil.



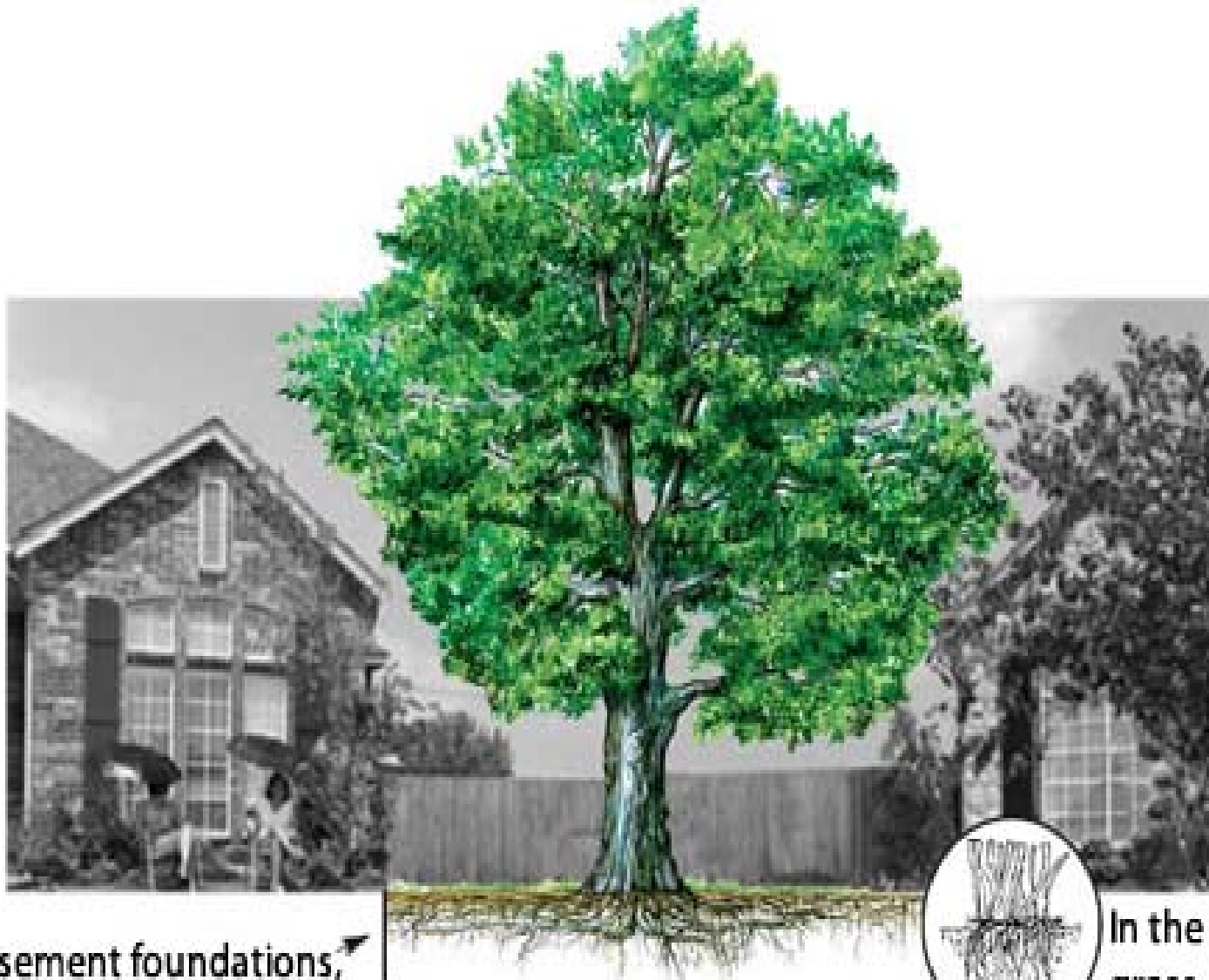
**Single Grained:** Soil is broken into individual particles that do not stick together. Always accompanies a loose consistence. Commonly found in sandy soils.





# ***Bulk Density***

- Weight/volume ( $\text{g/cm}^3$  or  $\text{lb/ft}^3$ )
- Used to assess pore space adequacy
- Reflects degree of soil compaction



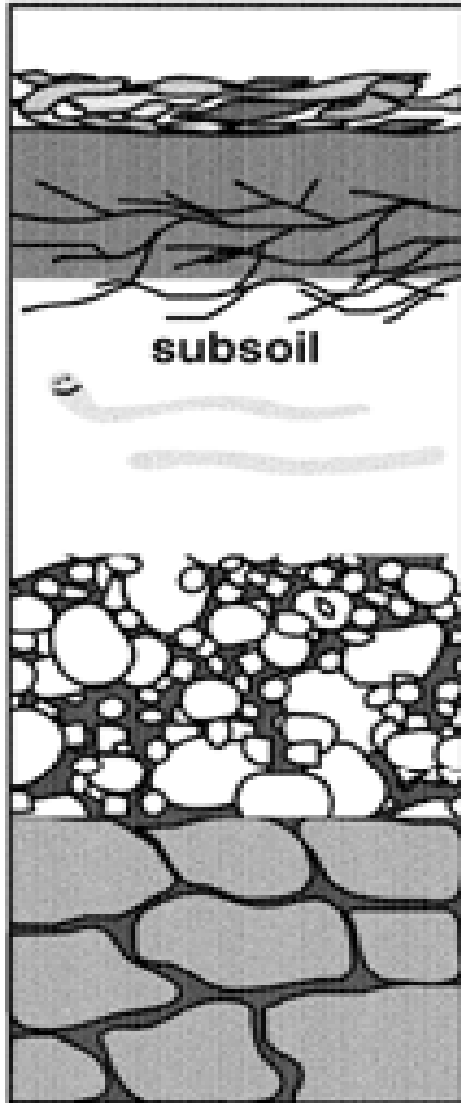
Basement foundations, driveways, utilities and streets restrict the root system and limit available resources.

In the urban environment grass competes with trees for water and minerals. This significantly reduces the capacity of a yard to support a tree.

# *Compaction*

- Disruption of soil particles, aggregates and accompanying pore spaces
- Increase in bulk density and decrease in total pore space
- Typically the result of traffic (foot & vehicular)
- Clay soils are more easily compacted than sandy soils
- Moist soils more easily compacted than dry soils

## Undisturbed native soil

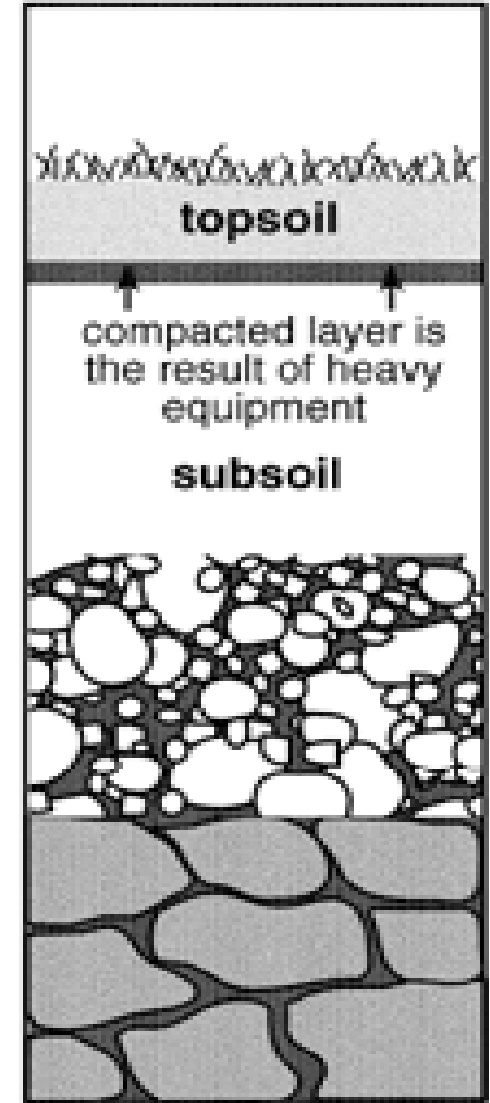


spongy surface layer

insects aerate the soil and turn debris into usable organic matter

bedrock

## Typical urban soil



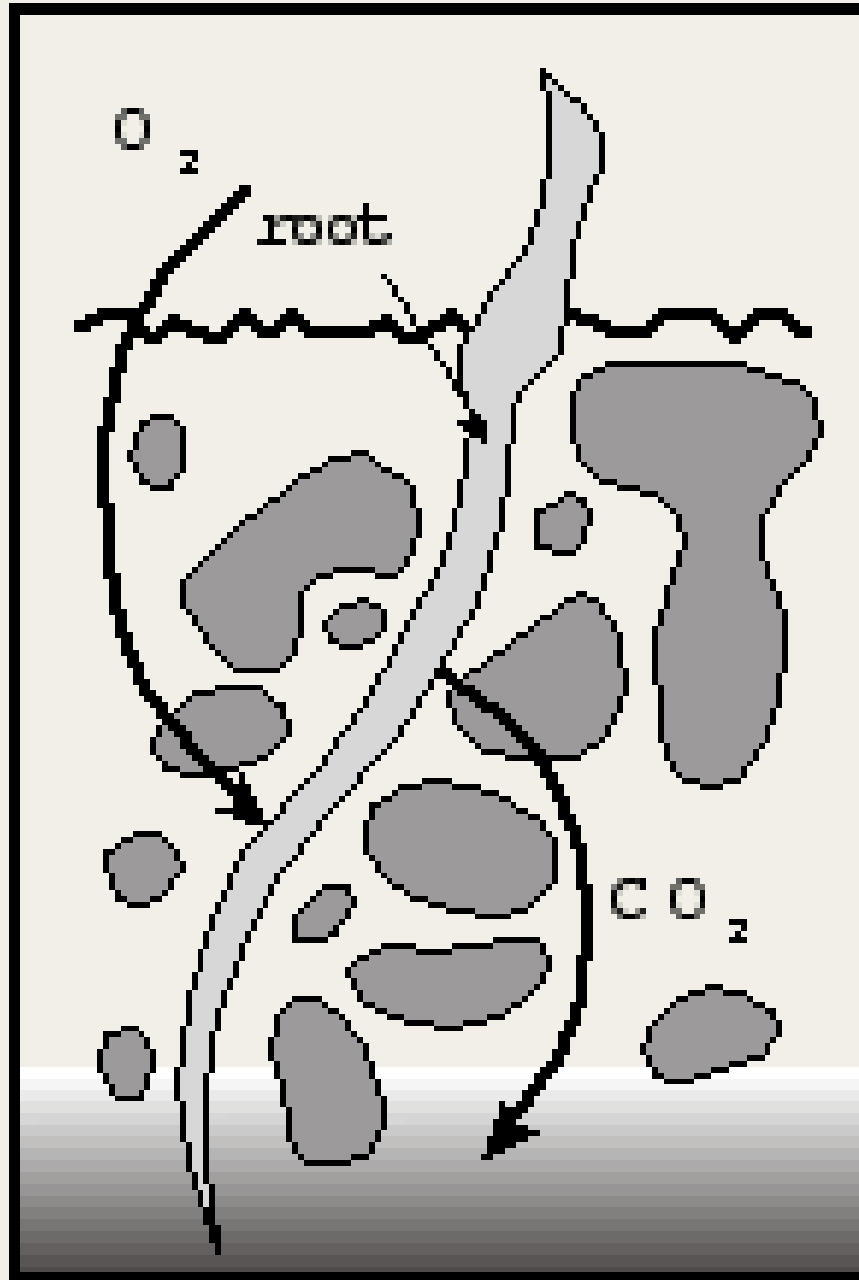
surface compacted for sod preparation

low organic matter

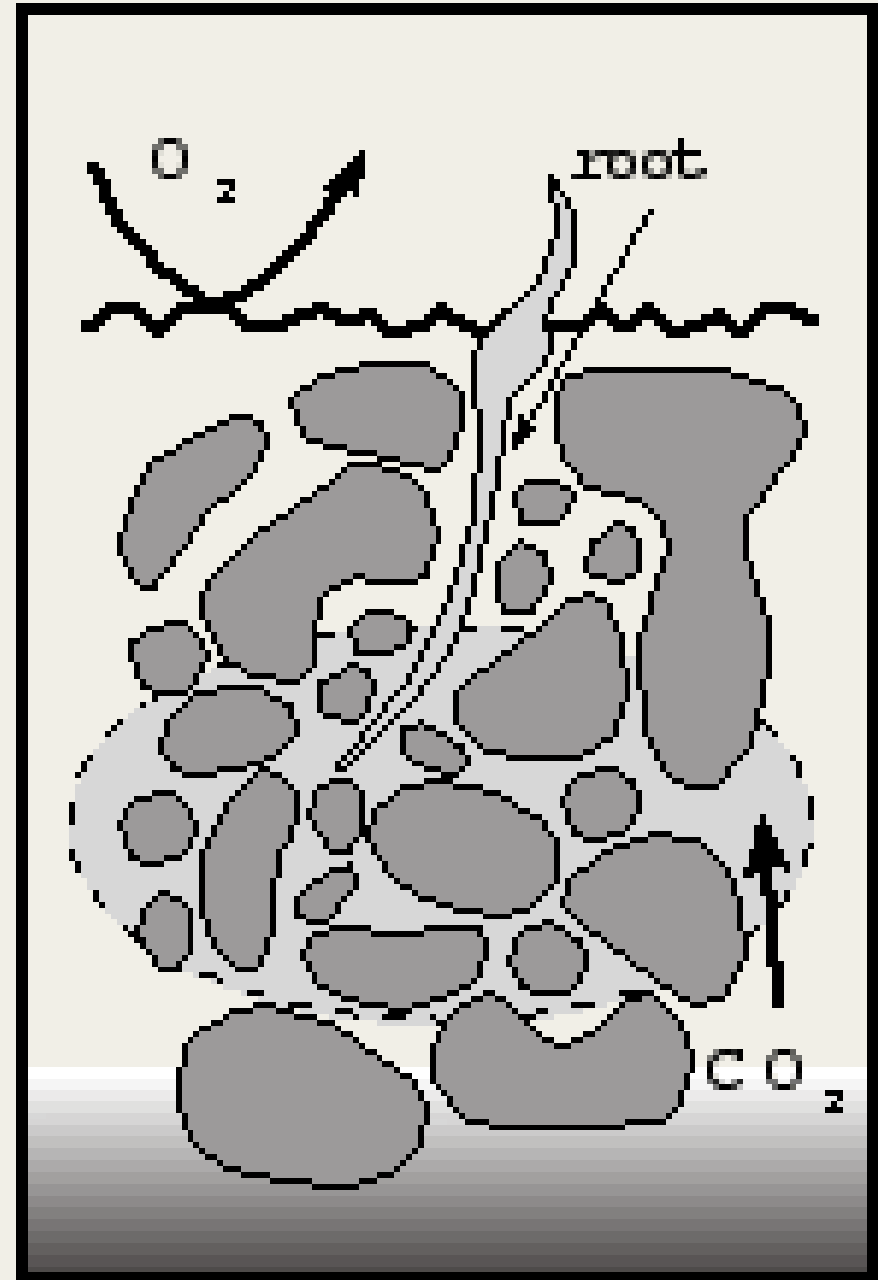
topsoil is often from another location and may not "match" the subsoil

bedrock

## Loose Soils

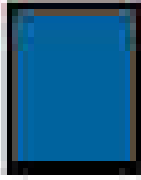


## Compacted Soils

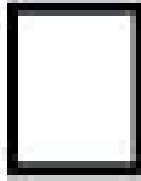




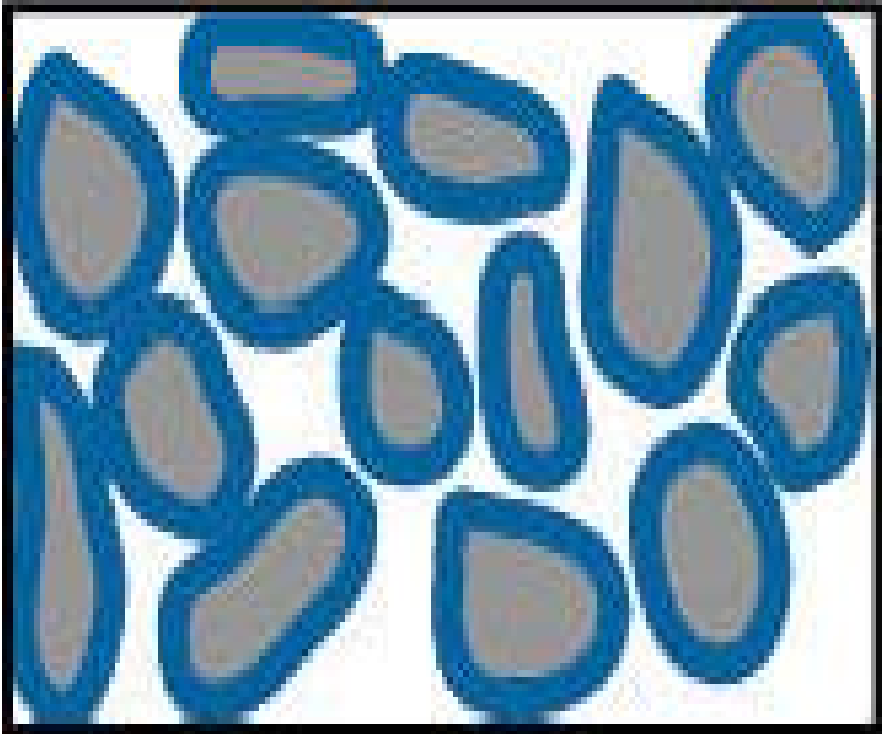
Soil Particles



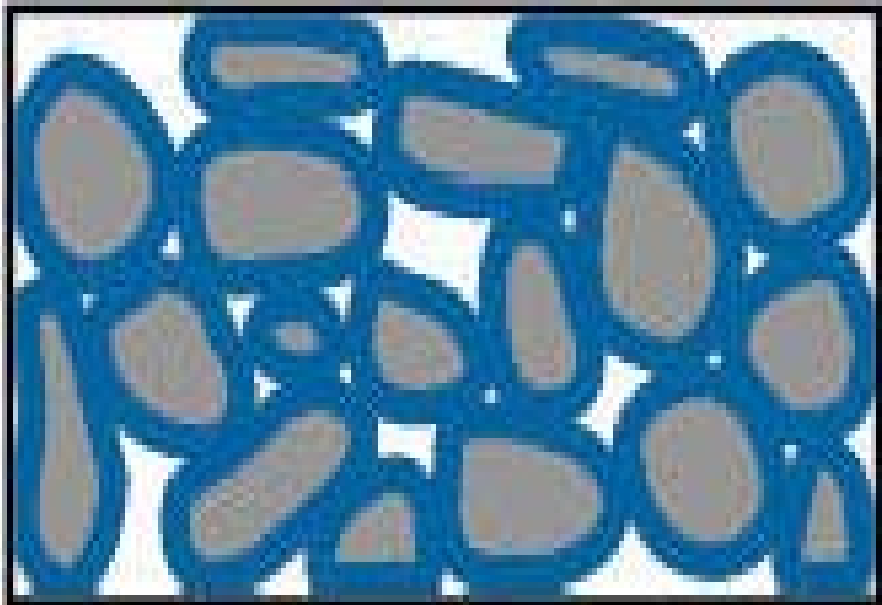
Water



Air



Non-compacted

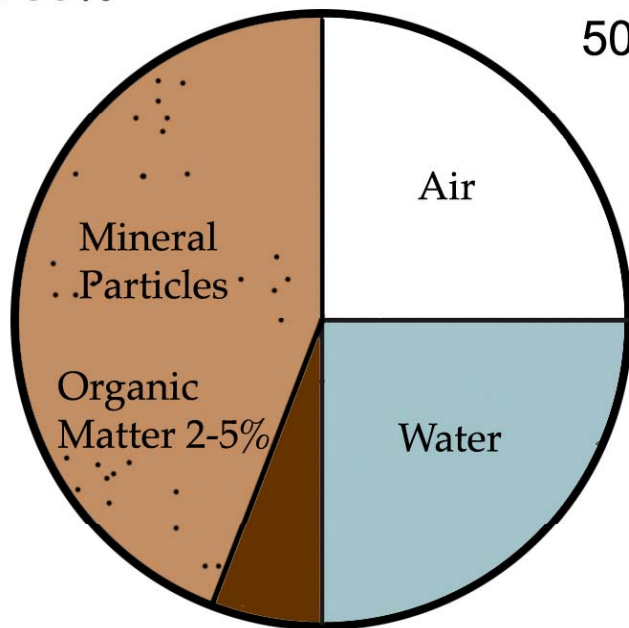


Compacted



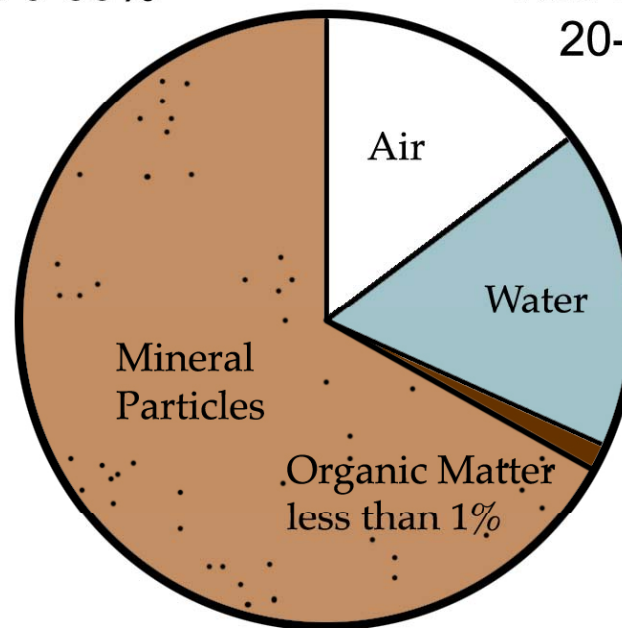
Jim Urban

Solid 50%      Void Space 50%



IDEAL FOREST SOILS

Solid 70-80%      Void Space 20-30%

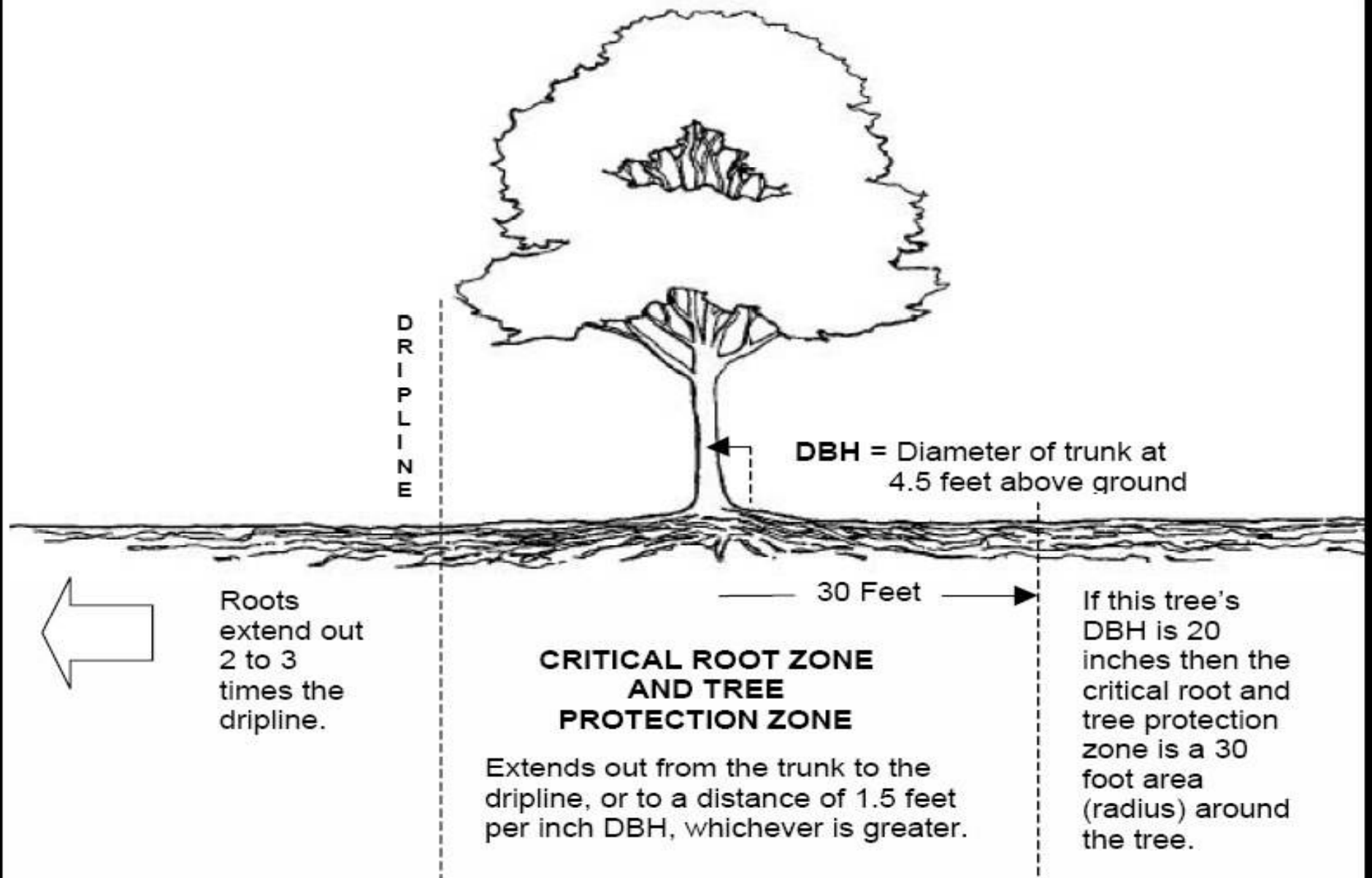


URBAN SOILS

As compaction increases, pore space for water and air decreases



Figure 1. CRITICAL ROOT ZONE





# *Organic matter*

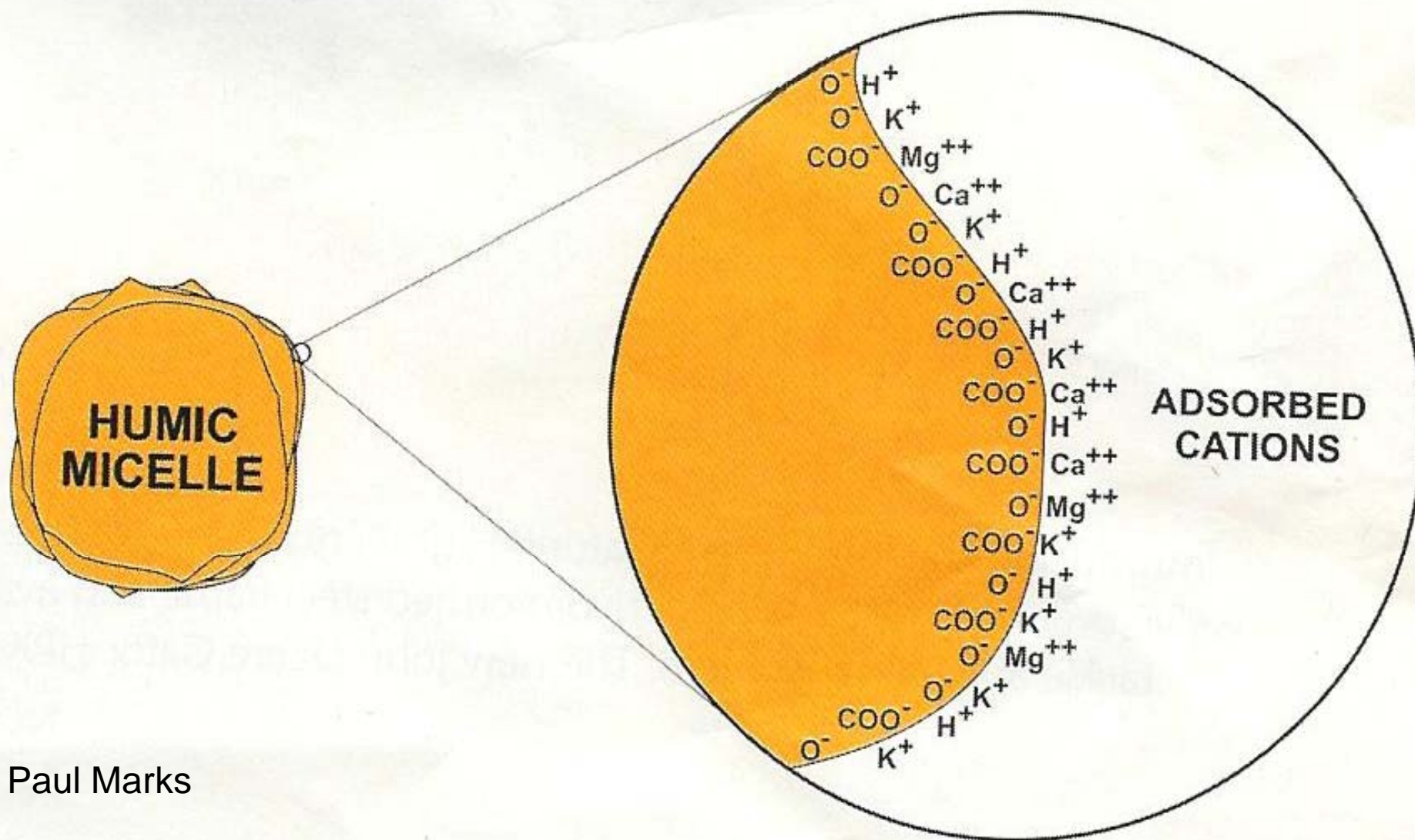
- Accumulated decomposition products of dead plants and animals
- Carbohydrates, nitrogenous compounds (proteins, amino acids), lignin, fats/resins/waxes
- End stage of decomposition is humus

# *Humus*

- Chemically active
- Porous
- Improves water/nutrient holding capacity of coarse textured soils
- Reduces or moderates water/nutrient holding capacity of fine textured soils

Figure 3

# ORGANIC COLLOID



Paul Marks

# *Soil water*

- Ability to hold or leach water varies from soil to soil
- Influenced by soil texture, vegetation, cultivation, amendments, cultural methods (mulching, etc.)
- Volume occupied by water = volume not occupied by air
- Important because all soil chemistry takes place in solution

# ***Soil water***

- Gravitational – drains from macropores under force of gravity
- Capillary – remains in micropores after gravitational effects – available for plant use
- Hygroscopic – exists as vapor – unavailable to plants



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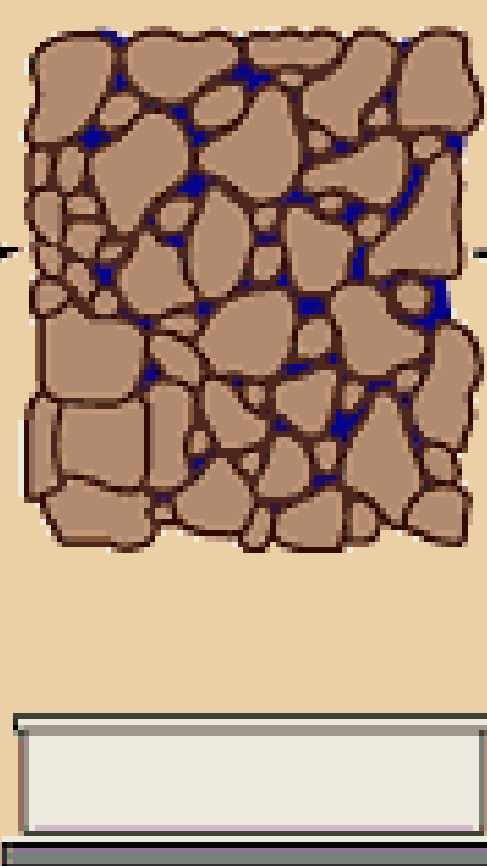






### **Saturation**

All pores are full of water. Gravitational water is lost



### **Field Capacity**

Available water for plant growth



### **Wilting Point**

No more water is available to plants

# *Soil water*

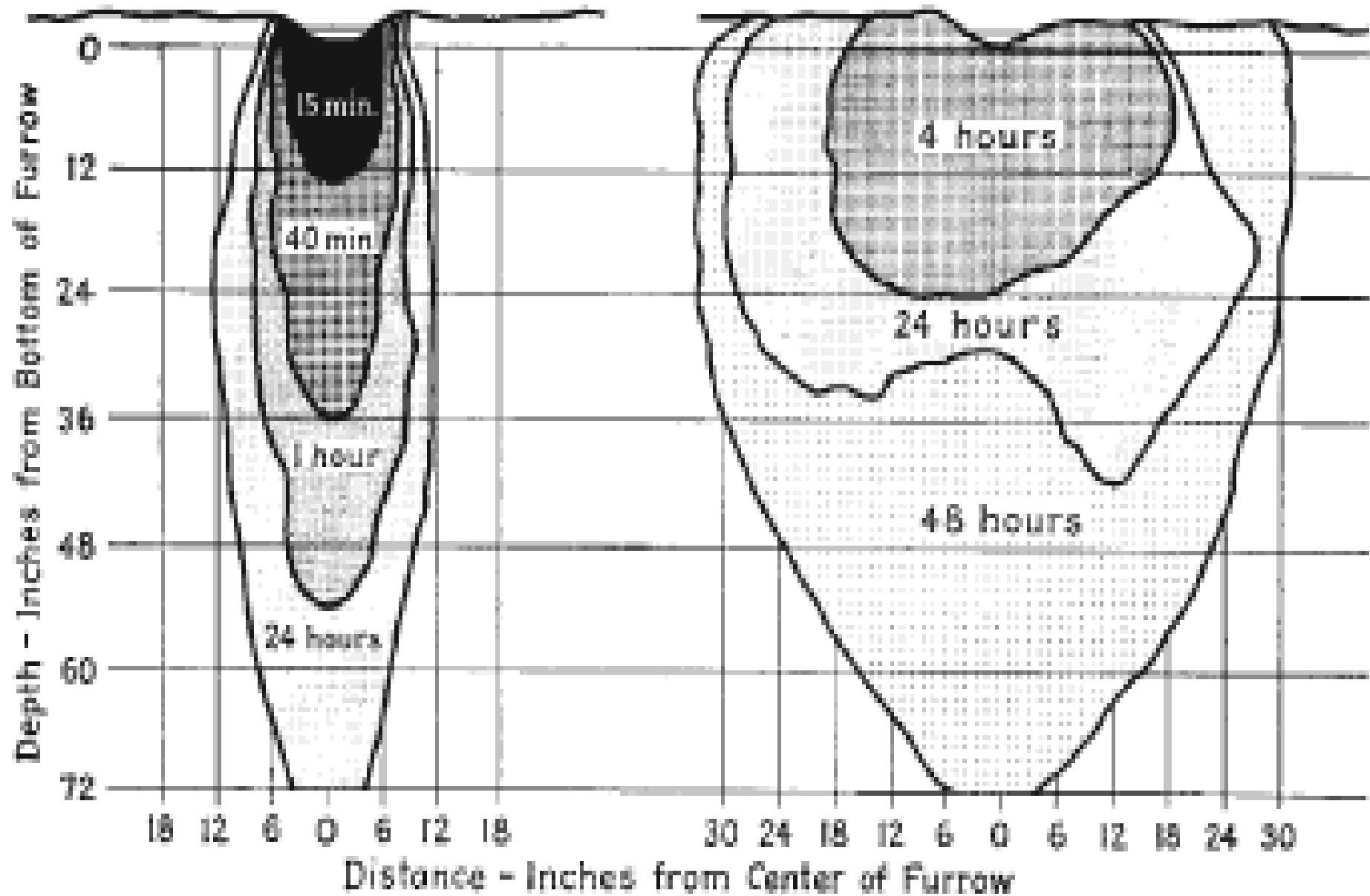
- Saturation – all pore space occupied by water – flooding can occur
- Field capacity – after gravitational water has passed through profile
- Wilting point – no water left for plant use – plants begin to wilt
- Infiltration rate – speed with which water moves through a soil profile
- Evapotranspiration – moisture loss from soil and plants

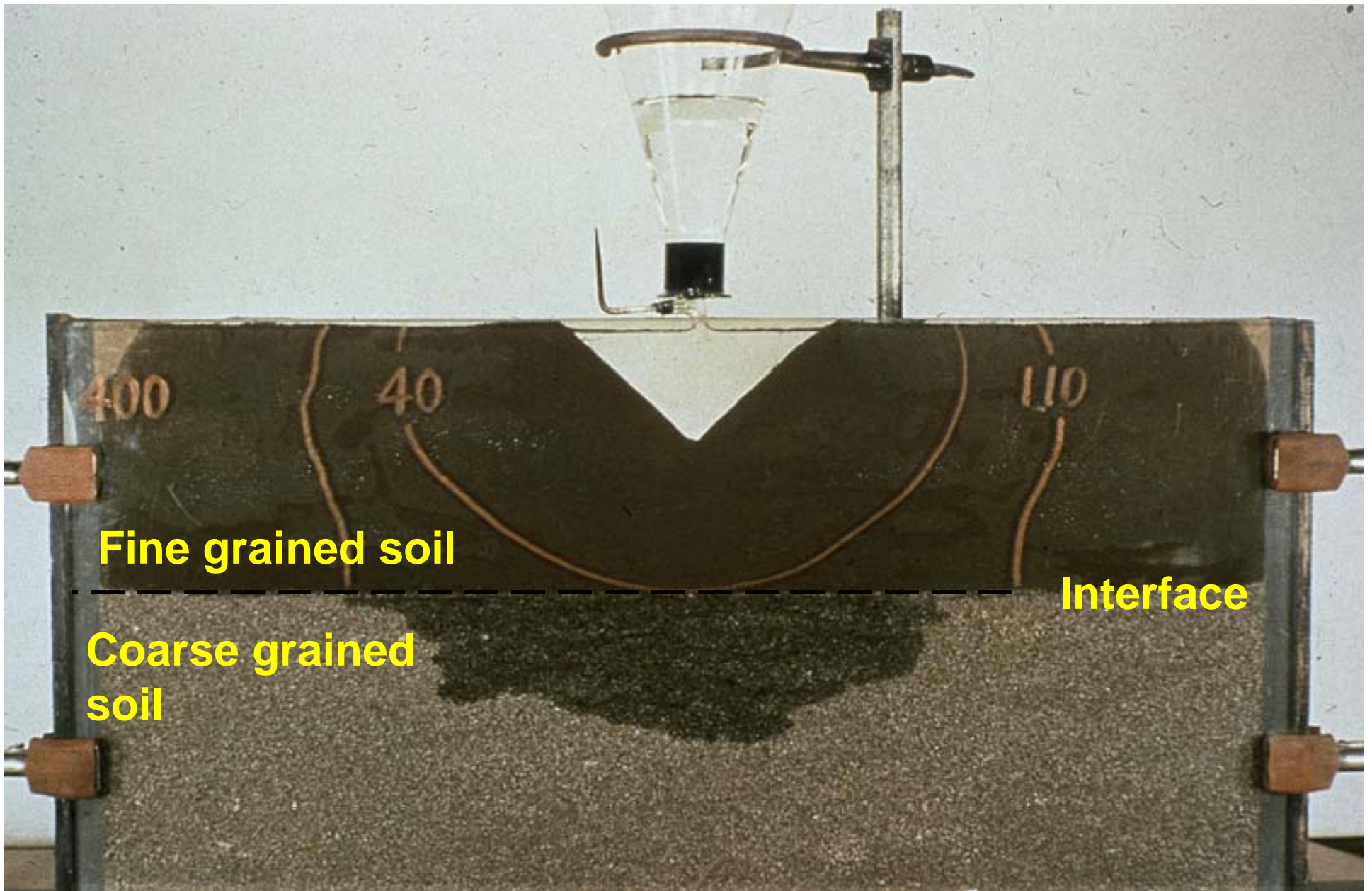


University of Arizona. Credit: John C. Palumbo

### SANDY LOAM

### CLAY LOAM

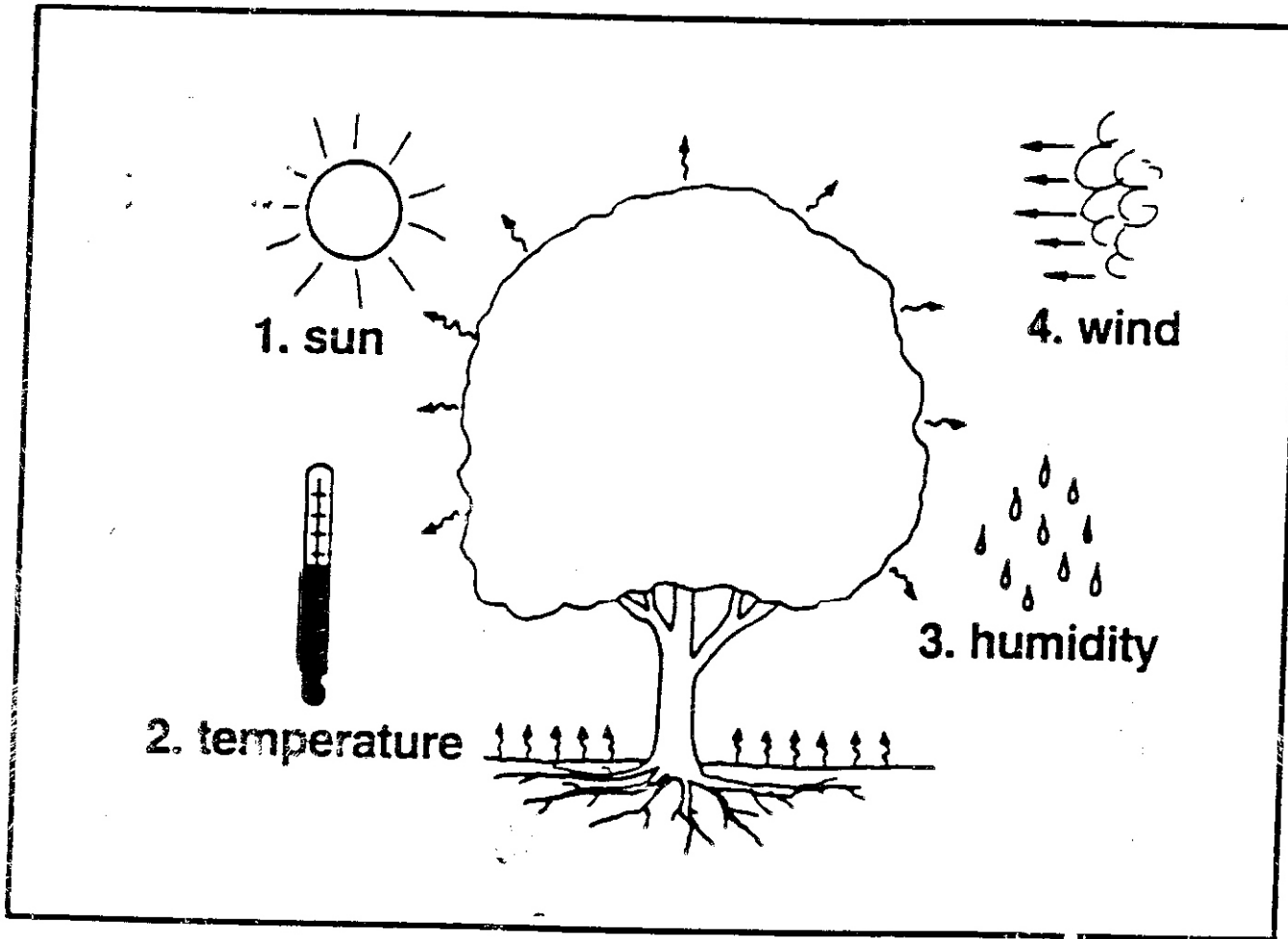




**Fine grained soil**

**Interface**

**Coarse grained  
soil**



**Fig. 4.6** Evapotranspiration rates are dependent on environmental conditions including light, temperature, wind, and humidity.

# *Percolation test*

- Dig one foot deep pit
- Fill pit with water and allow to drain (to saturate soil before the actual test)
- Within 24 hours, refill pit and record time needed for each inch of water to drain
  - poorly drained =  $< 1$  in/hr
  - moderately drained =  $1 - 6$  in/hr
  - excessively drained =  $> 6$  in/hr



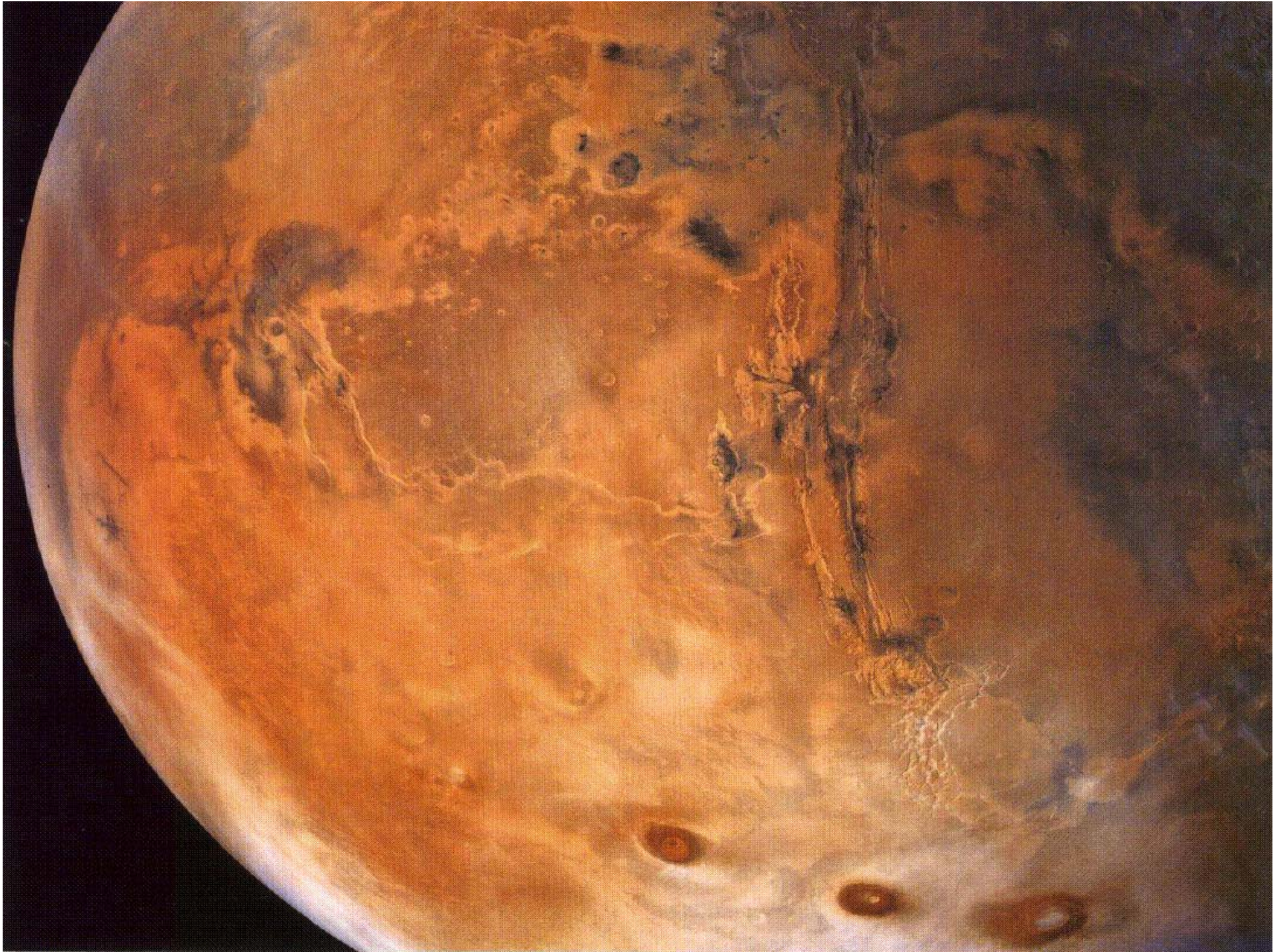
# *Irrigation*

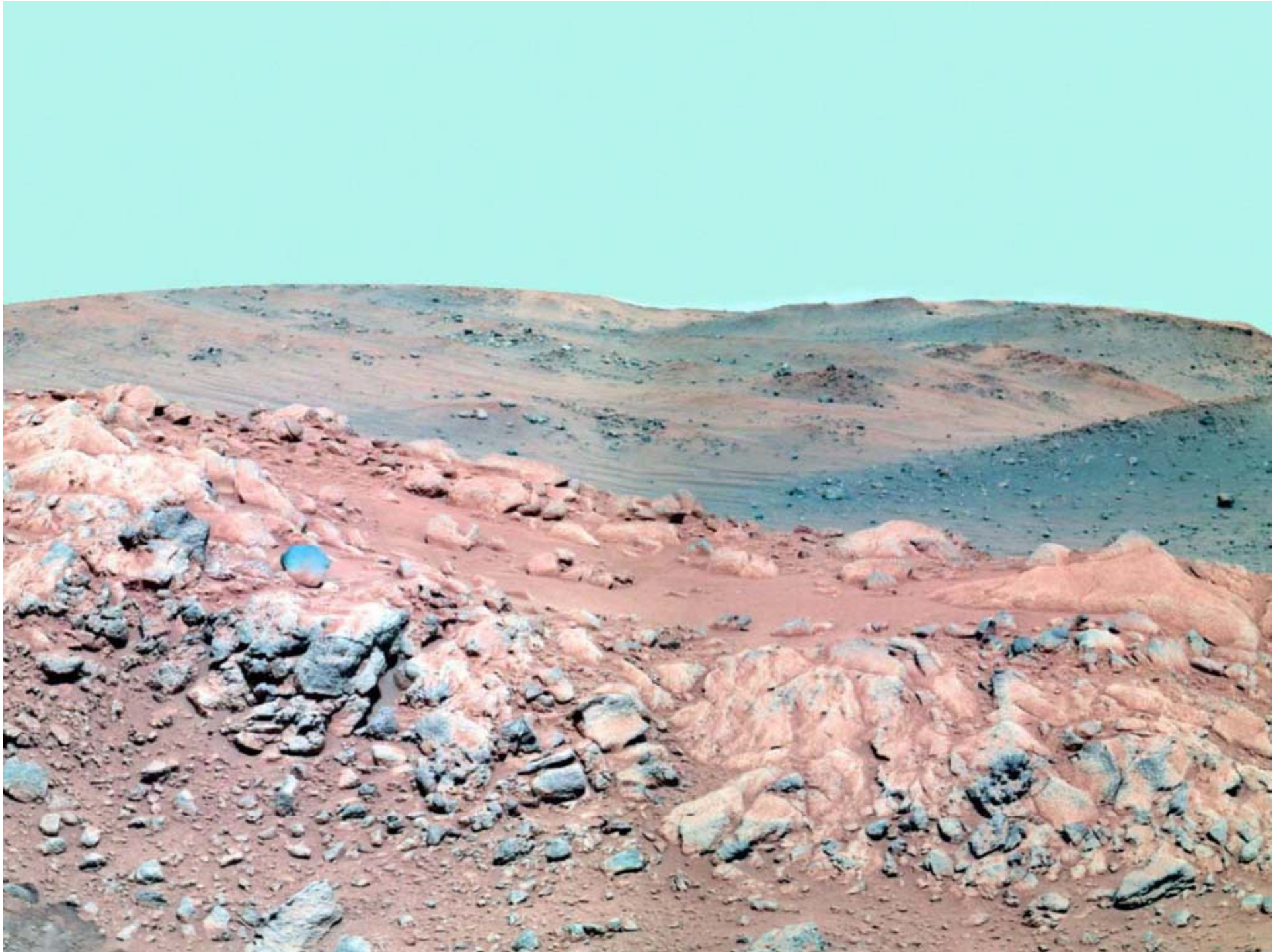
- Sprinkler irrigation – most common method; long duration/infrequent intervals is best strategy; greater risk for evaporation
- Drip irrigation – low volumes applied over time directly to soil; helps conserve water
- Minimum irrigation – least amount of water needed to maintain plants; plant/soil dependant

# *Irrigation terms*

- Tensionometer – device that senses soil moisture; helpful for irrigation scheduling
- Hydrogels – polymeric crystals that absorb and slowly supply water when incorporated in soil
- Antitranspirants – applied to foliage to limit moisture loss
- Mulches – placed on soil surface to limit moisture loss
- Desiccation – plants drying to death
- Phytotoxic – poisonous to plants







# *Soil biology*

- Microfauna

bacteria

fungi

algae

protozoans

nematodes

- Macrofauna/flora

insects

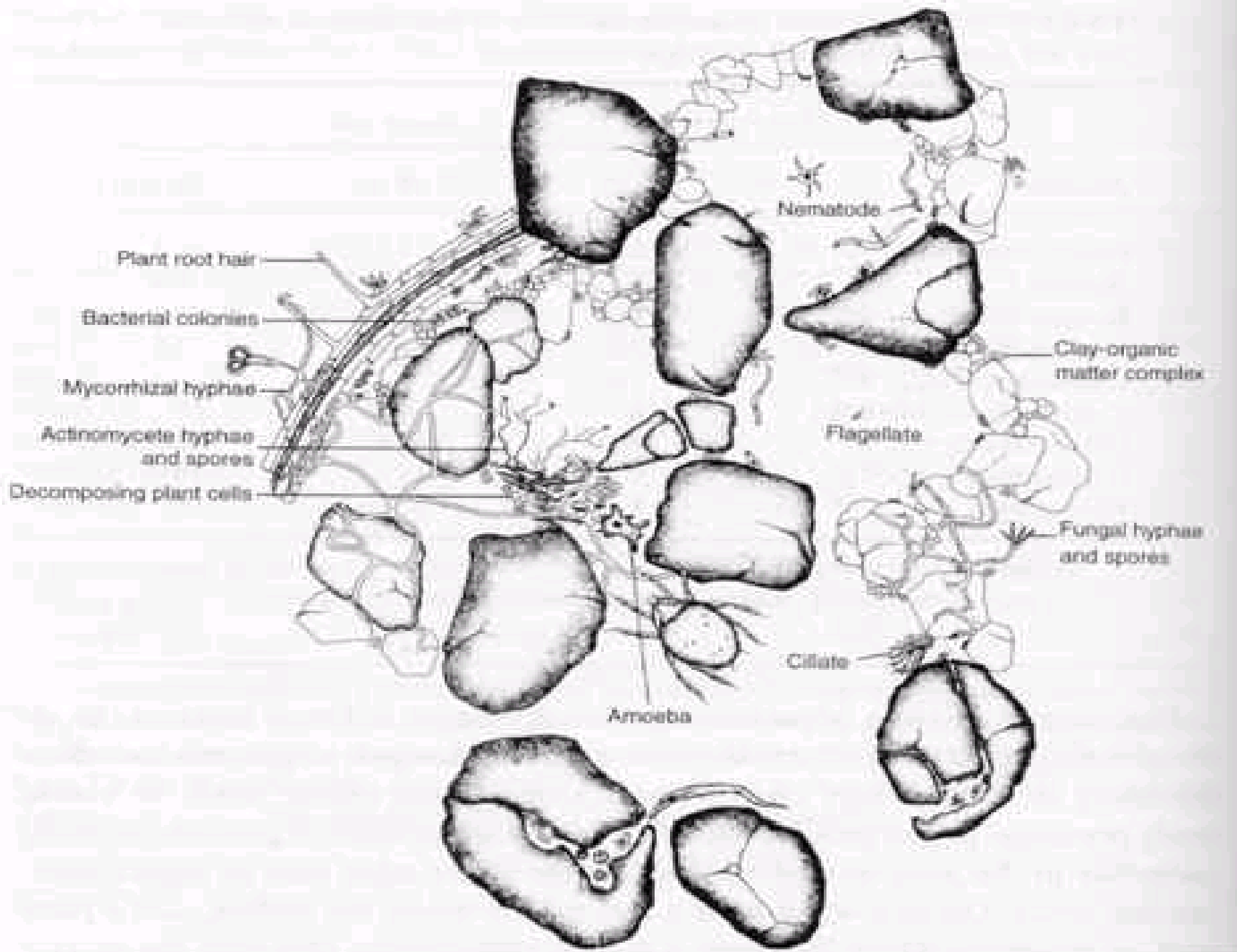
mites

earthworms

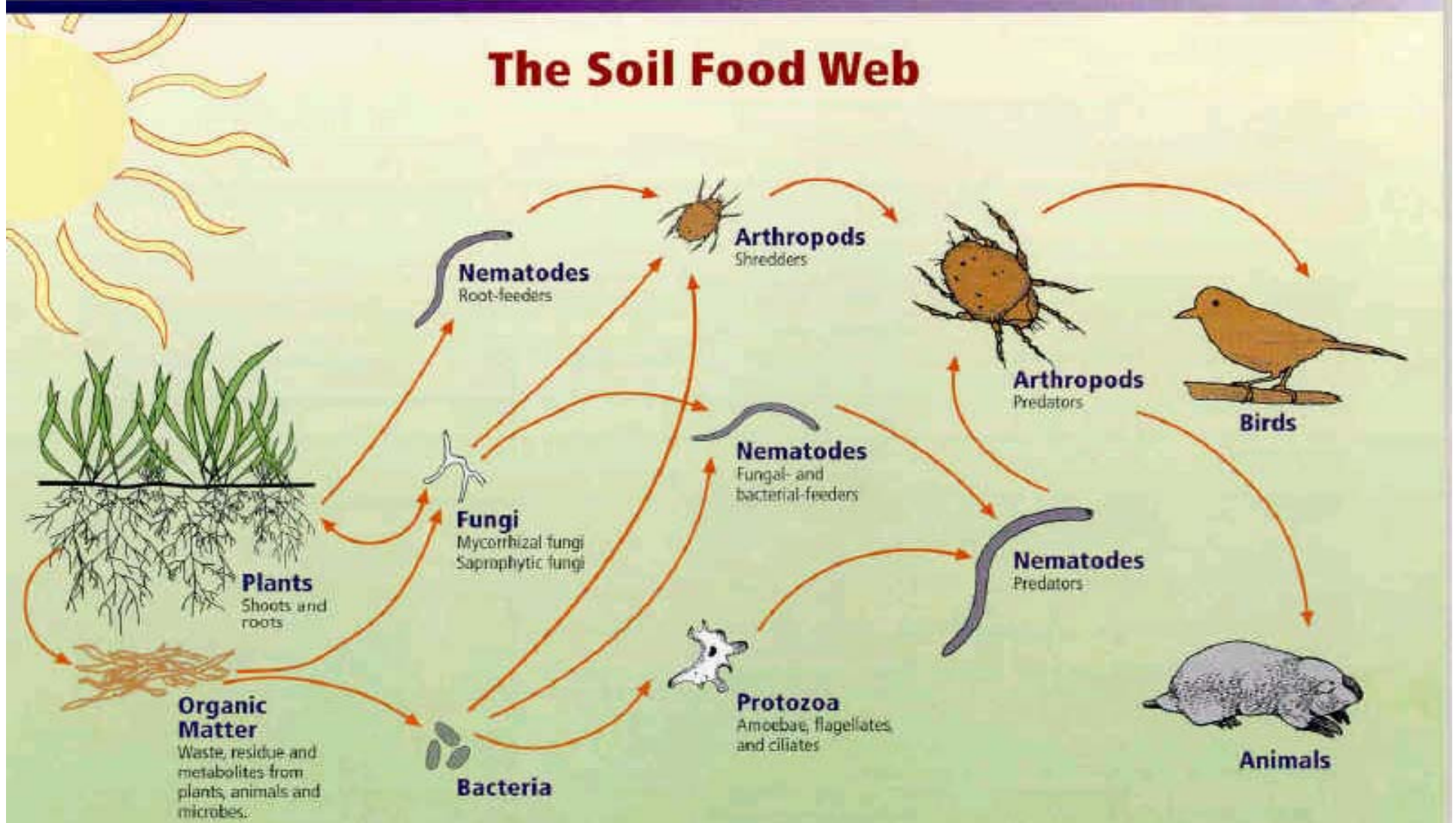
rodents

mycorrhizae

plant roots



# The Soil Food Web



**First trophic level:**  
Photosynthesizers

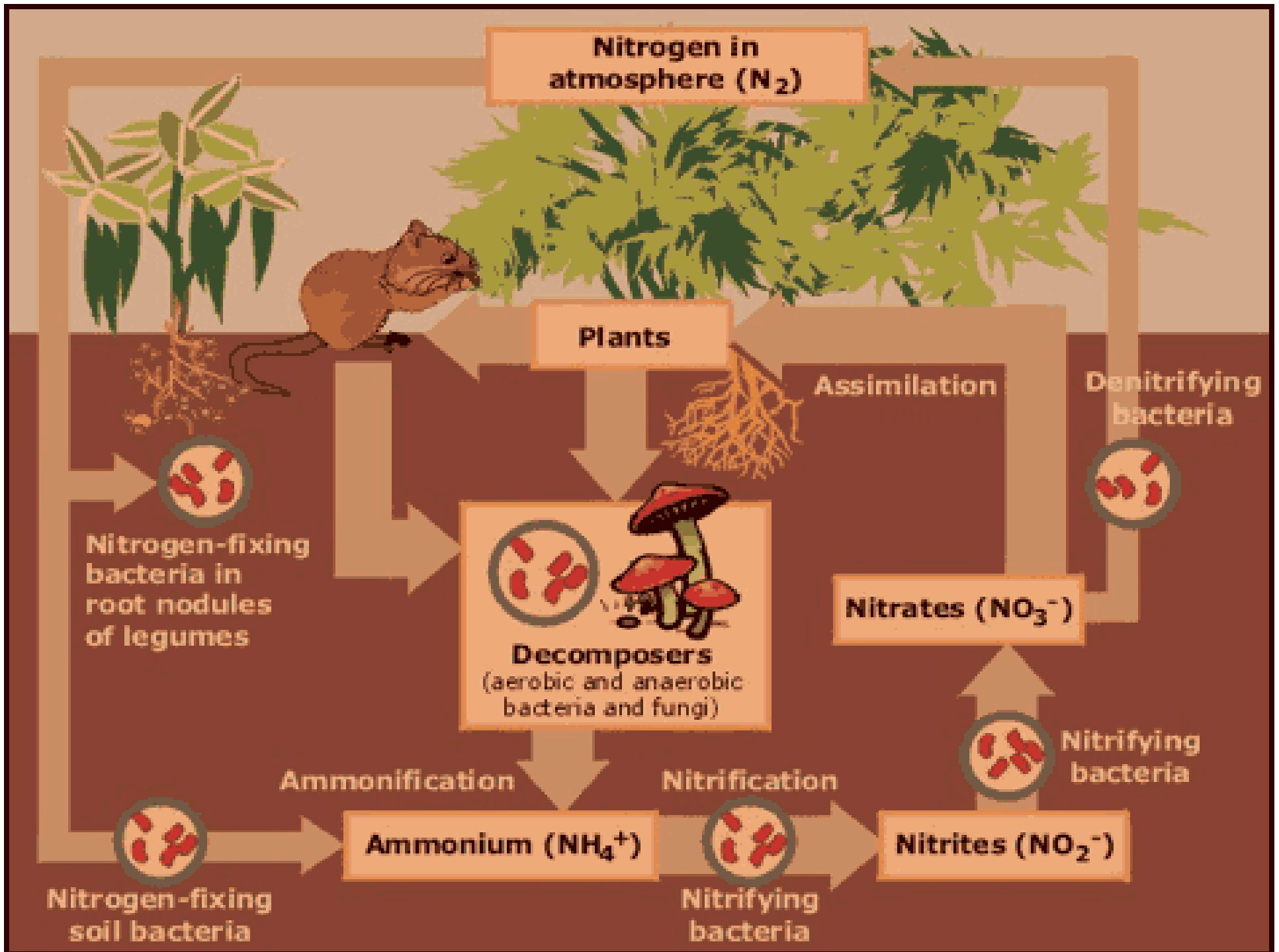
**Second trophic level:**  
Decomposers  
Mutualists  
Pathogens, parasites  
Root-feeders

**Third trophic level:**  
Shredders  
Predators  
Grazers

**Fourth trophic level:**  
Higher level predators

**Fifth and higher trophic levels:**  
Higher level predators





## Biomass of Soil Organisms in Four Ecosystems

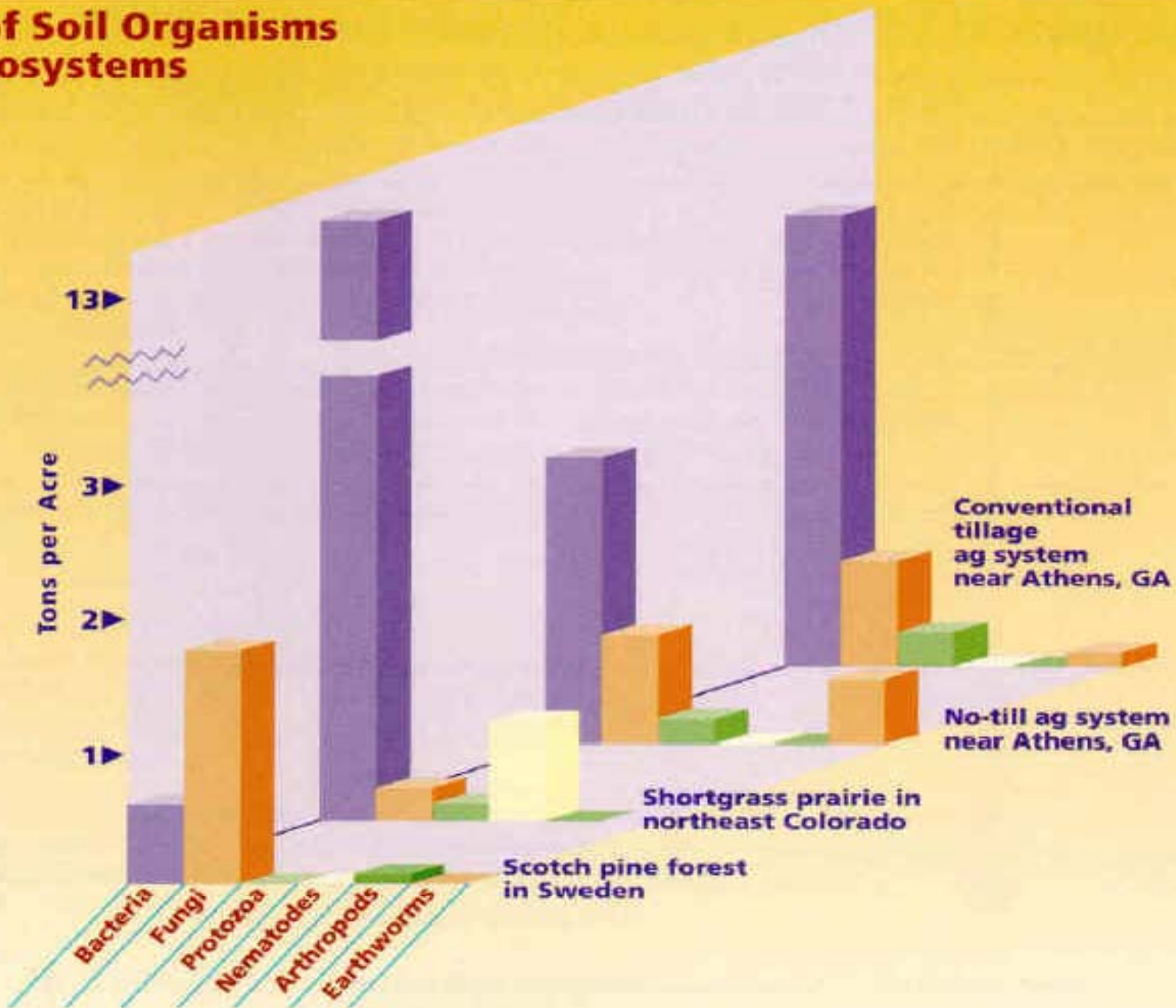


Figure 2

# *Microbial activity*

- Breakdown complex organic substances
- Consume carbon, nitrogen
- Nitrogen cycle carried out by microorganisms

# ***Mycorrhizae – fungal roots***

- Myco = fungal
- Rhizae = roots
- Exist in many natural settings
- Parasitize plant roots
- Improve roots' ability to take up moisture and nutrients (especially P)
- Example of symbiosis = mutually beneficial

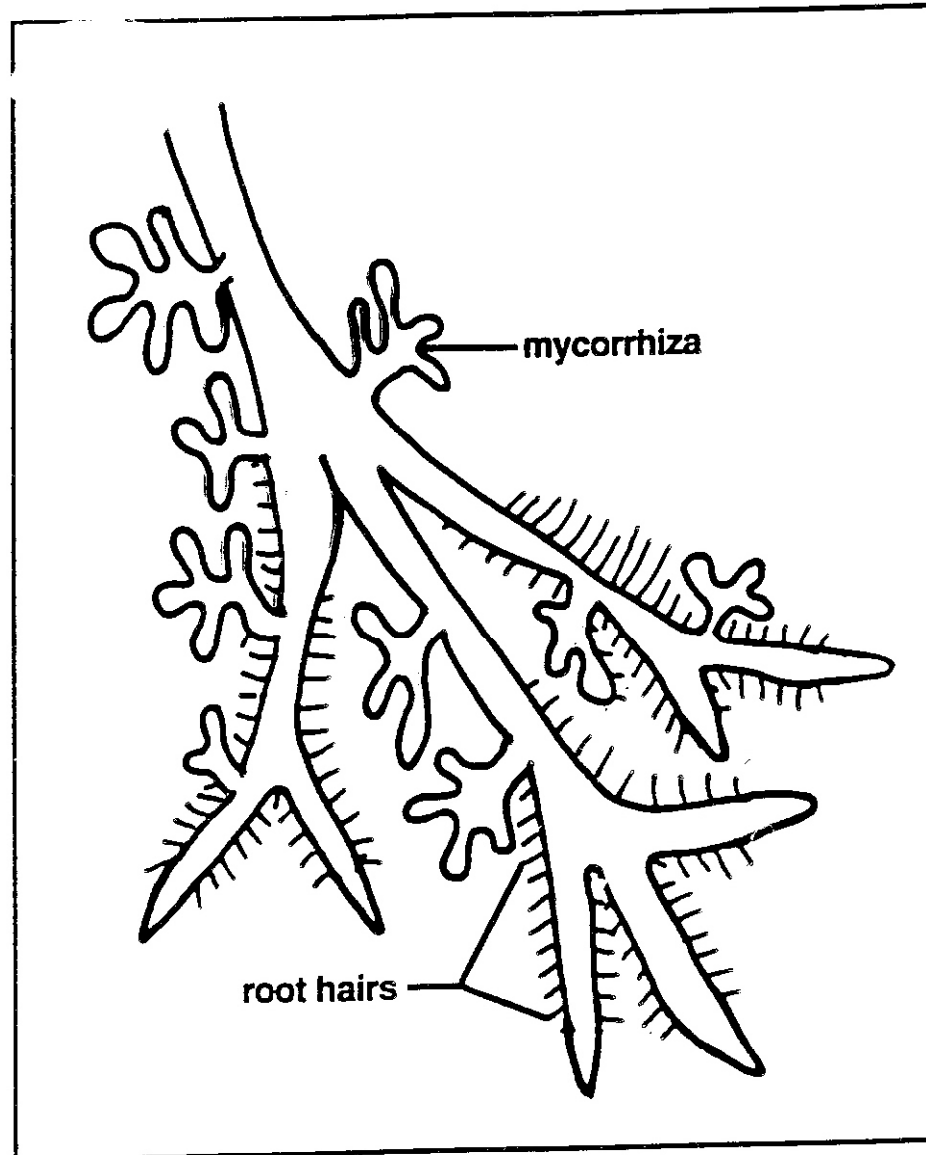
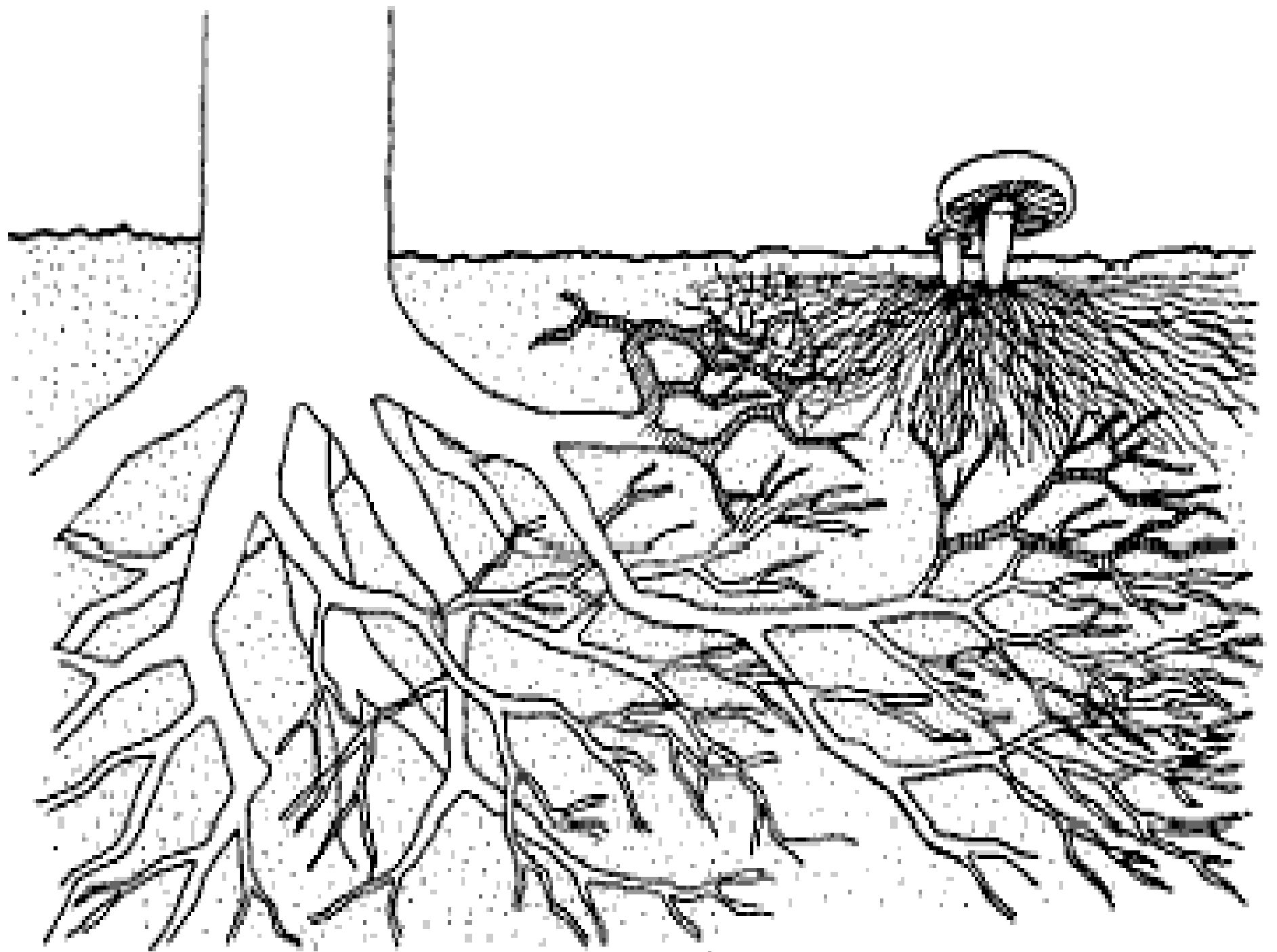


Fig. 3.6 Mycorrhizal roots aid the tree in the uptake of water and minerals.





# *Soil chemistry – surface reactivity*

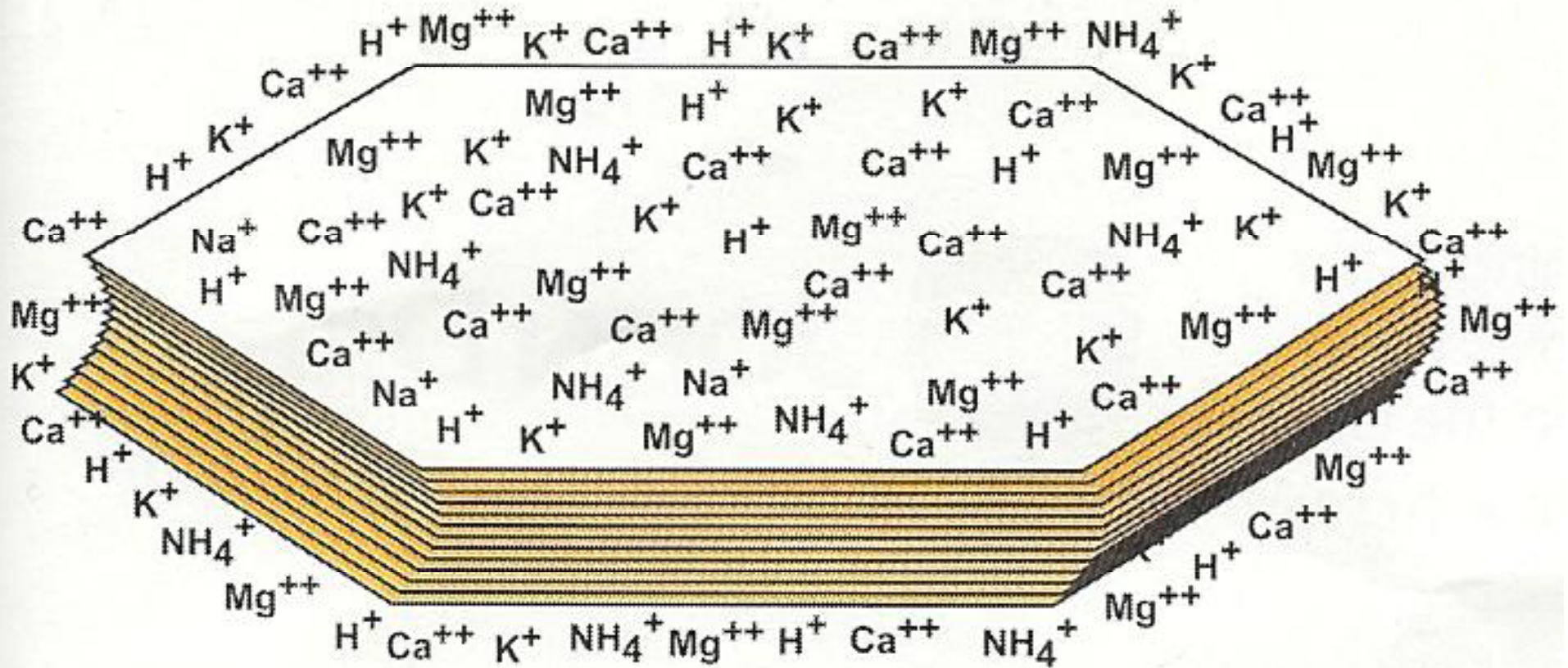
- Plants take up nutrients only in ionic form
- Ions = elements or collection of elements in solution and are positively or negatively charged
- Cations – positively charged ions (Ca, Mg, K, Na,  $\text{NH}_4$ , H; attached to soil (- charge)
- Anions – negatively charged ions (Cl,  $\text{CO}_3$ ,  $\text{NO}_3$ ,  $\text{SO}_4$ ); not attached to soil - at mercy of gravity



Figure 1

Paul Marks

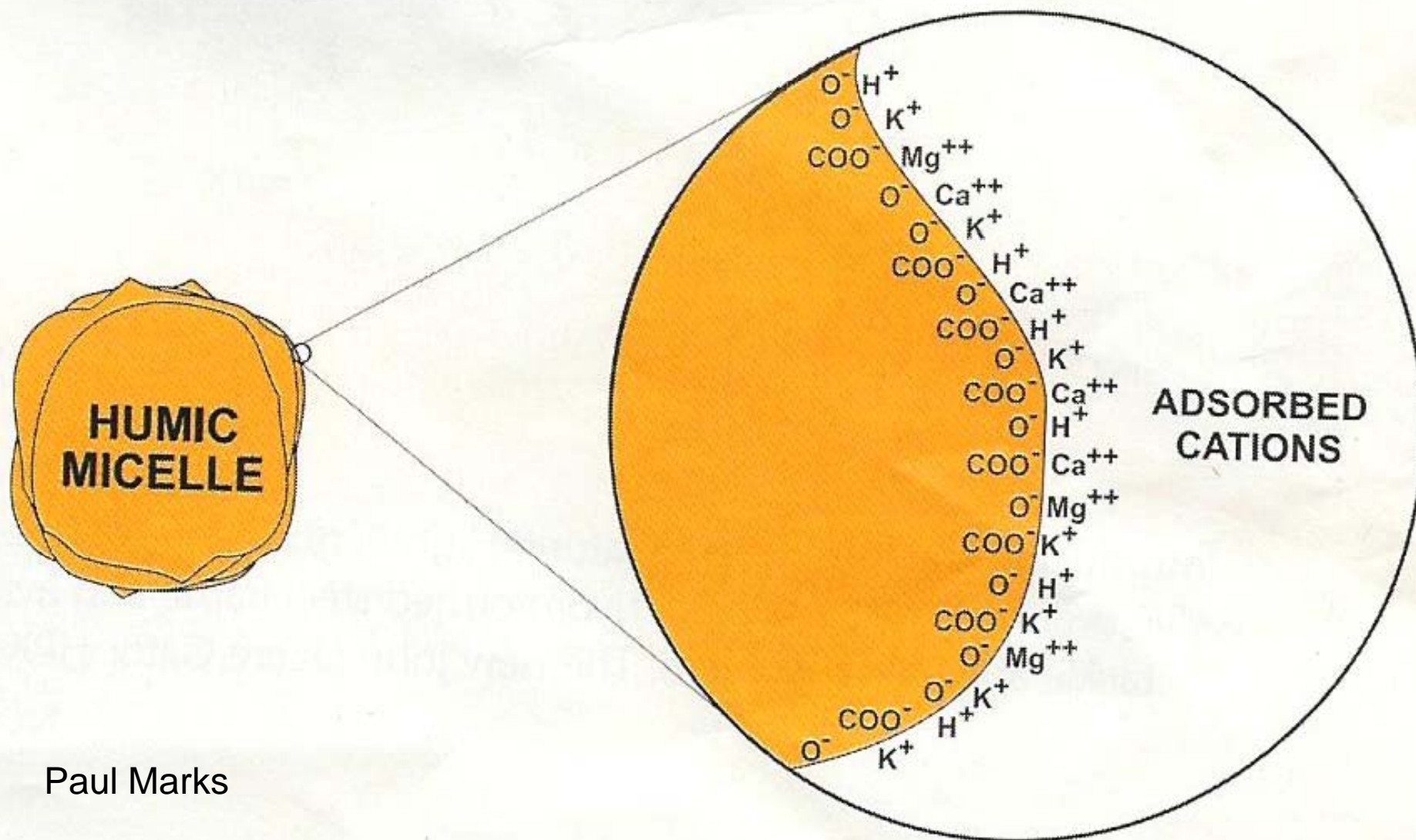
# CLAY PARTICLES



Negatively charged clay particles  
shown with typical plate-like  
appearance and swarm of adsorbed cations

Figure 3

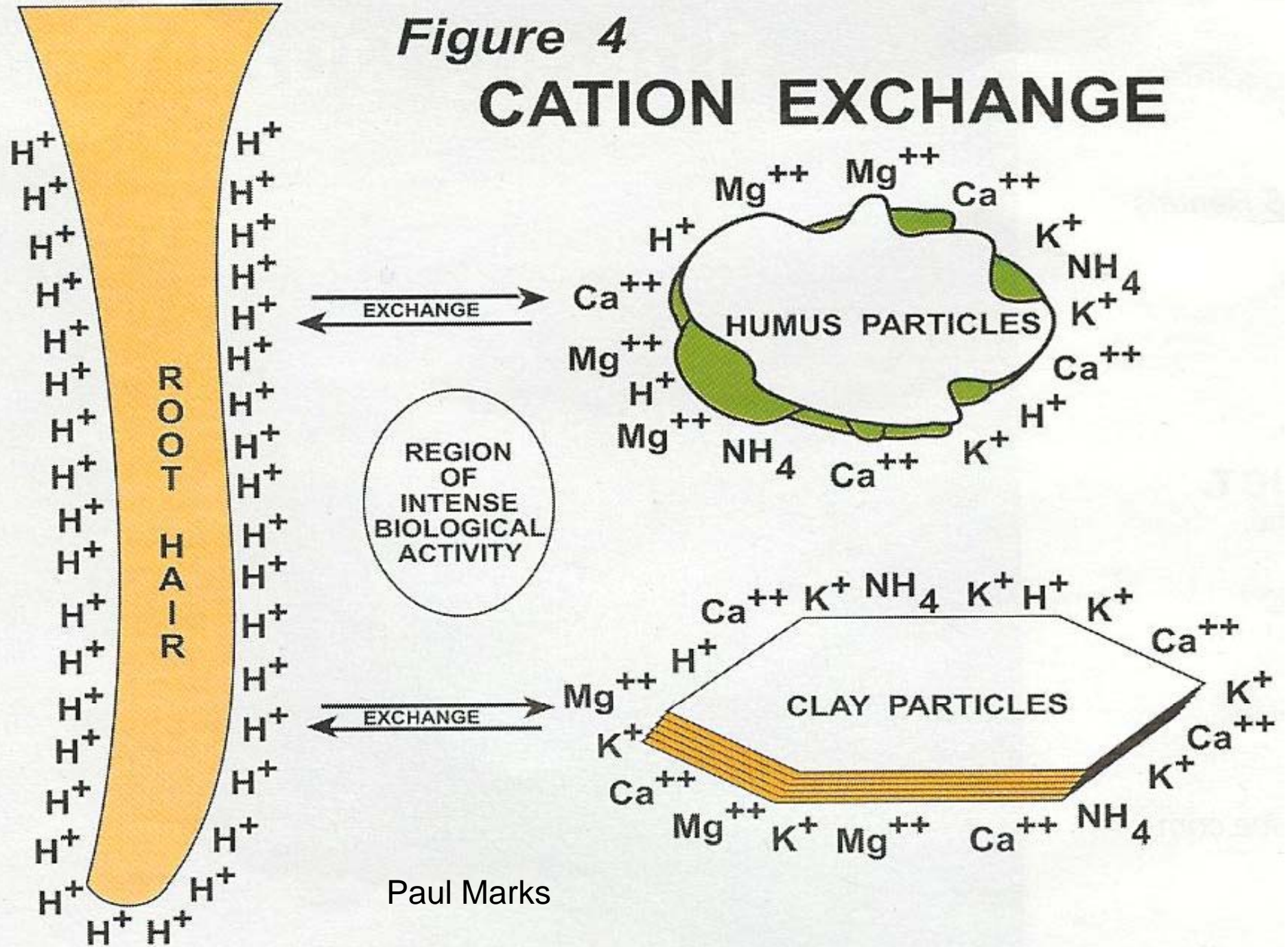
# ORGANIC COLLOID



Paul Marks

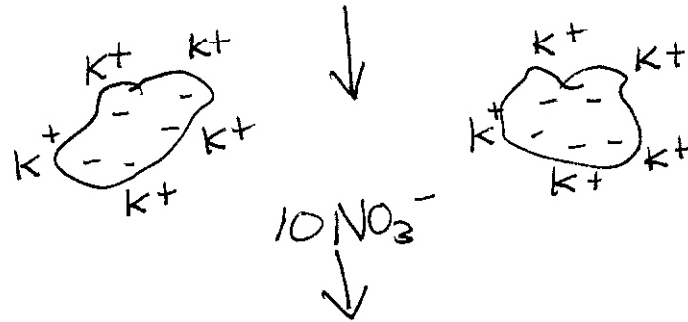
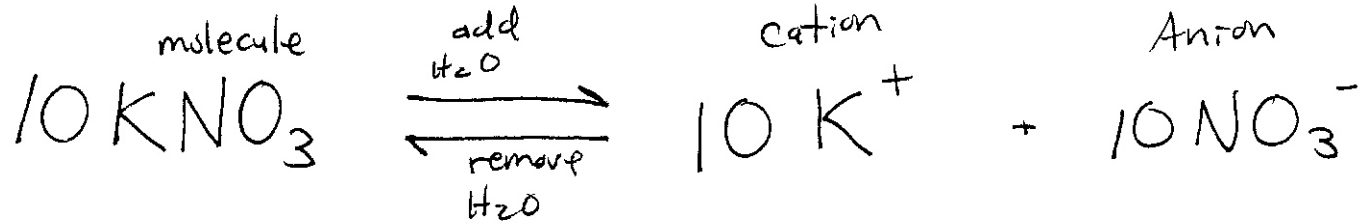
Figure 4

# CATION EXCHANGE



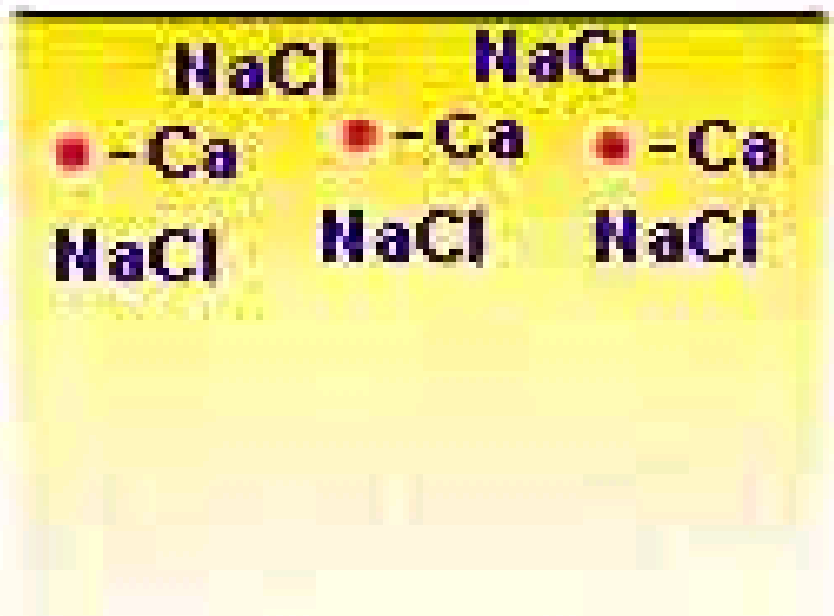
Paul Marks

# Potassium Nitrate

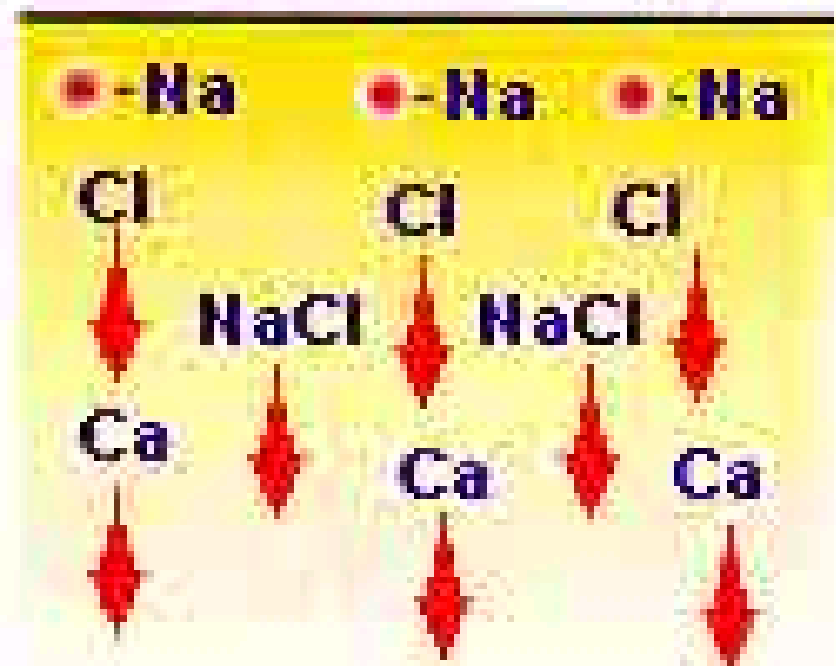


Rain

Irrigation



**SALINE SOIL**

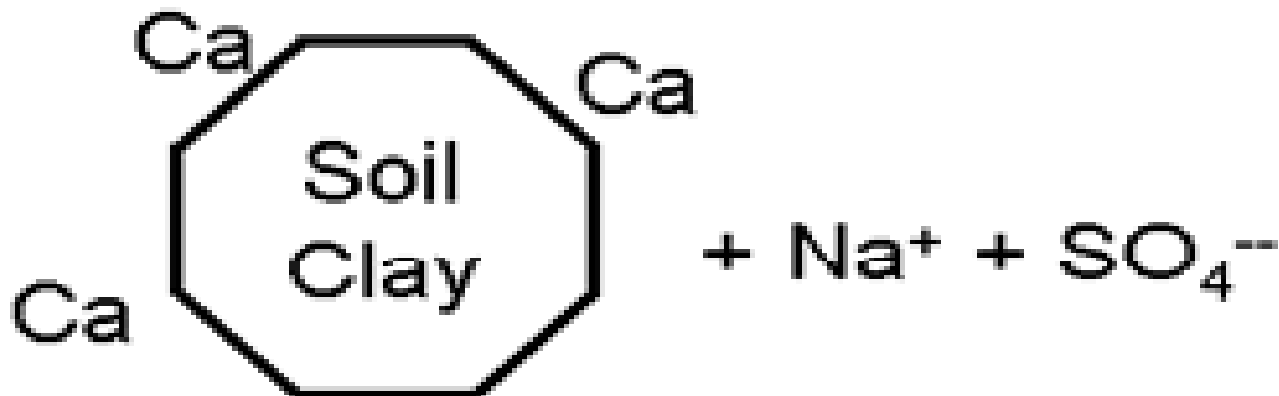
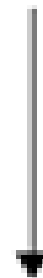
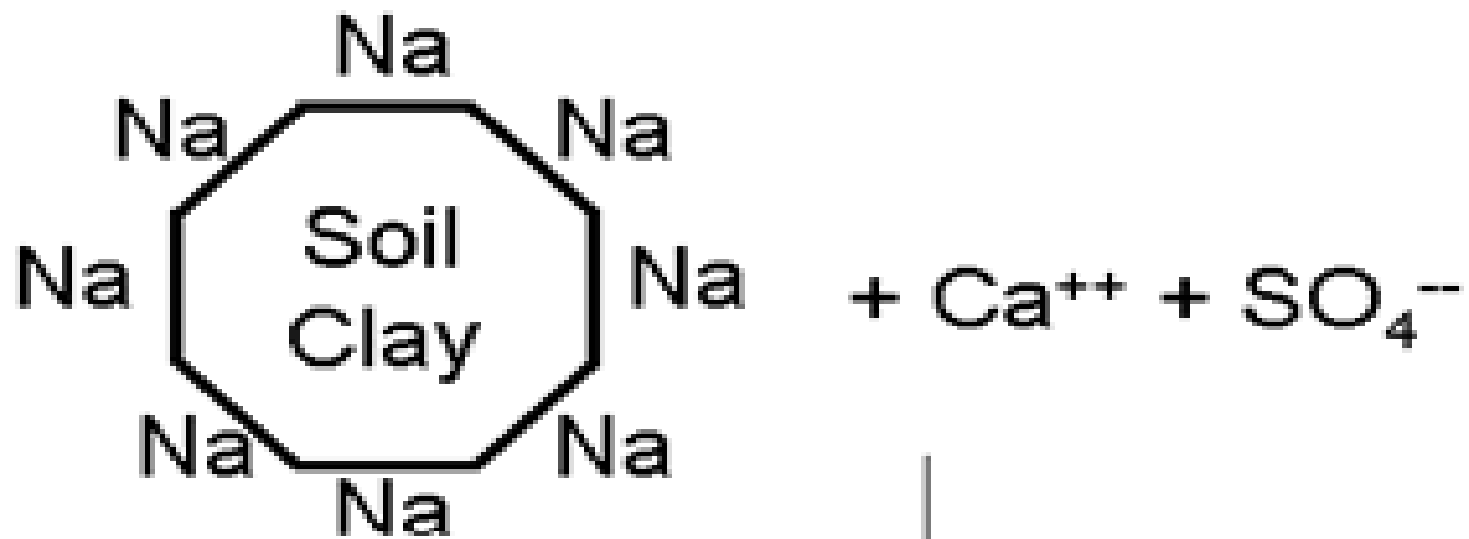


**SODIC SOIL**

**LEGEND**

NaCl Sodium chloride (salt)  
Na Sodium  
Cl Chloride

Ca Calcium  
● clay particles



# *Cation exchange capacity*

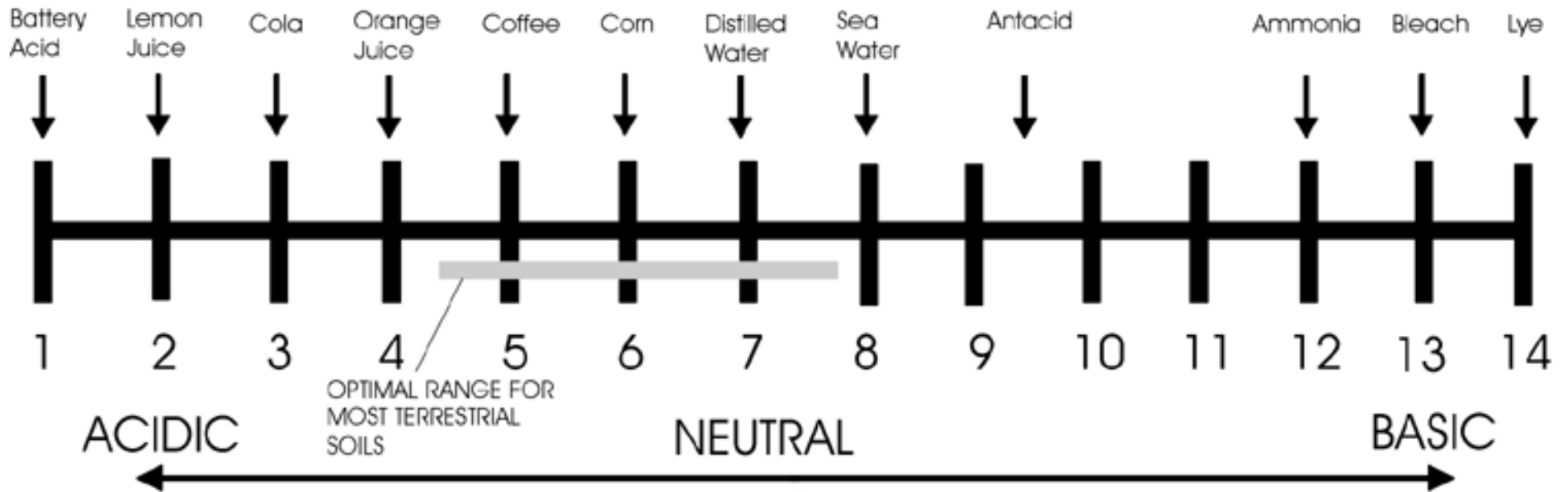
- Soil ability to attract, retain and exchange cations
- Indirect measurement of clay content because clays have more active (-) sites
- Coarser soils with fewer active sites less able to hold cations

# *Soil pH*

- Measure of the acidity or alkalinity of soil.
- Measures concentration of H in soil
- pH scale ranges from 0 – 14
- 0 – 7 = acidic; 7 = neutral; 7-14 = alkaline
- Negative logarithmic function – each whole number is a factor of 10 different from the next – 4.0 is 10 times more acidic than 5.0, 100 times more than 6.0, 1000 times more than 7.0



# pH Scale for Soils



## The pH Tolerance Levels for Plants





# *Soil pH*

- Most plants have a soil pH preference range that mirrors the soil conditions where the plant evolved
- Soil pH can be manipulated by additions of limestone (Ca and Mg oxides and carbonates to raise pH) or sulfur (lower pH)
- Buffering capacity – ability of soil to resist a change in pH – clays are better than sands

# Liming

- ✓ Increases soil pH



- ✓  $H^+$  is exchanged for  $Ca^{2+}$  or  $Mg^{2+}$  on cation exchange sites and the acidity of the soil is neutralized

# ***18 essential elements***

- Macronutrients – N, P, K, Ca, Mg, S – usually in 0.25% - 2.0% of woody plant dry matter
- Micronutrients – Fe, Mn, Cu, Zn, B, Mo, Cl, Co, Ni – content in woody plant dry matter usually in ppm range

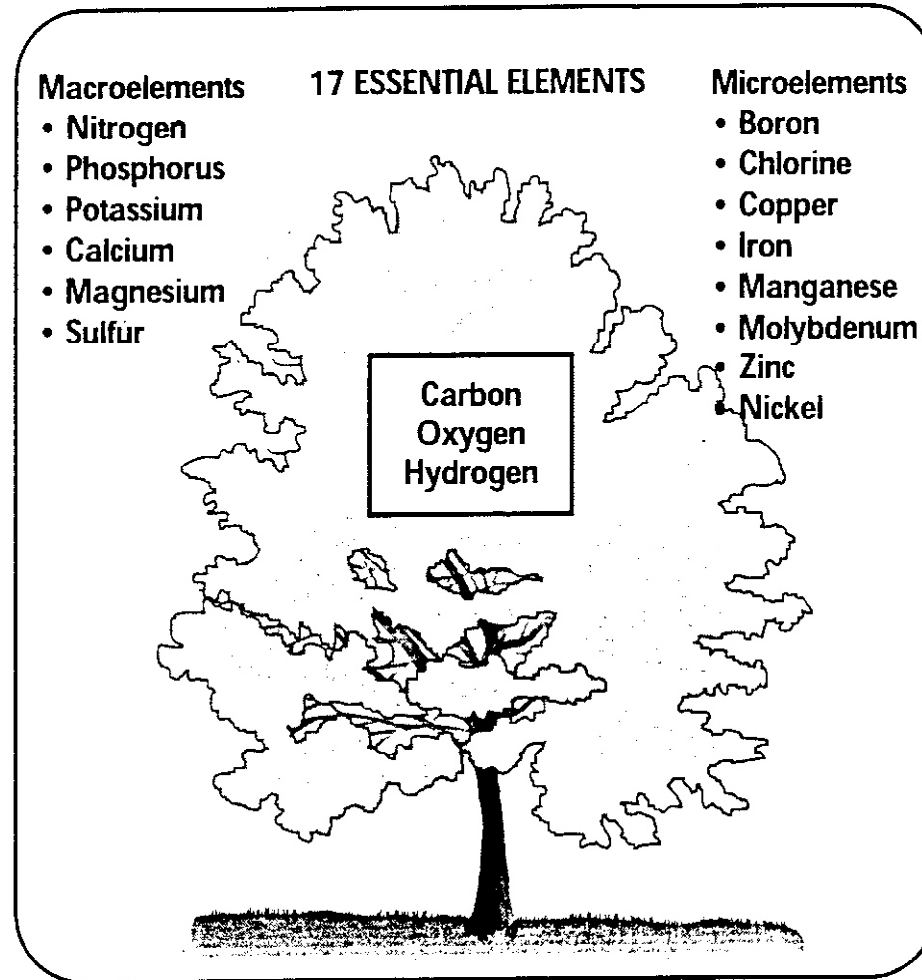


Figure 5.1 Trees require 17 essential elements.

# ***Carbon, hydrogen and oxygen***

- The basics for carbohydrates – the building blocks of plant tissue
- Make up 95 – 98% of most plants
- Derived from air (to roots and foliage) and water
- Availability depends on soil and root health

# ***Soil – a storehouse for nitrogen***

- Organic N average 0.15% in US soils
- Over 3000 lb N/A (6 inch depth)
- 68 lb N/1000 sq ft
- Slow availability
- Provides the bulk of the N taken up by woody plants in natural settings and landscapes





10-10-10 Fall



0-10-10 Spring

Fraser firs planted  
in 2001  
Treatments applied  
annually since  
2002

Photos taken  
3/8/06



No Fertilization



10-10-10 Spring



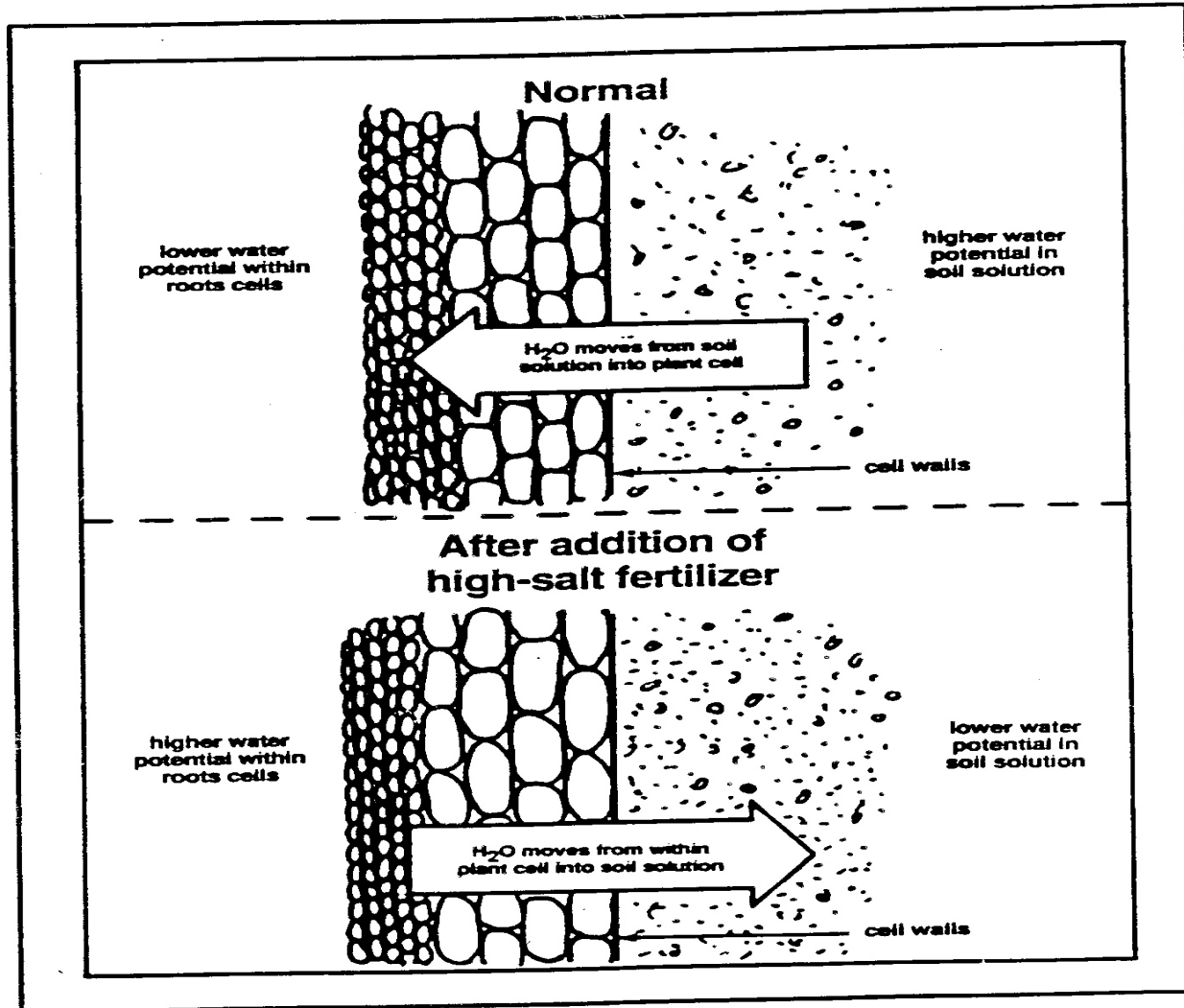
10-10-10 Spring & Fall

# *Fertilization*

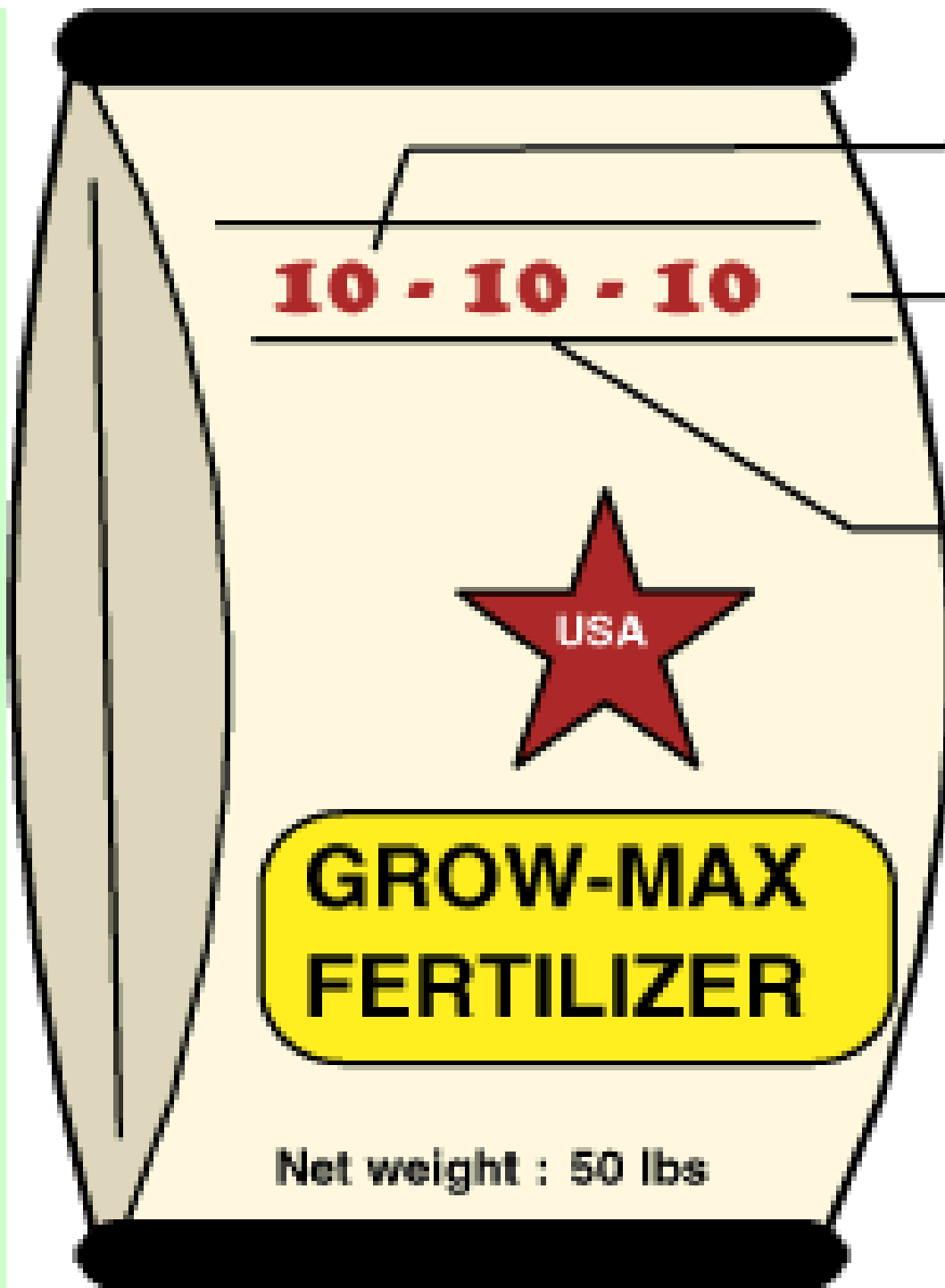
- Most nutrients supplied by soil in survival amounts but often not enough to meet modern landscape standards
- Fertilization is done to supplement nature's supply, often making the difference between thriving and surviving landscapes

# *Fertilizer terms*

- Complete fertilizer – contains nitrogen, phosphorus and potassium
- Fertilizer analysis – relative percentage of each element supplied and guaranteed on the label – a product with a 5-10-10 analysis has 5% N, 10% P ( $P_2O_5$ ) and 10% K ( $K_2O$ )
- Fertilizer burn – presence of excess fertilizer salts in root zone, pulling water out of roots by osmosis, causing scorched foliage or plant death



**Fig. 5.8** Excessive soluble salts in the root zone can reverse the osmotic process, drawing water out of the roots.



Available  
nitrogen ( N )

Available  
potash (  $K_2O$  )

Available  
phosphate (  $P_2O_5$  )

**10 - 10 - 10**

USA

**GROW-MAX  
FERTILIZER**

Net weight : 50 lbs

# ***Fertilizer terms***

- Inorganic fertilizer – release their essential elements when they contact water – quick, short term availability
- Controlled release fertilizer – releases essential elements over extended period; the release mechanism may be slow solubility, dependant on microbial degradation or coated with a permeable material
- Water insoluble nitrogen – 20-30% dissolution rate of N

# *Organic fertilizer*

- Chemical definition – Any fertilizer composed of carbon based molecules, synthetic or natural; chemical or biological decomposition needed to make essential elements available
- Traditional definition – any non-synthetic product derived from natural biological or geologic sources; can be chemically organic or inorganic

# ***Fertilization frequency***

- Dependant on soil conditions
- Fine textured soils can hold nutrients longer – larger amounts can be applied less frequently
- Coarse textured soils can not hold nutrients for very long – smaller, more frequent applications are more appropriate

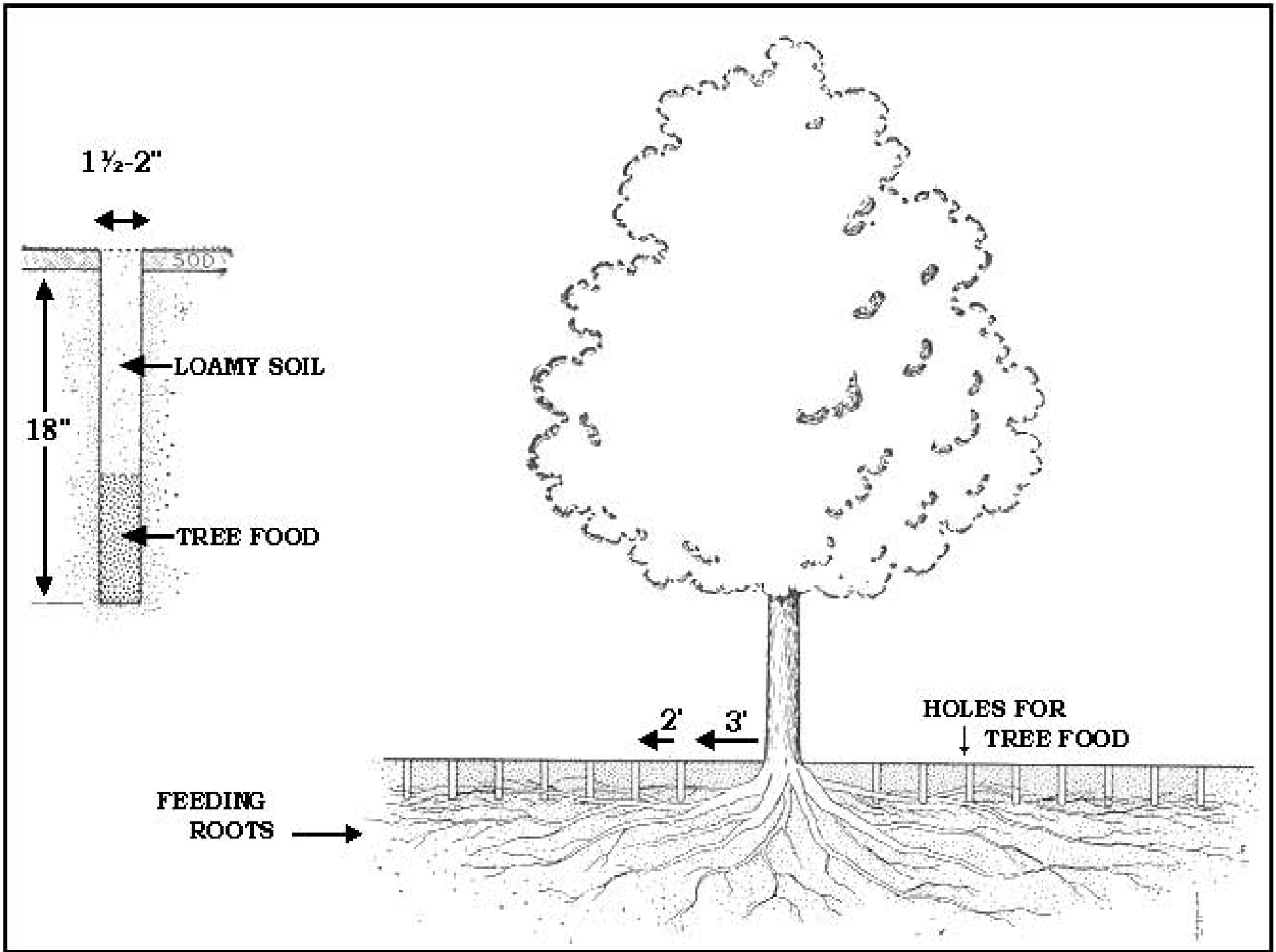


# ***Fertilization methods***

- Broadcast – easiest and least expensive; usually, granular fertilizer products are spread uniformly over soil surface – may interfere with other plantings/turf
- Drill hole method – concentric grid pattern of drill holes into which granular fertilizers are poured – affects trees but not other plants/turf

# *Fertilization methods*

- Subsurface injection (deep root feeding) – pressurized injection of dissolved or suspended fertilizer into the root zones of woody plants – little effect on other plants
- Foliar feeding – low concentration of soluble fertilizer in a fine mist directly to foliage – impractical for general fertilization but can be useful for temporary deficiency remedy
- Implants and tree injection – in capsule or solution injected directly into xylem – can work for some deficiencies or to stimulate local growth



# ***Fertility related analyses***

- Soil analysis – info on physical and chemical characteristics; soil pH, soil texture, organic matter content and major nutrients; limestone and fertilizer suggestions made – very useful to monitor over time; sampling method critical
- Foliar analysis – useful in conjunction with soil analysis; can accurately reflect how tree uses nutrition; sampling method critical





Adult exit hole

Oviposition sites

# The Emerald Ash Borer



Maine.gov

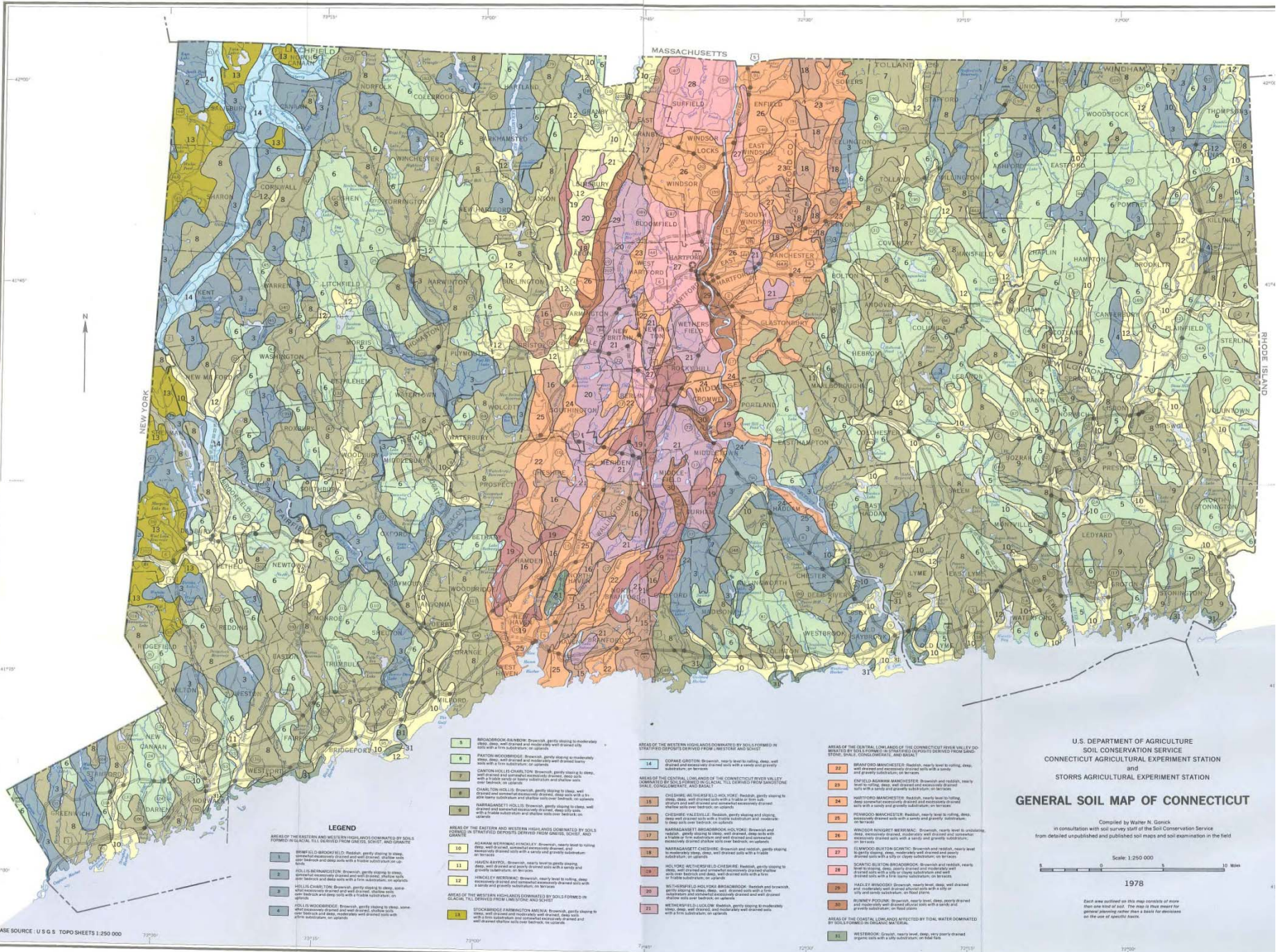
# Serpentine Galleries





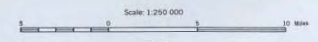


Jim Urban



U. S. DEPARTMENT OF AGRICULTURE  
SOIL CONSERVATION SERVICE  
CONNECTICUT AGRICULTURAL EXPERIMENT STATION  
and  
STORRS AGRICULTURAL EXPERIMENT STATION  
**GENERAL SOIL MAP OF CONNECTICUT**

Compiled by Walter R. Gunkel  
in consultation with soil survey staff of the Soil Conservation Service  
from detailed unpublished and published soil maps and soil examination in the field



1978

Each area outlined on this map consists of more than one kind of soil. The map is that meant for general planning rather than a basis for decisions on the use of specific lands.

BASE SOURCE: U. S. G. S. TOPO SHEETS 1:250,000